## SUPPLEMENTARY MATERIALS FOR:

## Mixing brine with oil triggered sphalerite deposition at Pine

## Point, Northwest Territories, Canada

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## ADDITIONAL INFORMATION ON METHODS

## Microthermometry of Fluid Inclusions

Samples were prepared into doubly-polished sections with a thickness of $60 \mu \mathrm{~m}$. The sections were mounted using an ambient-temperature, acetone-soluble glue to avoid any heating of the samples prior to microscopy, so as to avoid any stretching of the contained fluid inclusions. Petrographic analyses were conducted using a custom-built Olympus BX53 microscope at the University of Alberta, equipped for transmitted light microscopy in the visible and near-infrared ranges, as well as reflected light and ultraviolet-light fluorescence.

Fluid inclusion microthermometry was conducted using a Linkam THGMS 600 heatingfreezing stage mounted on the aforementioned Olympus BX53 microscope at the University of Alberta. Temperature of the stage was calibrated according to the triple point of $\mathrm{CO}_{2}\left(-56.6^{\circ} \mathrm{C}\right)$, the triple point of pure $\mathrm{H}_{2} \mathrm{O}\left(0.0^{\circ} \mathrm{C}\right)$, and the critical point of pure $\mathrm{H}_{2} \mathrm{O}\left(374.1^{\circ} \mathrm{C}\right)$ using synthetic fluid inclusion standards.

Primary and pseudo-secondary fluid-inclusion assemblages were identified according to the criteria outlined by Roedder (1984). Specifically, assemblages were deemed primary when they occurred along well-defined growth zones, whereas pseudo-secondary assemblage were those that occurred along cross-cutting trails that were truncated by later growth zones.

Because the brine inclusions analyzed here had very low volume fractions of vapor at ambient temperature (Fig. 2), we were cautious to avoid artificially high homogenization temperatures resulting from accidental stretching (Roedder, 1984). Therefore, during microthermometry, the samples were first heated to measure the homogenization temperatures, prior to freeing to measure the melting temperatures.

Whenever possible, we used the combination of ice-melting and hydrohalite-melting temperatures in an inclusion to estimate both total salinity and $\mathrm{Ca}: \mathrm{Na}$ ratio the model system $\mathrm{H}_{2} \mathrm{O}-\mathrm{NaCl}-\mathrm{CaCl}_{2}$ (Steele-MacInnis et al., 2011). In cases where only one or the other melting
temperature was observed, we estimated salinity in terms of equivalent weight percent NaCl (Steele-MacInnis et al., 2012).

## Laser Raman spectroscopy

Raman analyses were conducted at MacEwan University using a Bruker SENTERRA spectrometer and a 532 nm Ar+ laser, focused to a $1 \mu \mathrm{~m}$ spot through a 100x objective mounted on a standard petrographic microscope. All spectra were acquired on unoriented grains using a laser power of 20 mW and two to three, 5-20s exposures summed to the final reported spectra. Baseline subtraction and background reduction were applied using the Fityk ${ }^{\mathrm{TM}}$ software packages. Spectra were interpreted using the RRUFF database (Lafuente et al., 2016; mineral species), Frezzotti et al. (2012; hydrocarbons), and Walter et al. (2016; Brine sulfate).

## Laser ablation inductively coupled plasma mass spectrometry

The elemental composition of the fluid inclusions was analyzed at the Canadian Center for Isotopic Microanalysis at the University of Alberta using a Thermo Scientific ICAP-Q quadrupole inductively coupled plasma mass spectrometer (ICP-MS) coupled with a New Wave UP-213 laser ablation system. Helium ( $0.5 \mathrm{~L} / \mathrm{min}$ ) gas transported ablated particles from the ablation cell to the argon plasma. The laser operated at 10 Hz and $70 \%$ power. Beam diameter was adjusted as needed to fully ablate the fluid inclusions, between $25 \mu \mathrm{~m}$ and $65 \mu \mathrm{~m}$. In sphalerite crystals that showed poorer coupling with the laser (particularly, light-colored to colorless sphalerite), we set the laser to 20 Hz and $80 \%$ power to improve ablation. We used both
the NIST 610 and NIST 612 reference materials as external standards to calibrate sensitivity and drift.

The ICP-MS was set to detect 18 elements: ${ }^{43} \mathrm{Ca},{ }^{23} \mathrm{Na},{ }^{39} \mathrm{~K},{ }^{24} \mathrm{Mg},{ }^{137} \mathrm{Ba},{ }^{88} \mathrm{Sr},{ }^{208} \mathrm{~Pb},{ }^{57} \mathrm{Fe}$, ${ }^{55} \mathrm{Mn},{ }^{7} \mathrm{Li},{ }^{85} \mathrm{Rb},{ }^{133} \mathrm{Cs},{ }^{63} \mathrm{Cu},{ }^{68} \mathrm{Zn},{ }^{73} \mathrm{Ge},{ }^{111} \mathrm{Cd},{ }^{115} \mathrm{In}$ and ${ }^{75} \mathrm{As}$, all at a dwell time of 10 ms . Of the elements analyzed, $\mathrm{Na}, \mathrm{K}, \mathrm{Mg}, \mathrm{Ca}, \mathrm{Sr}, \mathrm{Rb}, \mathrm{Ba}$, and Pb were detected in the inclusions; all other elements were either below their respective detection limits, or masked by the sphalerite matrix (particularly Zn and Fe ). The LA-ICP-MS data were reduced using the SILLS software package (Guillong et al., 2008). The internal standard for the brine inclusions was either equivalent weight percent NaCl (for inclusions in which only ice melting was observed), or the estimated "true" NaCl concentration in terms of $\mathrm{H}_{2} \mathrm{O}-\mathrm{NaCl}-\mathrm{CaCl}_{2}$ (for inclusions in which both ice and hydrohalite melting were observed; Schlegel et al., 2012; Steele-MacInnis et al., 2016). The internal standard for the Oil inclusions was an estimated Na concentration of 100 ppm , based on a survey of ICP-OES analyses of various crude oils (de Oliviera Souza et al., 2015). The Na concentrations of natural crude oils show little variability, generally on the order of $\sim 10$ to $<200$ ppm (Wauquier, 1998). The calculated concentrations of metals in the oil inclusions will scale proportionally to estimated Na concentration as internal standard. For example, if the specified Na concentration of the oil is reduced by $50 \%$, then the calculated Pb concentrations will be similarly reduced by $50 \%$. Based on available analytical data for Na concentrations of crude oils (de Oliviera Souza et al., 2015), a value of 100 ppm was selected as a conservative estimate, and the true Pb concentrations are expected to fall within the range of 0.1 to 2 x the reported values based on this internal standard.

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22A2

22B2
Sample
Number

Location

DATA REPOSITORY TABLES

Table DR1: Samples and sample locations.

| Sample Number | Location | Description |
| :---: | :---: | :---: |
| 002.1 | $\begin{gathered} \text { N38 (Main } \\ \text { Trend) } \end{gathered}$ | Coarse sphalerite grown into open space filled by late-stage calcite. |
| 006.3 | L37 (Main Trend) | Series of cross-cutting "blue-vein" dolomite veins wherein coarse sphalerite postdates the "blue-vein" dolomite in each successive vein. |
| 007 | L37 (Main Trend) | Sphalerite plus dolomite grown into open space filled by late-stage calcite. |
| 011.1 | K77 (Main Trend) | "Blue-vein" dolomite vein containing coarse galena and sphalerite plus late-stage calcite. |
| 22A2 | $\begin{aligned} & \text { N38 (Main } \\ & \text { Trend) } \end{aligned}$ | Breccia with incorporated fragments of carbonate wall rock and earlier-formed (now brecciated) fragments of coarse sphalerite. |
| 22B2 | $\begin{aligned} & \text { N38 (Main } \\ & \text { Trend) } \end{aligned}$ | Breccia with incorporated fragments of carbonate wall rock and earlier-formed (now brecciated) fragments of coarse sphalerite. |

Table DR2: Microthermometry of brine inclusions.

| Inclusion Number | Inclusion Type | $\mathbf{T}_{\mathbf{H}}$ <br> $\left({ }^{\circ} \mathbf{C}\right)$ | $\mathbf{T}_{\mathbf{M} \text { Ice }}$ <br> $\left({ }^{\circ} \mathbf{C}\right)$ | Salinity <br> (eq. wt\% NaCl) |
| :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |
|  |  |  |  |  |
| 1 | Brine | 59.7 | -39.7 | 35.9 |
| 2 | Brine | 93.7 | -26.6 | 26.6 |
| 3 | Brine | 97.9 | -34.3 | 31.5 |
| 4 | Brine | 115.6 | -33.9 | 31.2 |
| 5 | Brine | 114.6 | -38.8 | 34.3 |
| 6 | Brine | 96.7 | -38.0 | 35.1 |
| 7 | Brine | 96.7 | -43.8 | 40.0 |
| 8 | Brine | 100.7 | -43.8 | 40.0 |
| 9 | Brine | 86.8 | -34.0 | 31.3 |
| 10 | Brine | 86.8 | -34.0 | 31.3 |
| 11 | Brine | 90.3 | -21.7 | 34.1 |
| 12 | Brine | 79.4 | -38.0 | 36.4 |
| 13 | Brine | 79.7 | -34.9 | 34.3 |
| 14 | Brine | 79.7 | -34.9 | 34.3 |
| 15 | Brine | 80.9 | -35.7 | 35.0 |
| 16 | Brine | 79.4 | -38.0 | 36.4 |
| 17 | Brine | 100.4 | -38.8 | 35.0 |
| 18 | Brine | 91.4 | -40.1 | 36.4 |
| 19 | Brine | 91.4 | -40.1 | 36.3 |
| 20 | Brine | 100.9 | -40.3 | 36.4 |
| 21 | Brine | 94.6 | -32.6 | 30.3 |
| 22 | Brine | 63.4 | -19.6 | 22.1 |

Table DR3: Microthermometry of Oil inclusions.

| Inclusion Number | Inclusion Type | $\mathbf{T}_{\mathbf{H}}$ <br> $\left({ }^{\circ} \mathbf{C}\right)$ | $\mathbf{T}_{\mathbf{M}}$ <br> $\left({ }^{\circ} \mathbf{C}\right)$ |
| :---: | :---: | :---: | :---: |
|  |  | 71.8 | - |
| 23 | Oil | 70.1 | - |
| 24 | Oil | 77.3 | - |
| 25 | Oil | 101.8 | - |
| 26 | Oil | 101.9 | - |
| 27 | Oil | 82.3 | - |
| 28 | Oil | 66.8 | - |
| 29 | Oil | 71.3 | - |
| 30 | Oil | 81.3 | - |
| 31 | Oil | 66.3 | - |
| 32 | Oil | 51.4 | - |
| 33 | Oil | 53.2 | - |
| 34 | Oil | 49.8 | - |
| 35 | Oil | 52.6 | - |
| 36 | Oil | 52 | - |
| 37 | Oil | 56.9 | - |
| 38 | Oil | 71.3 | - |
| 39 | Oil | 66.2 | - |
| 40 | Oil | 72.6 | - |
| 41 | Oil | 73.4 | - |
| 42 | Oil | 69.8 | - |
| 43 | Oil | 74.8 | - |
| 44 | Oil |  |  |
|  |  |  |  |
| *Oil inclusions did not freeze. |  |  |  |

## 139 Table DR4: LA-ICP-MS results

| Inc. <br> Number | Inc. <br> Type | $\mathbf{N a}$ <br> $\mathbf{( p p m})$ | $\mathbf{M g}$ <br> $(\mathbf{p p m})$ | $\mathbf{K}$ <br> $(\mathbf{p p m})$ | $\mathbf{C a}$ <br> $(\mathbf{p p m})$ | $\mathbf{M n}$ <br> $(\mathbf{p p m})$ | $\mathbf{C u}$ <br> $(\mathbf{p p m})$ | $\mathbf{G e}$ <br> $(\mathbf{p p m})$ | $\mathbf{R b}$ <br> $(\mathbf{p p m})$ | $\mathbf{S r}$ <br> $(\mathbf{p p m})$ | $\mathbf{B a}$ <br> $(\mathbf{p p m})$ | $\mathbf{P b}$ <br> $(\mathbf{p p m})$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | Brine | 42885.2 | 11939.6 | 4694.7 | 62899.7 | $<0.7$ | 12.1 | 404.3 | 11.6 | 1536.8 | 25.2 | $<0.009$ |
| 2 | Brine | 31709.6 | 9201.6 | 3982.1 | 45745.8 | 36.9 | 48.0 | 41.4 | 11.7 | 1156.7 | 28.1 | 336.3 |
| 3 | Brine | 30751.3 | 11735.8 | 18414.6 | 51955.7 | $<0.7$ | 29.6 | 107.4 | 17.1 | 1163.0 | 21.8 | 333.1 |
| 4 | Brine | 31808.2 | 10430.7 | 11862.2 | 55469.0 | $<8.0$ | $<2.5$ | 30.2 | 9.0 | 1219.8 | 11.7 | 318.2 |
| 5 | Brine | 41277.0 | 11481.2 | 17451.5 | 55911.8 | 48.5 | 6.1 | $<3.7$ | 10.1 | 1283.1 | 19.6 | 62.2 |
| 6 | Brine | 46303.1 | 12988.0 | 20242.0 | 64160.2 | 20.7 | 3.9 | 211.7 | 10.4 | 1530.3 | 17.7 | 564.4 |
| 7 | Brine | 48308.1 | 13598.1 | 18797.4 | 62180.5 | $<8.7$ | $<2.5$ | 31.8 | 21.0 | 1528.8 | 19.6 | 392.0 |
| 8 | Brine | 26877.5 | 8798.9 | 4852.2 | 95790.2 | 159.3 | $<2.9$ | 299.2 | 2.7 | 1987.9 | 20.1 | 116.2 |
| 9 | Brine | 35535.1 | 9114.1 | 11241.8 | 55079.8 | $<17.2$ | 362.9 | 26.8 | 5.0 | 1390.8 | 24.4 | 299.7 |
| 10 | Brine | 38083.4 | 10884.0 | 11185.4 | 50040.1 | $<2.1$ | $<0.5$ | $<0.7$ | 12.9 | 1235.0 | 22.7 | $<0.03$ |
| 11 | Brine | 37429.4 | 10908.8 | 8191.8 | 61591.8 | 26.7 | 587.2 | 55.8 | 4.3 | 1404.7 | 19.1 | 191.7 |
| 12 | Brine | 41904.1 | 11805.1 | 7566.5 | 64263.4 | 8.5 | 243.4 | $<0.2$ | 7.7 | 1548.4 | 19.6 | 129.2 |


| 151 Table DR4，Continued |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
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| © | $\stackrel{\square}{\circ}$ | $\overline{\text { ®on }}$ | $\underset{\hat{0}}{\hat{O}}$ | $\stackrel{\circ}{\dot{N}}$ | $\stackrel{3}{\square}$ | $\stackrel{\widetilde{n}}{\stackrel{i}{v}}$ | $\begin{aligned} & \underset{\sim}{a} \\ & \stackrel{\infty}{v} \end{aligned}$ | $\stackrel{丸}{\stackrel{\rightharpoonup}{\circ}}$ | － | $\stackrel{\underset{\vec{F}}{v}}{ }$ | 声 | $\stackrel{Y}{\mathrm{~F}}$ | $\stackrel{\infty}{\dot{j}}$ | $\stackrel{\square}{\square}$ | $\stackrel{\rightharpoonup}{\mathbf{0}}$ | $\begin{aligned} & \stackrel{\circ}{\dot{玉}} \\ & \underset{\underline{0}}{ } \end{aligned}$ | N | $\begin{gathered} \infty \\ \underset{\sim}{⿺} \end{gathered}$ | $\begin{gathered} \tilde{C} \\ \stackrel{\rightharpoonup}{\partial} \\ \hline \end{gathered}$ | $\stackrel{\text { I }}{\square}$ | $\stackrel{\circ}{\dot{7}}$ | $\stackrel{\square}{2}$ |
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## DATA REPOSITORY FIGURES



Figure DR1. Histograms of homogenization temperatures of brine (top) and oil (bottom) inclusions.

