## METHODS

### **Grain Boundary Imaging and Orientation Analysis**

Backscatter electron (BSE) maps of thin sections were acquired using the FEI Verios XHR scanning electron microscope (SEM) at the Centre for Microscopy and Analysis (CMCA) at The University of Western Australia, Perth. Analytical conditions are listed in Table 1.

The two different grain boundary assemblages were colour coded on each of the thin section BSE maps. The assemblage shape and orientation (Fig. 3 A) was analysed using the Analyze Particles tool in ImageJ (Schneider et al., 2012). Orientation data was then plotted using Stereonet 10 (Cardozo and Allmendinger, 2013) and shape data was plotted and fitted with Gaussian curves using the Curve Fitting Toolbox in MATLAB.

# Mineral Chemistry and Calculation of Assemblage Composition

The average modal composition of grain boundary assemblages was calculated through combined electron backscatter diffraction (EBSD) and energy dispersive X-ray spectroscopy (EDS) using the MIRA variable pressure field-emission scanning electron microscope (VP-FE-SEM) at the John de Laeter Centre, Curtin University, Perth. Analytical conditions are listed in Table 1.

Modal compositions of grain boundary assemblages were then combined with mineral compositions acquired by wavelength-dispersive X-ray spectroscopy (WDS) on the JEOL 8530F electron probe micro analyser (EMPA) at the CMCA. Analytical conditions are listed in table 1.

The precursor plagioclase composition used in the isocon chemical diagram (Fig. 4 A) was calculated through quantitative WDS mapping of the core of a large (> 1mm diameter) plagioclase grain in the least altered anorthosite, using the operating conditions listed in Table 1.

Verios XHR SEM	MIRA VP-FESEM	JEOL 8530F EMPA
BSE imaging	EDS and EBSD	WDS
4 mm	20 mm	13 mm
15 keV	20 keV	15 keV
	17 nA	
0.8 nA	5 nA	15 nA
	BSE imaging 4 mm 15 keV	BSE imagingEDS and EBSD4 mm20 mm15 keV20 keV17 nA

#### Table DR1 – Analytical conditions for instruments used.

#### **Bulk Rock Composition and Thermodynamic Modelling**

The bulk rock concentrations of major element oxides in samples have been determined by X-Ray Fluorescence (XRF) spectrometry. The obtained compositions of the anorthosite and amphibolite samples are consistent with those calculated based on modal abundance (TIMA) and phase composition (EPMA) and are thus considered representative (for comparison see Table DR1).

Thermodynamic modelling was undertaken using the software THERMOCALC 3.45 (Powell and Holland, 1988) and the internally consistent thermodynamic dataset and equation of state for  $H_2O$  of Holland and Powell (2011). The NCKFMASHTO (Na<sub>2</sub>O-CaO -K<sub>2</sub>O-FeO-MgO-Al<sub>2</sub>O<sub>3</sub>-SiO<sub>2</sub>-H<sub>2</sub>O-TiO<sub>2</sub>-Fe<sub>2</sub>O<sub>3</sub>) system was used with the solution models given in Table 2.

Solution model	used abbreviation	References
Clinopyroxene	dio	Green et al. (2016)
Clinoamphibole	hb	Green et al. (2016)
Epidote	CZ	Holland and Powell (2011)
Garnet	g	White et al. (2014)
Orthopyroxene	орх	White et al. (2014)
Muscovite	mu	White et al. (2014)
Biotite	bi	White et al. (2014)
Ternary feldspar	pl	Holland and Powell (2003)
Tonalitic 'metabasite' melt	L	Green et al. (2016)

Table DR2– Mineral solution models used for the computation of P-T pseudosections in THERMOCALC 3.45

#### **REFERENCES CITED**

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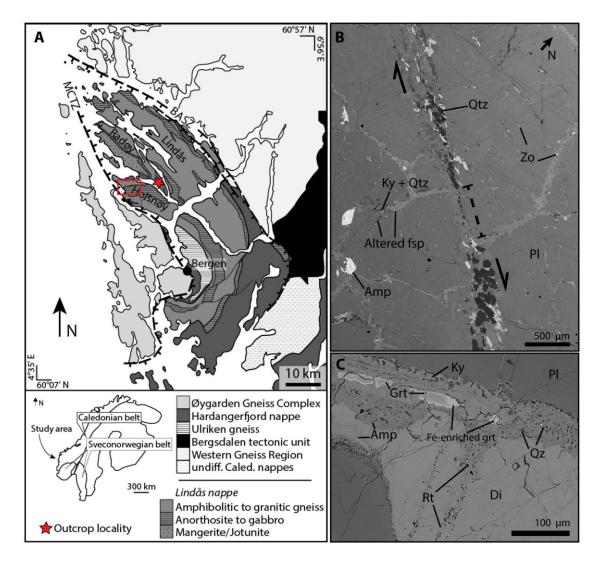


Figure DR1 – Study locality and details of alteration (A) Sample location (red star) at Radøy in the Lindås Nappe of the Bergen Arc, Western Norway. Sample location in Universal Transverse Mercator [UTM] coordinates relative to World Geodetic System 1984 [WGS84]; zone 31V, 620260E, 6718524N. The red box indicates the location of eclogite shear zones studied by Raimbourg et al. (2005). MCTZ = Main Caledonian Thrust Zone. BASZ = Bergen Arc Shear Zone. Modified from Boundy et al. (1997) and Glodny et al. (2008) (B) Backscatter electron (BSE) image showing the offset of an original plagioclase grain boundary (indicated by dashed line) along an extensional quartz-vein, indicating a dextral top-to-the-ESE sense of displacement during hydration of the anorthosite. (C) BSE image of the alteration assemblage occurring at diopside and garnet grain boundaries, primarily consisting of hornblende with minor quartz, rutile and Fe-enriched garnet.

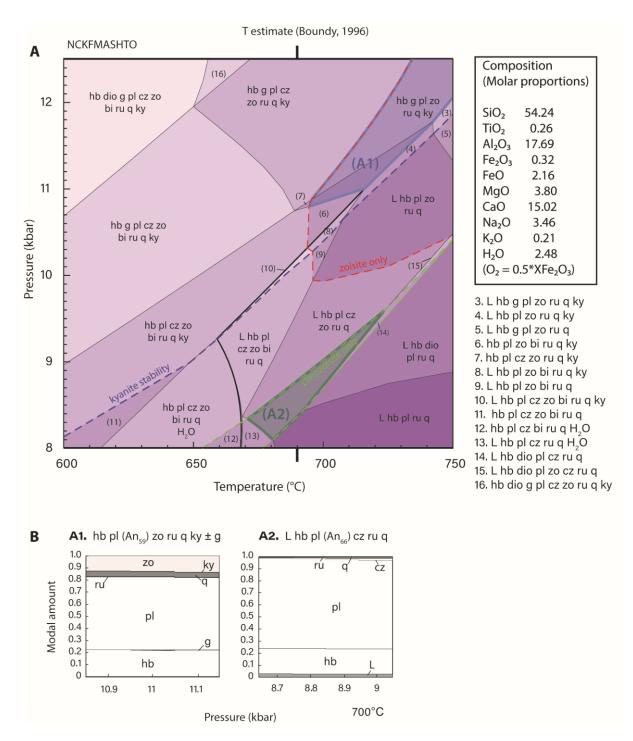


Figure DR3 – Thermodynamic modelling of amphibolite bulk rock (A) Pseudosection for amphibolite assemblages. The PT space of the two grain boundary assemblages is outlined in bold lines and coloured according to the scheme used in the manuscript. (B) Modelled modal compositions for assemblages at 700°C are given for each of the assemblages. The An-component estimates produced by THERMOCALC are consistent with the observed plagioclase compositions. An<sub>58</sub> for A1, consistent with the composition of the host plagioclase and An<sub>66</sub> for A2.

Table DR3 – Bulk XRF (used in isocon diagram Fig. 4 A) and TIMA map calculated whole rock composition for granulitic anorthosite and amphibolite, and, local
composition of grain boundary assemblages I and II, calculated using modal mineral volumes and electron microprobe mineral compositions (All values are weight%
oxide; sample density is in g/cm³; n.m. = not measured; LOI = loss on ignition)

Analytical technique	XRF*	TIMA-X	XRF*	TIMA-X	EBSD + WDS	EBSD +WDS	Quantitative WDS mapping*
Analytical technique	Granulitic anorthosite		Amphibolite		Assemblage I	Assemblage II	Precursor plagioclase
Sample density	2.87		2.86				
SiO <sub>2</sub>	49.9	50.5	48.9	49.9	57.8	52.8	53.7
TiO <sub>2</sub>	0.11	0.12	0.21	0.32	0.02	0.02	0.20
Al <sub>2</sub> O <sub>3</sub>	27.0	26.5	26.5	27.6	30.8	28.3	29.3
Fe <sub>2</sub> O <sub>3</sub>	3.58	0.60	3.50	0.81	0.71	0.14	0.19
FeO	n.m.	2.28	n.m.	2.37	0.01	0.23	n.m.
MgO	3.68	3.33	4.11	3.08	0.03	0.44	0.03
CaO	11.9	12.6	11.8	12.9	9.46	11.4	11.9
Na <sub>2</sub> O	3.24	3.55	3.23	3.28	0.00	3.49	4.71
K <sub>2</sub> O	0.27	0.27	0.28	0.30	0.35	2.33	0.28
LOI	0.22	n.m.	0.83	n.m.	n.m.	n.m.	n.m.
H <sub>2</sub> O	n.m.	0.01	n.m.	0.68	0.83	0.12	n.m.
Total	99.8	99.8	100	99.8	100	99.3	100

\* All Fe calculated as Fe<sub>2</sub>O<sub>3</sub>