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1 SUPPLEMENTAL INFORMATION

2 LOSCAR SIMULATION OF ETM-2

3 We apply the LOSCAR (Long-term Ocean-atmosphere-Sediment CArbon cycle Reservoir Model v2.0.4; Zeebe, 2012) climate/carbon-cycle model to simulate the ETM-2 carbon 4 emission scenario given observations of CCD, planktic δ^{13} C and Δ SST (see Fig. DR2), with the 5 goal of generating ΔpH estimates across the event so that pH-adjustments may be applied to 6 planktic δ^{18} O and Mg/Ca. This model has been used extensively in previous research to simulate 7 8 PETM and modern emission scenarios (Zachos et al., 2008; Zeebe et al., 2009; Sluijs et al., 2012; Zeebe, 2013; Zeebe and Zachos 2013; Penman et al., 2014; Zeebe et al., 2016). We apply 9 the same boundary conditions (i.e., early Eocene ocean geometry and climate response functions 10 based on early Eocene climate sensitivity) used in recent PETM simulations (Zeebe et al., 2016). 11 We then simulate the carbon emission scenario by releasing 1,300 Pg C over 25 kyr, at time = 0 12 kyr (consistent with the astronomically-paced onset described in Lourens et al. (2005)), using an 13 intermediate δ^{13} C of -40‰, with the aim of matching simulations to observations of the 14 15 magnitude of the CIE and CCD changes (Fig. DR2). The ETM-2 simulation indicates a pH decrease of ~ 0.05 . The temperature anomaly produced by this simulation falls within the range 16 of pH-adjusted and non-adjusted Mg/Ca-based temperatures (Fig. DR3). 17

18 ANALYTICAL CHEMISTRY

Foraminifera (25 individuals) were crushed, homogenized, and split into two samples,
one for trace elements and one for stable isotopes (δ¹³C/δ¹⁸O). Trace element samples were
cleaned following the oxidative reductive protocol of Barker et al. (2003), dissolved in 0.075N
HNO₃ and analyzed via ICP-MS on a Thermo Element XR following the methodology of Brown

et al. (2011). The long-term reproducibility of consistency standard measurements indicates
inter-run precision for Mg/Ca is <3% (2 s.d.).

25	The sample portion used for stable isotope analysis was not cleaned following
26	oxidation/reduction protocol. These samples were analyzed on a Thermo MAT 253 IR-MS
27	coupled to a Kiel IV carbonate device. Based on replicate measurements of consistency
28	standards, inter-run precision for δ^{13} C and δ^{18} O is <0.1% (2 RSD) and <0.16% (2 RSD),
29	respectively. The bulk %CaCO ₃ ETM-2 record for Site 1209 was generated using a UIC Carbon
30	Coulometer Analyzer.

31 Δ SST AND Δ SSS COMPUTATIONS

32 Planktic Mg/Ca is pH-adjusted using the LOSCAR-simulated ΔpH for ETM-2 following
33 the logistic pH-adjustment from Evans et al. (2016):

34
$$Mg/Ca = \frac{0.66}{1 + \exp(6.9*(pH - 8.0))} + 0.76. (1)$$

Additionally, because Evans et al. (2016) could not rule out a linear fit to their data, we also
include the linear pH-adjustment in our SST and SSS anomaly envelopes:

37
$$Mg/Ca = -0.70 * pH + 6.7. (2)$$

These pH-adjustments shift temperatures by <0.1°C due to the small pH decrease (~-0.05 pH units) we simulate for ETM-2. Furthermore, when larger magnitude boron-based ΔpH (~-0.11 pH units) is applied, the added effect on ΔSST is less than -0.1°C, and less than -0.2 ppt for ΔSSS (Fig. DR3). SST anomalies are generated using the pH-adjusted planktic Mg/Ca following Zachos et al. (2003):

43
$$\Delta T = \frac{1}{A} \ln \left[\left(\frac{C}{100} \right) + 1 \right]. (3)$$

Where 'C' is the percent change in Mg/Ca relative to baseline and 'A' is the exponential constant
in the Mg/Ca-temperature calibration (i.e., Mg/Ca-temperature sensitivity for a species of

46	foraminifera). Here we apply a range of 'A' values derived using culture calibration
47	measurements of the modern species G. ruber in which Mg/Caseawater was varied, following
48	Evans et al. (2016). The best regression fits through the culture calibration data indicate a
49	decrease in sensitivity (i.e., lower 'A' value) with lower Mg/Ca _{seawater} . Given the possible range
50	of Mg/Ca _{seawater} in the early Eocene, we apply a range in 'A' values of 0.05 to 0.09 to generate
51	SST anomaly envelopes. This method incorporates errors due to the potential uncertainty in
52	Mg/Ca _{seawater} for the early Eocene, however, it is still limited due to the fact that sensitivities are
53	based on culture calibration of a single modern planktic foraminifera species. Anomaly
54	envelopes do not incorporate any changes in Mg/Ca _{seawater} across the hyperthermal, which is
55	appropriate given the residence times of Mg and Ca in seawater (~13 Ma and ~1 Ma,
56	respectively; Broecker and Peng, 1982) and the time interval of the anomaly envelope (~200
57	kyr).
58	To generate SSS anomaly envelopes, we assume any difference in pH-adjusted δ^{18} O-
59	based Δ SST compared to pH-adjusted Mg/Ca-based Δ SST is due to the effect of local surface
60	salinity changes on δ^{18} O. Mg/Ca-based Δ SSTs are converted into expected δ^{18} O temperature
61	anomalies following the relationship of 0.213‰/°C from Zachos et al. (2003). We then subtract
62	the pH-adjusted (-2.51‰ per pH unit following Zeebe, 1999) observed δ^{18} O anomaly from a
63	theoretical temperature-based $\delta^{18}O$ record generated using the Mg/Ca-based temperature change.

64 This produces a residual δ^{18} O anomaly. This residual value represents the surface salinity signal

65 in planktic δ^{18} O. SSS anomaly envelopes incorporate both the range in the Mg/Ca-temperature

calibration constants and the possible range of the $\Delta \delta^{18}$ Oseawater / Δ SSS relationship (0.25-

67 0.50‰/salinity unit) from Zachos et al. (2003).

68 **REFERENCES CITED**

69	Barker, S., Greaves, M., and Elderfield, H., 2003, A study of cleaning procedures used for
70	foraminiferal Mg/Ca paleothermometry: Geochemistry Geophysics Geosystems, v. 4, no.
71	9.
72	Broecker, W. S., and Peng, T. H., 1982, Tracers in the Sea, p. 125-159.
73	Brown, R. E., Anderson, L. D., Thomas, E., and Zachos, J. C., 2011, A core-top calibration of
74	B/Ca in the benthic foraminifers Nuttallides umbonifera and Oridorsalis umbonatus: A
75	proxy for Cenozoic bottom water carbonate saturation: Earth and Planetary Science
76	Letters, v. 310, no. 3, p. 360-368.
77	Evans, D., Wade, B. S., Henehan, M., Erez, J., and Müller, W., 2016, Revisiting carbonate
78	chemistry controls on planktic foraminifera Mg/Ca: implications for sea surface
79	temperature and hydrology shifts over the Paleocene-Eocene Thermal Maximum and
80	Eocene–Oligocene Transition: Climate of the Past, v. 11, p. 3143-3185.
81	Penman, D. E., Hönisch, B., Zeebe, R. E., Thomas, E. and Zachos, J. C., 2014, Rapid and
82	sustained surface ocean acidification during the Paleocene-Eocene Thermal Maximum:
83	Paleoceanography, v. 29, p. 357–369.
84	Sluijs, A., Zachos, J. C., and Zeebe, R. E., 2012, Constraints on hyperthermals: Nature
85	Geoscience, v. 5, no. 4, p. 231-231.
86	Zachos, J. C., Wara, M. W., Bohaty, S. M., Delaney, M. L., Rose-Petrizzo, M., Brill, A.,
87	Bralower, T. J. and Premoli-Silva, I., 2003, A transient rise in tropical sea surface
88	temperature during the Paleocene-Eocene Thermal Maximum: Science, v. 302, p. 1551-
89	1554.
90	Zachos, J. C., Dickens, G. R. and Zeebe, R. E., 2008, An early Cenozoic perspective on
91	greenhouse warming and carbon-cycle dynamics: Nature, v. 451, p. 279–283.

92	Zeebe, Richard E., 1999, An explanation of the effect of seawater carbonate concentration on
93	foraminiferal oxygen isotopes: Geochimica et Cosmochimica Acta, v. 63, no. 13, p.
94	2001-2007.
95	Zeebe, R. E., Zachos, J. C., and Dickens, G. R., 2009, Carbon dioxide forcing alone insufficient
96	to explain Palaeocene-Eocene Thermal Maximum warming: Nature Geoscience, v.2, no.
97	8, p. 576-580.
98	Zeebe, R. E., 2012, LOSCAR: Long-term Ocean-atmosphere-Sediment CArbon cycle Reservoir
99	Model v2.0.4: Geoscientific Model Development, v. 5, p. 149-166.
100	Zeebe, R. E., 2013, What caused the long duration of the Paleocene-Eocene Thermal
101	Maximum?: Paleoceanography, v. 28, no. 3, p. 440-452.
102	Zeebe, R. E. and Zachos, J. C., 2013, Long-term legacy of massive carbon input to the Earth
103	system: Anthropocene vs. Eocene: Royal Soc. London Phil. Trans. A., v. 371, no. 2001.
104	Zeebe, R. E., Ridgwell, A., and Zachos, J. C., 2016, Anthropogenic carbon release rate
105	unprecedented during the past 66 million years: Nature Geoscience, v. 9, no. 4, p. 325-
106	329.
107	FIGURE CAPTIONS
108	Figure DR1. Bulk carbonate and carbonate nodule δ^{13} C from pelagic ocean sites (ODP Sites
109	1265 and 1209; Stap et al., 2009 and Gibbs et al., 2012) and terrestrial Big Horn Basin Site
110	(Abels et al., 2012) illustrate that the perturbation to carbon cycle during ETM-2 was global. Site
111	1265 ages are based on the bulk carbonate Site 1263 age model of Lauretano et al. (2016).

113 Figure DR2. LOSCAR carbon emission scenario for ETM-2: 1,300 Pg of carbon is released to 114 atmosphere at time=0 kyr with δ^{13} C=-40‰. pH simulations for the emission scenario are used to 115 pH-adjust planktic Mg/Ca and δ^{18} O.

116

117 Figure DR3. Sensitivity test showing the potential influence of pH adjustment on our SST and SSS records at Sites 1209 and 1265 using 3 different pH change scenarios. The black line 118 119 represents constant pH. The red line represents a pH change of -0.05 pH units derived from 120 LOSCAR simulations and used in this study. We show both the linear and logistic pH 121 adjustments of Evans et al. (2016). Intermediate values for the Mg/Ca temperature calibration sensitivity ('A' value of 0.075) and intermediate δ^{18} O-salinity sensitivity (0.33%) per salinity 122 unit) are applied instead of the ranges used previously to clearly display the change in the Δ SST 123 124 and Δ SSS records for each Δ pH scenario.

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126 Figure DR4. Δ SST and Δ SSS for the PETM at ODP Site 1209 using foraminiferal data (mixed-127 layer dweller Acarinina soldadoensis) from Zachos et al. (2003), which is pH-adjusted using 128 Pacific surface ΔpH data from LOSCAR PETM simulations (Zeebe et al., 2009). SST envelopes 129 are generated similarly to ETM-2 envelopes (Fig. 2) using a range of Mg/Ca-temperature sensitivities ('A' values from 0.05 to 0.09) and include both linear and logistic pH-adjustments 130 131 from Evans et al. (2016). Note that the larger range in both SST and SSS envelopes shown here, 132 compared with Zachos et al. (2003), is a function of the larger range of Mg/Ca temperature 133 sensitivities given the recommendations of Evans et al. (2016) (see main text). The range in Δ SSS incorporates a range of δ^{18} O-salinity sensitivities (the same range as the Δ SSS envelopes 134 135 displayed in Fig. 2; 0.25-0.5‰ per salinity unit).







