

Pulsed magmatic fluid release for the formation of porphyry deposits: tracing fluid evolution in absolute-time from the Tibetan Qulong Cu-Mo deposit

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53 grain to reach oxygen equilibrium with fluid at 300 $^{\circ}\text{C}$ with a salinity of 5.8 NaCl equiv. %,
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57 analyzed at Qulong, with an absolute timeframe determined by molybdenite Re-Os dating (Li
58 et al., 2017a). Fluid $\delta^{18}\text{O}$ values are calculated using formation temperatures determined from
59 a fluid inclusion study (Li et al., 2017b). Fluid $\delta^{18}\text{O}$ values are also calculated at hypothesized
60 temperatures of at 420, 350 and 300 $^{\circ}\text{C}$, which are the upper, average and lower limits of
61 mineralization temperature at Qulong and PCD in general (Richards, 2011; Kouzmanov and
62 Pokrovski, 2012). Also presented is the magmatic water percentage contribution at these
63 variable temperatures. The $\delta^{18}\text{O}_{\text{fluid}}$ and magmatic water contribution trends demonstrate the
64 presence of three intermittent pulses, and show a continuous drop in the flux of magmatic

65 derived fluid during ore formation. This figure is an expanded version of Figure 3 in the main
66 text.

67 **A2. Geological framework of Qulong**

68 Qulong is China's largest porphyry Cu-Mo deposit, containing 11 Mt of Cu and 0.6 Mt
69 of Mo (Li et al., 2017b). The deposit geology has been extensively studied and is summarized
70 here. The Qulong deposit occurs in the eastern Gangdese porphyry copper belt that hosts tens
71 of porphyry copper systems (Fig. DR1A). The deposit geology is dominated by Jurassic
72 volcanic/intrusive rocks and Miocene porphyries (Yang et al., 2009; Xiao et al., 2012; Hu et
73 al., 2015; Zhao et al., 2016; Li et al., 2017a; Li et al., 2017b). The order of emplacement of
74 the Mid-Miocene ore related intrusions is observed to be the pre-ore Rongmucuola
75 monzogranite pluton, syn-ore aplite and monzogranite P porphyry, monzogranite X porphyry,
76 and finally the post-ore quartz diorite (Fig. DR1B-C). In addition, a syn-ore hydrothermal
77 breccia pipe, comprising two stages of breccia, developed following the emplacement of the
78 X porphyry. High precision zircon U-Pb dating suggests that the Rongmucuola pluton, P
79 porphyry and quartz-diorite were emplaced at 17.142 ± 0.023 Ma (including ^{238}U decay
80 constant uncertainty), 16.009 ± 0.024 Ma and 15.166 ± 0.020 Ma, respectively (Li et al.,
81 2017a).

82 The Cu-Mo mineralization at Qulong is predominately hosted by the western
83 Rongmucuola pluton with a small portion of ore occurring in the Jurassic volcanic rocks (Fig.
84 DR1B). Mineralization is spatially associated with potassic and, to a lesser extent, with
85 propylitic and phyllic alteration assemblages (Yang et al., 2009; Li et al., 2017a; Li et al.,
86 2017b). The potassic alteration assemblages are characterized by K-feldspar and biotite in the
87 center of the deposit with the most intensive alteration spatially associated with the P
88 porphyry and the north-south fault. Propylitic alteration assemblages are widely distributed in

the western Rongmucuola pluton and overprint the potassic alteration assemblages, which are best illustrated by the selective replacement of biotite and plagioclase by chlorite bearing selvages. Phyllic alteration assemblages are represented by the pervasive replacement of feldspar by fine-grained sericite and ultra-fine granular quartz, and occur ubiquitous in the western Rongmucuola pluton and overprint most of the potassic and propylitic alteration assemblages.

The main ore minerals are chalcopyrite and molybdenite which occur predominately in veinlets, which are classified as A-type, B-type and D-type veins, with B-type veins being the most abundant and hosting the majority of the Cu-Mo mineralization (Yang et al., 2009; Li et al., 2017a). Previous high precision Re-Os geochronology (Li et al., 2017a) demonstrates that the majority of metals at Qulong were precipitated within 266 ± 13 kyrs between 16.126 ± 0.077 and 15.860 ± 0.075 Ma (including ^{187}Re decay constant uncertainty). Coupling the high precision dating, vein types and cooling history indicates that the mineralization process comprise three short-lived (38 - 59 kyrs) pulses between 16.126 ± 0.008 and 16.050 ± 0.005 Ma, 16.040 ± 0.007 and 15.981 ± 0.007 Ma, and $\sim 15.981 \pm 0.007$ and 15.860 ± 0.010 Ma. With vein formation ages and temperatures constrained by molybdenite Re-Os dating and fluid inclusion analysis (Li et al., 2017a; Li et al., 2017b), respectively, the cooling rate of the Qulong hydrothermal system has been constrained to be 0.55 ± 0.11 °C/kyr, with much faster cooling rates (~ 1.2 °C/kyr) for the individual mineralization pulses.

A3. Samples and analytical methods

A3.1 Samples

The selected sample-set were quartz-molybdenite veinlets which have been previously studied in detail; with formation ages and temperatures (Table DR1) constrained by high

precision molybdenite Re-Os dating and fluid inclusion studies (Li et al., 2017a; Li et al., 2017b). The formation ages of these veins ($n = 12$) were 16.126 ± 0.008 Ma, (2σ , analytical uncertainty), 16.107 ± 0.015 Ma, 16.098 ± 0.013 Ma, 16.088 ± 0.007 Ma, 16.050 ± 0.005 Ma, 16.040 ± 0.007 Ma, 16.036 ± 0.012 Ma, 16.011 ± 0.015 Ma, 15.981 ± 0.007 Ma, 15.943 ± 0.007 Ma, 15.939 ± 0.006 Ma, and 15.860 ± 0.010 Ma, with vein formation temperatures (trapping temperatures of fluid inclusion groups) of 425 ± 20 °C, 398 ± 20 °C, 390 ± 20 °C, 380 ± 20 °C, 360 ± 20 °C, 365 ± 20 °C, 365 ± 20 °C, 340 ± 20 °C, 290 ± 20 °C, 345 ± 20 °C, 310 ± 20 °C, 280 ± 20 °C, respectively.

Quartz grains closely associated with molybdenite grains from the parallel chips (Fig. DR3G) of the veinlets (Fig. 2B, Fig. DR3D) were recovered during the molybdenite separation process (Li et al., 2017a) for SIMS oxygen isotope analysis (Fig. DR3G). The textural association between the quartz and molybdenite (Fig. DR3D-E) and previous fluid inclusion study (Li et al., 2017b) suggested that these minerals were co-precipitated, which permit the linkage of the nature of the fluid recorded in the quartz to the timing of the ore (molybdenite)-forming process. To the best of our knowledge, this is the first approach evaluating the ore-forming fluid evolution path under an absolute timeframe.

A single sinusoidal quartz vein hosted by aplite (Fig. 2A, Fig. DR3A), which was formed before the main mineralization stage (Li et al., 2017b), was analyzed to evaluate the fluid nature prior to the main stage mineralization. Further, a total of 12 molybdenite bearing quartz veins (Table DR1) with Re-Os dates (Li et al., 2017a) were analyzed by CL and SIMS to yield successive generations of quartz and oxygen isotopic compositions, respectively. One vein (1605-211) from (Li et al., 2017a) was not analyzed due to the inability to conform the co-precipitated association between quartz grains and molybdenite samples.

To provide constraints on the oxygen isotopic composition of the original magmatic fluid during ore formation, magmatic zircon and quartz grains were separated from two fresh Rongmucuola monzogranite hand specimens, one from the west and the other from the east of the Rongmucuola pluton. The fresh samples from ore-bearing and barren parts of the pluton permitted the evaluation of potential variations of magmatic water oxygen isotopic composition during ore formation. The P porphyry was not used for this purpose as it has been intensive hydrothermally altered (Li et al., 2017a; Li et al., 2017b).

A3.2 Methods

Following the recovery during molybdenite separation, the quartz grains were cleaned and mounted with the NBS-28 and Qinghu quartz standards in epoxy resin, polished and then coated with gold (Li et al., 2017c). CL imaging was conducted with a Hitachi SU-70 SEM at Durham University (Li et al., 2017c). The imaging conditions were 15 kv, 100 nA and 20 mm for accelerating voltage, beam current and working distance, respectively, under a high vacuum mode. Secondary electron images were taken using the same conditions, but at a beam current of 1 nA.

Quartz SIMS oxygen isotope analysis was carried out at the Institute of Geology and Geophysics, Chinese Academy of Sciences with a CAMECA IMS-1280 SIMS (Li et al., 2010). The Cs^+ primary ion beam was accelerated at 10 kV, with an intensity of ca. 2 nA and rastered over a 20 μm area. The spot size as an ellipse was $\sim 10 \times 20 \mu\text{m}$ in diameter. Oxygen isotopes were measured using multi-collection mode on two off-axis Faraday cups, with the intensity of ^{16}O was typically 10^9 cps during this study. The instrumental mass fractionation factor (IMF) was corrected using NBS-28 with a $\delta^{18}\text{O}$ value (VSMOW) of 9.5 ‰ (Matsuhisa, 1974). Measured $^{18}\text{O}/^{16}\text{O}$ ratios were normalized to the VSMOW composition ($^{18}\text{O}/^{16}\text{O} = 0.0020052$), and then corrected for the IMF. The uncertainty on individual analysis was

usually better than 0.3 - 0.4 ‰ (2SE). The studied samples were run during the same period of a previous study (Li et al., 2017c), and the long-term drift during the course of study was monitored by analyzing the Qinghu quartz standard between samples. The Qinghu standard yields $\delta^{18}\text{O}$ values between 8.34 ± 0.23 and 8.92 ± 0.37 ‰ (Li et al., 2017c), with a mean of 8.64 ± 0.28 ‰ (2SD, n = 41). For each analysed quartz vein, CL images and the distribution of $\delta^{18}\text{O}$ values are used to evaluate the potential presence of recognizable multiple growth history. The analyzed positions are re-checked under a microscope, and analyses with the presence of inclusions or offset from the target, e.g., away from the growth zonation, are omitted. When single quartz growth history was demonstrated, e.g., no presence of dissolution/resorption textures and homogeneous $\delta^{18}\text{O}$ values (e.g., Fig. 3, see below for further details), it is interpreted that the weighted mean of the $\delta^{18}\text{O}$ values represents the best estimate of the $\delta^{18}\text{O}$ value of that vein.

Experimentally determined quartz-water oxygen isotope fractionation equation (equations 1 and 2) (Matsuhisa et al., 1979) was used to calculate the $\delta^{18}\text{O}$ values of ore-forming fluid ($\delta^{18}\text{O}_{\text{fluid}}$). For fractionation calculation, hydrothermal and magmatic quartz formation temperatures (T in unit of °C) were constrained by a previous quartz (molybdenite bearing quartz veins) fluid inclusion study (Li et al., 2017b) and quartz-zircon equilibrium oxygen isotope fractionation, respectively. Uncertainties of the $\delta^{18}\text{O}_{\text{fluid}}$ values were calculated with propagation of uncertainties from $\delta^{18}\text{O}_{\text{quartz}}$ measurements and fluid inclusion homogenization temperatures (± 20 °C) by Monte Carlo simulation.

$$\delta^{18}\text{O}_{\text{quartz}} - \delta^{18}\text{O}_{\text{fluid}} = 3.34 * (10^6 / (T + 273.15)^2) - 3.31 \quad 250\text{ °C} < T < 500\text{ °C} \quad (1)$$

$$\delta^{18}\text{O}_{\text{quartz}} - \delta^{18}\text{O}_{\text{fluid}} = 2.05 * (10^6 / (T + 273.15)^2) - 1.14 \quad 500\text{ °C} < T < 800\text{ °C} \quad (2)$$

To evaluate the percentage contribution (R) of magmatic water in the hydrothermal system, a binary mixing model with end members of meteoric water ($\delta^{18}\text{O}_{\text{meteoric}}$) and magmatic water ($\delta^{18}\text{O}_{\text{magmatic}}$), as presented in equation 3, was used.

$$\delta^{18}\text{O}_{\text{fluid}} = \delta^{18}\text{O}_{\text{magmatic}} * R + \delta^{18}\text{O}_{\text{meteoric}} * (1-R) \quad (3)$$

A4. Oxygen isotope equilibrium time between water and quartz

Here we further adjust the oxygen isotope equilibrium time between quartz and water. There is a lack of direct experimental studies to quantify the oxygen isotope exchange rate between quartz and water. Using the experimental study of a granite-fluid system (water/mineral ratio of 5, salinity of 5.8 wt. % NaCl equiv. and containing >22.6 wt. % quartz) as an alternative, with quartz of an average grain size of 0.4 cm (radius is 0.2 cm), oxygen isotope equilibrium can be achieved within 20 years (Fig. DR4B) at 300 °C (Cole et al., 1992). The experimental study further predicts a reduction in the equilibrium time with increased temperature, fluid salinity and water/mineral ratio.

A previous fluid inclusion study (Li et al., 2017b) suggests that the hydrothermal fluids at Qulong have formation temperatures of between 425 and 280 °C (average = 340 °C) with salinities of >5 wt. % NaCl equiv. (Table DR1). Considering the average Cu concentration in porphyry ore-forming fluid is ~0.1 wt.% (Landtwing et al., 2005; Kouzmanov and Pokrovski, 2012), the minimum volume of hydrothermal fluids to deposit the 11 Mt Cu at Qulong is $>1.1 \times 10^{10} \text{ m}^3$ assuming that the dissolved metals can be sufficiently and completely deposited. The Qulong deposit contains 2200 Mt of ore (Li et al., 2017b), which is equal to a volume of $\sim 8 \times 10^8 \text{ m}^3$. The vein volume density in porphyry deposits is suggested to vary between 5 and 10 % in the ore shells (Weis et al., 2012), as such the maximum quartz volume at Qulong is estimated to be $8 \times 10^7 \text{ m}^3$. The minimum fluid volume and maximum quartz

volume estimated above suggest a minimum water/quartz ratio of 10 for Qulong. As discussed above, the growth temperature, fluid salinity and water/mineral ratio of hydrothermal quartz grains at Qulong (425 – 280 °C, >5 wt. %, 10) are favorable for the vein quartz to reach oxygen equilibrium with fluid if their growth time is longer than 20 years.

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270

Figure DR1

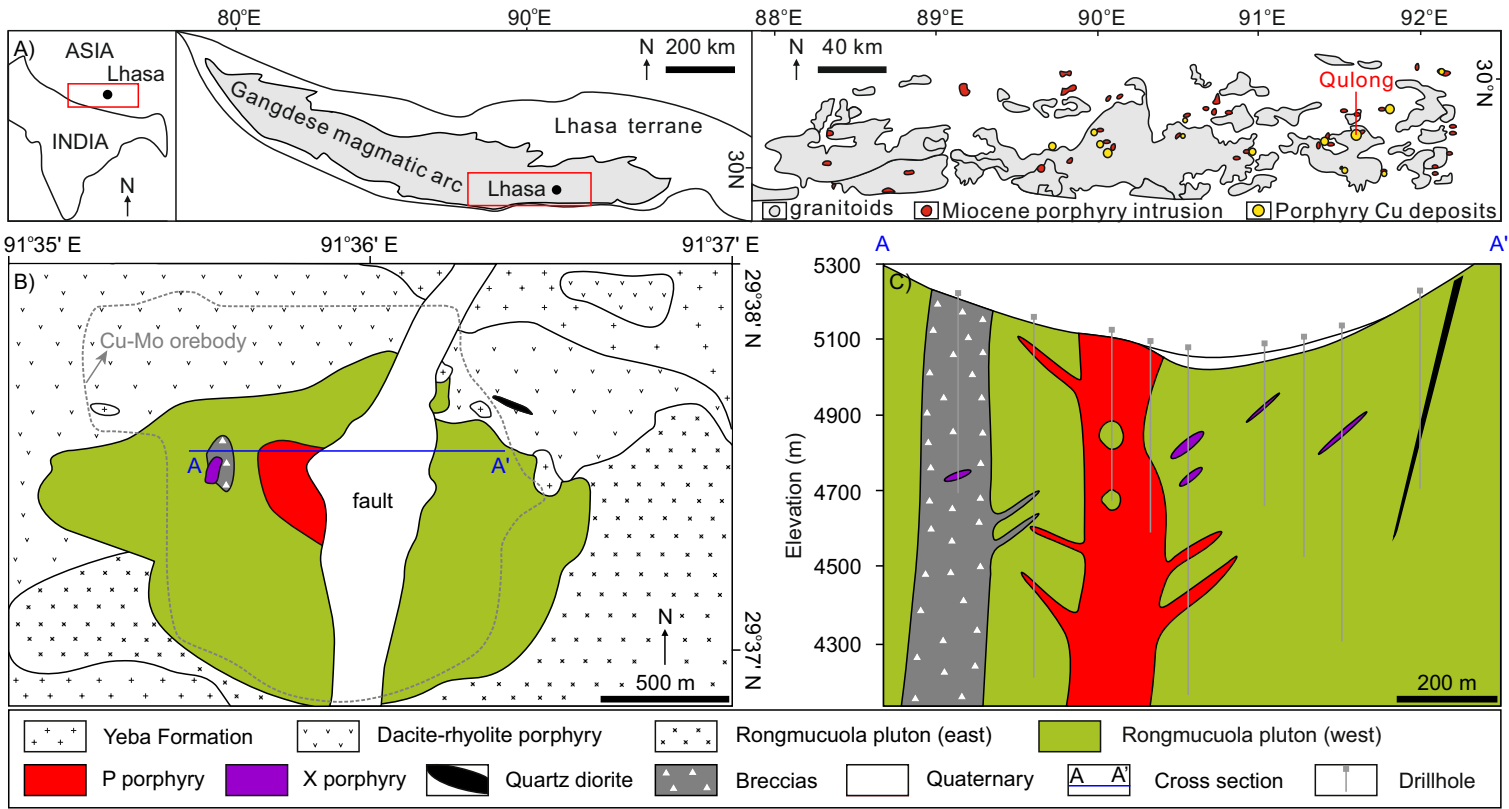


Figure DR2

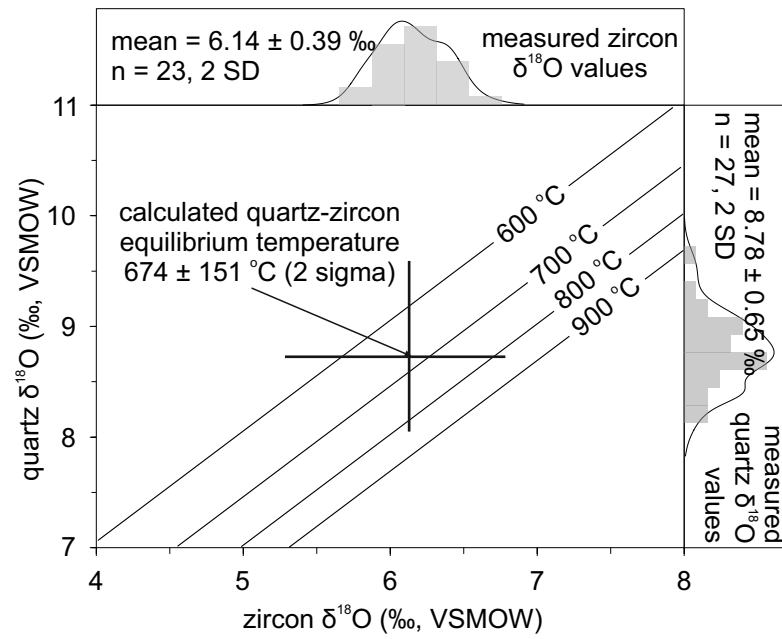


Figure DR3

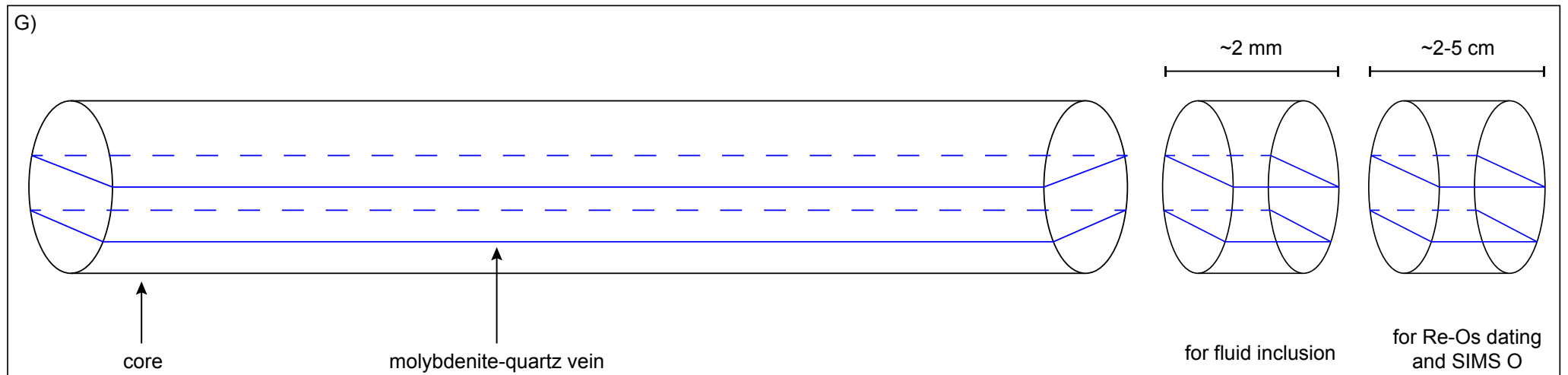
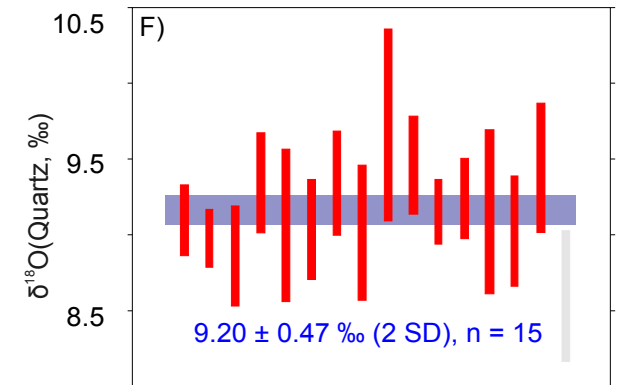
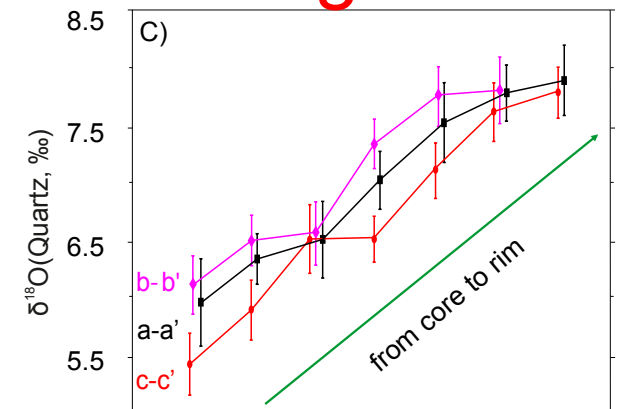
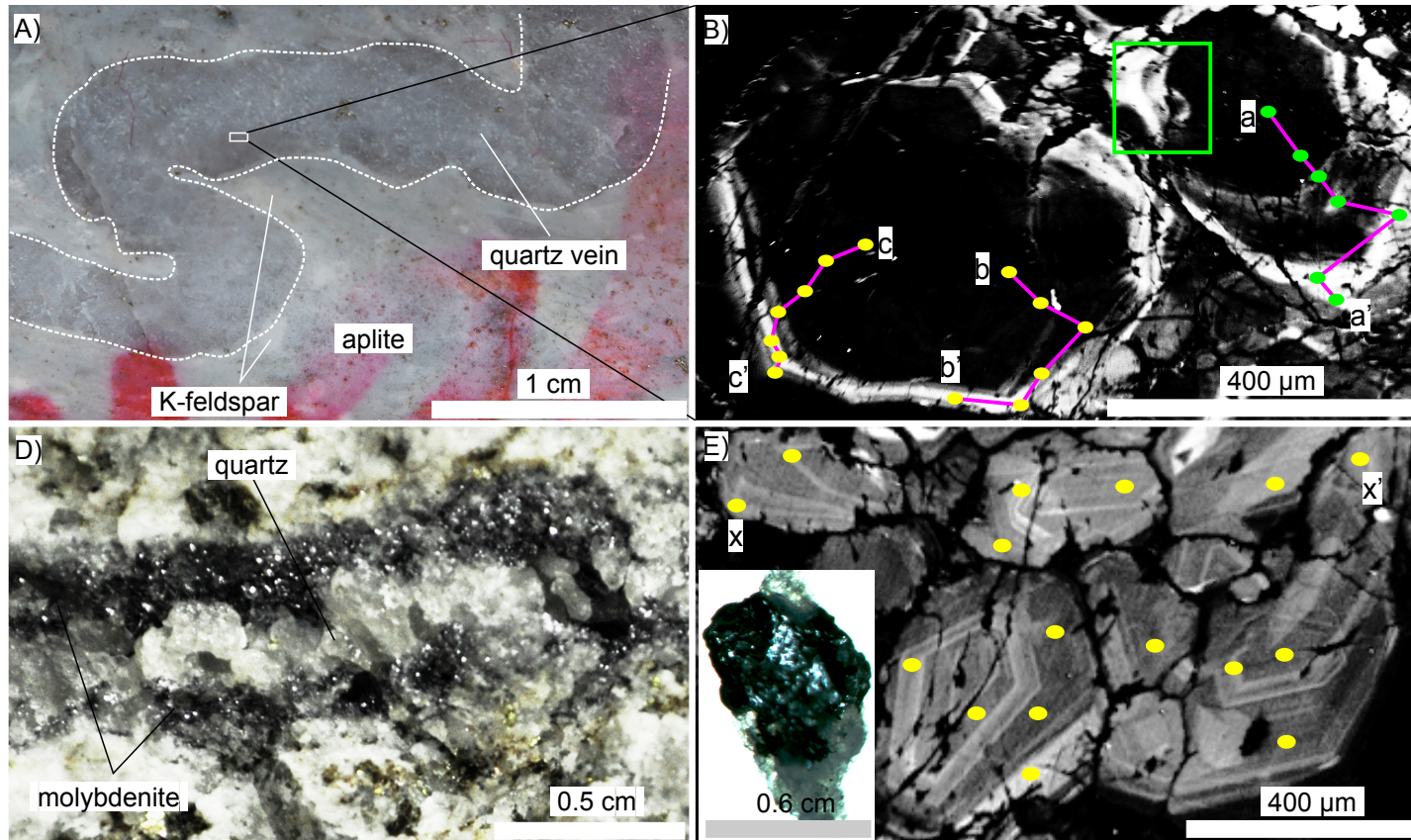


Figure DR4

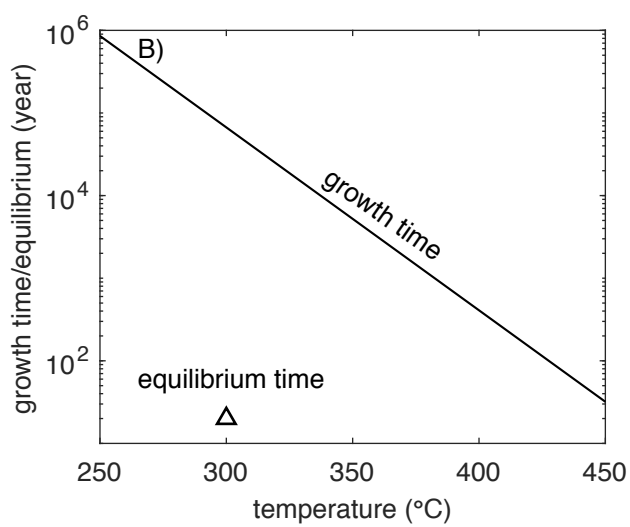
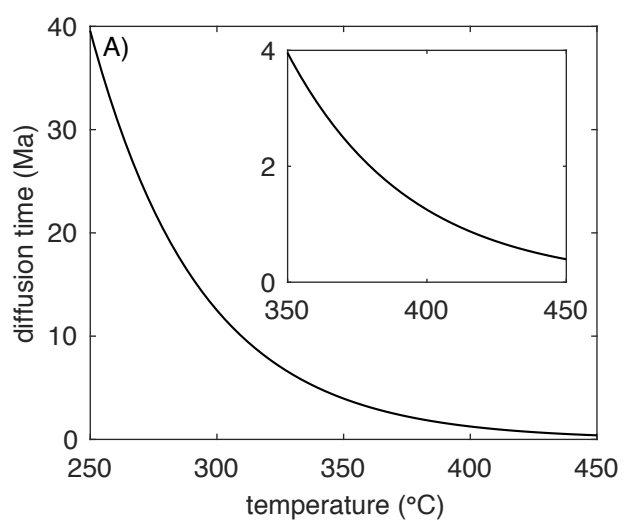


Figure DR5

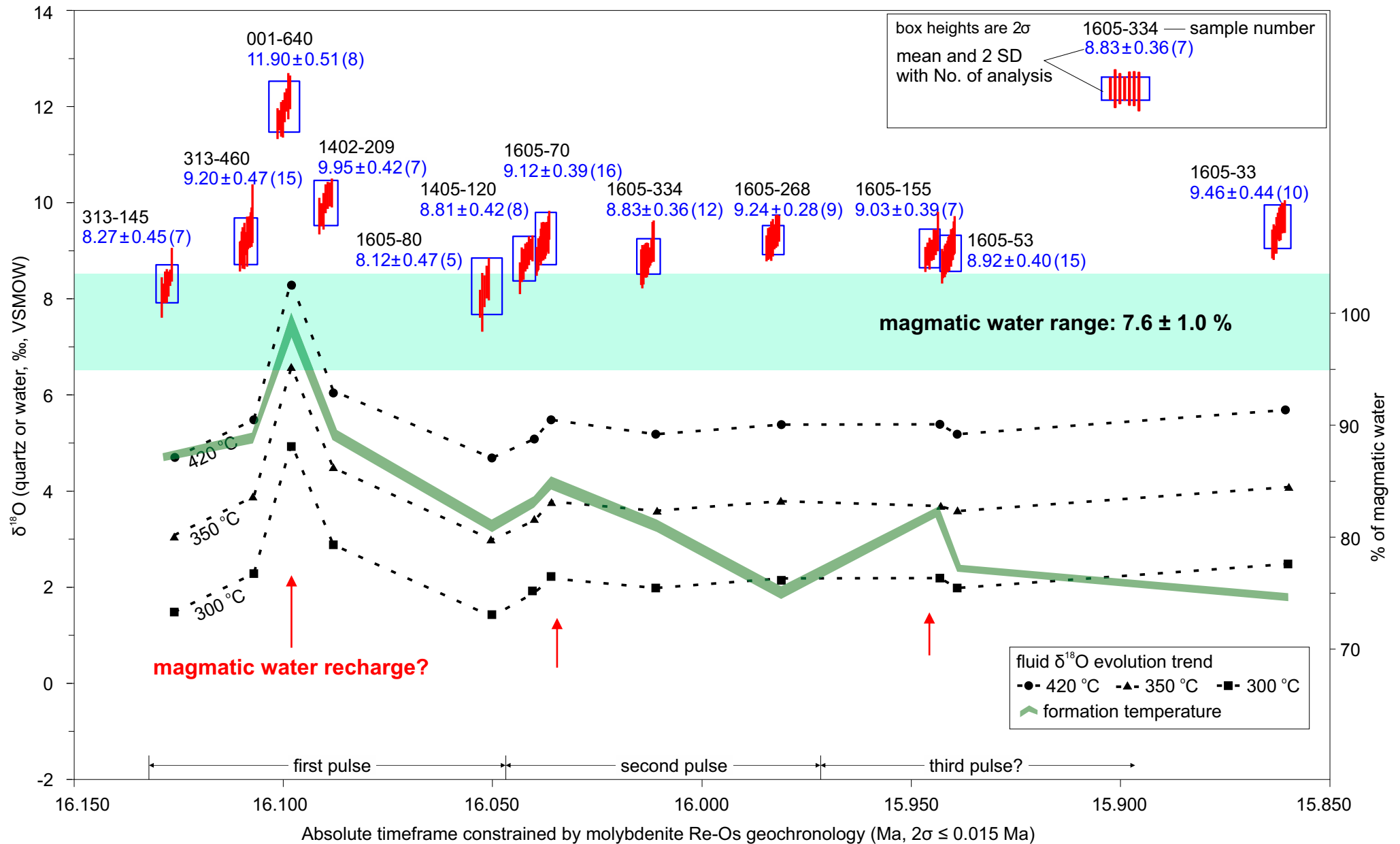


Table DR1. Oxygen isotopic composition of quartz grains enclosed by molybdenite, with estimations of fluid $\delta^{18}\text{O}$ and magmatic water contribution

Sample No.	Formation age ¹		Formation Temperature ²	$\delta^{18}\text{O}_{\text{quartz}}$		$\delta^{18}\text{O}_{\text{fluid}}$ with formation T and magmatic water % ³					$\delta^{18}\text{O}_{\text{fluid}}$ and magmatic % with upper, average and lower limit mineralization temperatures ⁴								
											420 ± 20 °C			350 ± 20 °C			300 ± 20 °C		
				$\delta^{18}\text{O}_{\text{fluid}}$		magmatic water %		$\delta^{18}\text{O}_{\text{fluid}}$		magmatic		$\delta^{18}\text{O}_{\text{fluid}}$		magmatic		$\delta^{18}\text{O}_{\text{fluid}}$		magmatic	
	Ma	2σ	°C	2σ	‰	2σ	‰	2σ	value	2σ	‰	2σ	water %	‰	2σ	water %	‰	2σ	water %
313-145	16.126	0.008	425	20	8.27	0.45	4.7	0.6	88%	5%	4.6	0.6	87%	3.0	0.7	80%	1.4	0.8	73%
313-460	16.107	0.015	398	20	9.20	0.47	5.1	0.6	89%	5%	5.6	0.6	91%	3.9	0.7	84%	2.3	0.9	77%
001-640	16.098	0.013	390	20	11.90	0.51	7.6	0.7	100%	5%	8.3	0.6	103%	6.6	0.8	96%	5.0	0.9	89%
1402-209	16.088	0.007	380	20	9.95	0.42	5.4	0.6	91%	5%	6.3	0.6	94%	4.7	0.7	87%	3.1	0.8	80%
1605-80	16.050	0.005	360	20	8.12	0.47	3.1	0.7	81%	6%	4.5	0.6	87%	2.8	0.7	79%	1.3	0.9	73%
1405-120	16.040	0.007	365	20	8.81	0.42	3.9	0.7	84%	5%	5.2	0.6	90%	3.5	0.7	82%	1.9	0.8	76%
1605-70	16.036	0.012	365	20	9.12	0.39	4.2	0.6	85%	5%	5.5	0.6	91%	3.8	0.7	84%	2.3	0.8	77%
1605-334	16.011	0.015	340	20	8.83	0.36	3.2	0.7	81%	6%	5.2	0.5	90%	3.5	0.7	82%	2.0	0.8	76%
1605-268	15.981	0.007	290	20	9.24	0.28	2.0	0.8	76%	6%	5.6	0.5	91%	3.9	0.6	84%	2.4	0.8	77%
1605-155	15.943	0.007	345	20	9.03	0.39	3.6	0.7	83%	5%	5.4	0.6	90%	3.7	0.7	83%	2.2	0.8	76%
1605-53	15.939	0.006	310	20	8.92	0.40	2.4	0.8	78%	6%	5.3	0.6	90%	3.6	0.7	83%	2.1	0.8	76%
1605-33	15.860	0.010	280	20	9.46	0.44	1.8	0.9	75%	6%	5.8	0.6	92%	4.2	0.7	85%	2.6	0.8	78%

1. Formation ages are constrained by molybdenite Re-Os geochronology from Li et al., 2017a.

2. Formation temperatures are constrained by trapping T of fluid inclusion group from the parellel wafers of the veins Li et al., 2017b.

3. Magmaic water and meteoric water $\delta^{18}\text{O}$ values are 7.6 ± 1.0 ‰ (this study) and -15.6 ± 3.6 ‰ (Currie et al., 2005), respectively.

4. To test the robustness of the temperatures used for oxygen isotope calculation, the uper, average and lower limit of mineralization temperatures represent mineralization temperature variations at Qulong and typical PCDs are used for testing.

Table DR2. SIMS $\delta^{18}\text{O}$ values of zircon and groundmass quartz of the Rongmucuola pluton

Sample NO.	$\delta^{18}\text{O}$	2σ	Mean and 2 SD (‰, VSMOW)	
<i>Zircon sample</i>				
QL14_23W@1	6.43	0.19	QL14_23W Mean = 6.14 ± 0.46 , n = 11.	
QL14_23W@2	6.00	0.18		
QL14_23W@3	5.82	0.16		
QL14_23W@5	5.92	0.28		
QL14_23W@6	6.44	0.21		
QL14_23W@7	6.37	0.17		
QL14_23W@8	5.80	0.26		
QL14_23W@9	6.10	0.26		
QL14_23W@10	6.28	0.25		
QL14_23W@11	6.16	0.32		
QL14_23W@12	6.19	0.29	QL14_23W + QL14_52E	
QL14_52E@1	6.17	0.26	QL14_52E Mean = 6.14 ± 0.39 , n = 23.	
QL14_52E@2	5.99	0.21		
QL14_52E@3	6.12	0.22		
QL14_52E@4	6.01	0.32		
QL14_52E@5	5.99	0.27		
QL14_52E@6	6.55	0.29		
QL14_52E@7	6.01	0.29		
QL14_52E@10	6.18	0.25		
QL14_52E@11	6.21	0.32		
QL14_52E@12	5.94	0.42		
QL14_52E@13	6.50	0.45		
QL14_52E@14	6.13	0.23		
QL14_52E@15	6.38	0.22		
<i>quartz sample</i>				
QL14_23W@1	8.49	0.36		QL14_23W Mean = 8.83 ± 0.83 , n = 12.
QL14_23W@3	9.12	0.22		
QL14_23W@4	8.95	0.29		
QL14_23W@5	8.45	0.29		
QL14_23W@8	9.05	0.33		
QL14_23W@9	9.99	0.22		
QL14_23W@10	8.74	0.25		
QL14_23W@11	8.87	0.30		
QL14_23W@12	9.38	0.46		
QL14_23W@13	8.79	0.30		
QL14_23W@14	9.56	0.31		
QL14_23W@15	8.40	0.16	QL14_23W + QL14_52E	
QL14_23W@16	8.16	0.36	QL14_52E Mean = 8.74 ± 0.48 , n = 15.	
QL14_52E@2	8.69	0.21		
QL14_52E@3	8.60	0.35		
QL14_52E@4	8.96	0.31		
QL14_52E@5	8.97	0.35		
QL14_52E@7	7.03	0.27		
QL14_52E@10	8.91	0.32		
QL14_52E@11	8.59	0.35		
QL14_52E@12	8.70	0.50		
QL14_52E@13	9.21	0.39		
QL14_52E@15	9.68	0.22		
QL14_52E@16	8.93	0.28		
QL14_52E@17	8.71	0.32		
QL14_52E@18	8.60	0.29		
QL14_52E@19	8.73	0.29		
QL14_52E@20	8.38	0.22		
QL14_52E@21	8.26	0.31		
QL14_52E@22	8.81	0.35	Mean = 8.78 ± 0.65 , n = 27.	

QL14_23W and QL14_52E are relative fresh samples from the west and east of the Rongmucuola pluton, respectively.

Data do not belong to the main population are rejected (with strikethrough) and are not used for mean value calculations

Table DR3. SIMS $\delta^{18}\text{O}$ values of hydrothermal quartz from Qulong

Sample No.	$\delta^{18}\text{O}$	2σ		
<i>Sinusoidal like quartz vein</i>				
<i>Section A</i>				
817-290@A1	5.95	0.38		
817-290@A2	6.33	0.22		
817-290@A3	6.50	0.33		
817-290@A4	7.01	0.25		
817-290@A5	7.52	0.35		
817-290@A6	7.78	0.24		
817-290@A7	7.89	0.31		
<i>Section B</i>				
817-290@B1	6.11	0.26		
817-290@B2	6.49	0.22		
817-290@B3	6.55	0.28		
817-290@B4	7.33	0.22		
817-290@B5	7.75	0.27		
817-290@B6	7.80	0.29		
<i>Section C</i>				
817-290@C1	5.40	0.27		
817-290@C2	5.88	0.26		
817-290@C3	6.49	0.30		
817-290@C4	6.50	0.20		
817-290@C5	7.10	0.24		
817-290@C6	7.61	0.26		
817-290@C7	7.78	0.22		
Molybdenite bearing veins				
Sample No.	$\delta^{18}\text{O}$	2σ	average	2 SD
313-145@1	8.09	0.20	8.27	0.45
313-145@2	8.41	0.17	n=7	
313-145@4	8.00	0.42		
313-145@5	8.28	0.26		
313-145@6	8.21	0.33		
313-145@9	8.24	0.35		
313-145@10	8.68	0.35		
313-460@1	9.10	0.22	9.20	0.47
313-460@2	8.99	0.19	n=15	

313-460@3	8.85	0.31		
313-460@4	9.36	0.32		
313-460@5	9.07	0.48		
313-460@6	9.05	0.32		
313-460@7	9.36	0.33		
313-460@8	9.03	0.43		
313-460@9	9.74	0.61		
313-460@10	9.46	0.31		
313-460@11	9.16	0.20		
313-460@12	9.25	0.25		
313-460@13	9.16	0.52		
313-460@14	9.03	0.34		
313-460@15	9.46	0.41		
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001-640@1	12.10	0.24	11.90	0.51
001-640@2	12.26	0.35	n=8	
001-640@4	11.68	0.21		
001-640@5	11.71	0.39		
001-640@7	12.18	0.48		
001-640@8	11.61	0.32		
001-640@9	11.95	0.31		
001-640@10	11.70	0.36		
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1402-209@1	10.17	0.30	9.95	0.42
1402-209@3	9.79	0.38	n=7	
1402-209@4	9.69	0.38		
1402-209@5	9.72	0.24		
1402-209@8	10.14	0.25		
1402-209@9	10.12	0.28		
1402-209@11	10.05	0.30		
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1605-80@2	7.87	0.29	8.12	0.47
1605-80@3	8.32	0.34	n=5	
1605-80@5	7.90	0.61		
1605-80@6	8.13	0.33		
1605-80@8	8.38	0.44		
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1405-120@1	8.69	0.32	8.81	0.42
1405-120@2	8.40	0.33	n=8	
1405-120@3	8.97	0.30		
1405-120@4	8.87	0.28		
1405-120@5	9.02	0.27		
1405-120@6	8.68	0.36		
1405-120@7	8.83	0.29		

1405-120@8	9.00	0.22		
1605-70@1	9.26	0.31	9.12	0.39
1605-70@2	9.28	0.25	n=16	
1605-70@3	8.97	0.24		
1605-70@4	9.50	0.30		
1605-70@5	9.01	0.26		
1605-70@6	8.75	0.30		
1605-70@7	8.90	0.33		
1605-70@8	9.21	0.34		
1605-70@9	9.16	0.40		
1605-70@10	9.05	0.38		
1605-70@11	9.18	0.34		
1605-70@12	9.18	0.38		
1605-70@13	9.17	0.24		
1605-70@16	8.78	0.24		
1605-70@17	9.17	0.40		
1605-70@18	9.30	0.39		
1605-334@1	8.76	0.34	8.83	0.36
1605-334@2	9.17	0.43	n=12	
1605-334@3	8.84	0.18		
1605-334@4	8.64	0.45		
1605-334@6	9.16	0.41		
1605-334@7	8.63	0.37		
1605-334@11	8.66	0.39		
1605-334@12	8.97	0.25		
1605-334@13	8.78	0.36		
1605-334@15	8.80	0.33		
1605-334@16	8.77	0.36		
1605-334@17	8.83	0.21		
1605-268@2	9.26	0.26	9.24	0.28
1605-268@3	9.17	0.41	n=9	
1605-268@4	9.13	0.33		
1605-268@7	9.37	0.34		
1605-268@8	9.43	0.27		
1605-268@9	9.39	0.33		
1605-268@10	9.11	0.33		
1605-268@11	9.02	0.27		
1605-268@13	9.24	0.39		
1605-155@1	9.37	0.41	9.03	0.39
1605-155@2	8.91	0.33	n=7	

1605-155@3	8.98	0.22		
1605-155@4	9.14	0.26		
1605-155@6	9.10	0.32		
1605-155@7	8.79	0.26		
1605-155@9	8.90	0.24		
1605-53@3	8.68	0.28	8.92	0.40
1605-53@4	8.92	0.34	n=15	
1605-53@5	8.82	0.23		
1605-53@6	8.92	0.37		
1605-53@7	9.23	0.35		
1605-53@8	9.05	0.28		
1605-53@9	9.30	0.39		
1605-53@10	8.70	0.25		
1605-53@12	9.11	0.33		
1605-53@13	9.03	0.39		
1605-53@14	8.80	0.32		
1605-53@15	8.75	0.33		
1605-53@16	8.65	0.36		
1605-53@17	8.85	0.33		
1605-53@19	9.00	0.30		
1605-33@1	9.15	0.37	9.46	0.44
1605-33@2	9.20	0.29	n=10	
1605-33@3	9.29	0.38		
1605-33@4	9.52	0.35		
1605-33@5	9.11	0.30		
1605-33@8	9.68	0.34		
1605-33@10	9.55	0.36		
1605-33@12	9.65	0.32		
1605-33@14	9.48	0.28		
1605-33@16	9.65	0.31		

Data do not belong to the main population or show evidence of mineral inclusions, or located at wrong targets after reexamination are rejected (with strikethrough), and thus are not used for mean values calculation.