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Episodic photic zone euxinia in the northeastern Panthalassic Ocean during the end-Triassic extinction

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1. Methods

1.1 Sample Collection

Samples were collected in the summer of 2004 at stratigraphic intervals of \sim 5 m in the upper Triassic and \sim 1 m in the lower Jurassic. The error in stratigraphic height of each sample was lower than 0.5m over the nearly 200 m section, delineated by a metal tape bolted to the outcrop.

1.2 Cleaning Protocol and Sample Preparation

Biomarker Sample Prep: Weathered edges were removed using a rock saw, followed by rinses with B-Pure water and acetone, and later sonicated in B-Pure water for 30 min to remove surface contamination. Samples were dried, crushed with a metal press, and pulverized to a fine power using an agate mortar and pestle. All laboratory equipment was rinsed with high-purity acetone, methanol (MeOH), and dichloromethane (DCM) between samples to avoid cross contamination. Powdered samples were then split for bulk elemental analysis and lipid extraction All glassware, aluminum foil, silica, quartz wool and quartz sand were combusted at 500°C for at least 12 hours to remove organic contamination; metal tools were rinsed in MeOH and DCM.

1.3 Nitrogen Isotope prep

A subsample of approximately 10 g selected for a lack of visible fracture fills and surface alteration was taken from each of the 51 samples, cleaned by sonication in ethanol for 1 hour, and in ultrapure deionized water (>18 M Ω) three consecutive times. Samples were powdered using an agate ball mill. Powders were acidified overnight at 40° C in an excess of 20 % HCl solution to remove carbonates, then rinsed three times with deionized water and dried overnight at 40° C.

1.4. Bulk Carbon analysis as a Check for Consistency with Previous Studies

Total carbon (TC) and the C isotopic composition of total organic carbon ($\delta^{13}C_{TOC}$) were measured on a Costech 4010 elemental analyzer system coupled to a Thermo Finnigan Delta-V-Plus Isotope Ratio Mass Spectrometer at Brown University. $\delta^{13}C_{TOC}$ values were compared to an existing record (Williford et al., 2007) to confirm that our samples were consistent with those used in previous studies. TC was measured in whole rock powders, whereas $\delta^{13}C_{TOC}$ was analyzed using decalcified samples. At least two replicates were run per sample. CaCO₃ was removed by reacting samples with 2N HCl at 60°C for 12 hours. Samples were rinsed with water until neutral pH was reached, and freeze-dried. Total inorganic carbon (TIC) was measured in duplicate using a UIC CM5012 CO₂ coulometer. Total organic carbon (TOC) values were derived from the difference between TC and TIC. TOC values derived in our lab were consistent with previous studies.

1.5. Lipid Extraction and Separation

Approximately 5-10 g of powdered rock were extracted with DCM: MeOH (9:1 v/v) using a Dionex ASE 250 accelerated solvent extraction system. Cells were packed with quartz filters and quartz sand. Before extraction, samples were spiked with 1 μ g of d4 C₂₉ $\alpha\alpha\alpha$ (20R)-ethylcholestane. The total lipid extract (TLE) was then concentrated, mixed with activated copper powder for 12 hours to remove elemental sulfur, and filtered through a pipette column packed with quartz wool to remove impurities. Copper was activated using 4N HCl for 1 hour, then rinsed with water to neutrality, and finally rinsed with MeOH and DCM (10x). Asphaltenes were precipitated from TLEs in 10-40 mL of hexane overnight at ~4°C and by centrifugation at 2,500 rpm for 30 minutes. The maltene fraction (supernatant) was pipetted out and collected, and the entire process of asphaltene precipitation was repeated three times. Maltenes (<5 mg) were then separated into three fractions using glass pipette columns filled with silica gel. The dead volume (DV) of each column was calculated by the addition of *n*-hexane. Aliphatic hydrocarbons, aromatic hydrocarbons, and polar compounds were eluted in *n*-hexane (3/8 DV), n-hexane:DCM (8:2 v:v, 4 DV), and DCM:MeOH (4:1 v:v, 4 DV), respectively. After separation, the

aromatic fraction was spiked with d4 $C_{29} \alpha \alpha \alpha$ (20R)-ethylcholestane, after ensuring a complete separation from the aliphatic fraction previously spiked as a TLE.

1.6. Gas Chromatography – Multiple Reaction Monitoring – Mass Spectrometry (GC–MRM–MS)

Aliphatic and aromatic hydrocarbons were analyzed in full scan by GC-MS and by GC-MRM-MS using predetermined precursor-product reactions (See Data Repository [DR] Table 1) on a Micromass AutoSpec Ultima mass spectrometer interfaced to an Agilent 6890 N gas chromatograph at Massachusetts Institute of Technology.

GC-MS Protocol for Aliphatic Hydrocarbons: GC was fitted with a DB-1 capillary column (60 m; 0.25 mm I.D.; 0.25 μ m film thickness; J&W Scientific) and He was used as the carrier gas. Temperature program: 60°C (2min) to 150°C at 10°C min⁻¹, to 315°C (held 24 min) at 3°C min⁻¹.

GC-MS Protocol for Aromatic Hydrocarbons: GC was fitted with a DB-5MS capillary column (60 m; 0.25 mm I.D.; 0.25 μ m film thickness; J&W Scientific) and He was used as the carrier gas. Temperature program was: 60°C (1 min) to 150°C at 15°C min⁻¹, to 325°C (held 39 min) at 4°C min⁻¹.

Autospec Parameters: The Autospec source was operated in electron ionization (EI, 70 eV) mode at 250° C, with 8 kV accelerating voltage for MRM. Full scan analyses were conducted over a range of m/z 50–600. Data were acquired and processed using MassLynx 4.0 (Micromass Ltd.).

Compound Identification: Identification of compounds was achieved by comparison with a synthetic mixture of oils (AGSO standard) that contains all known hopanes and steranes. For isorenieratane and aryl isoprenoids we used bitumens from the Cenomanian-Turonian boundary in Jordan (Sepúlveda et al., 2009) and the end-Permian extinction from Meishan, China (Cao et al., 2009), previously reported to contain these compounds. Biomarker concentrations were quantified by comparison with internal standards. Data Repository Figures 3-5 exhibit representative MRM chromatograms of the compounds analyzed in this study.

1.7. Nitrogen Isotopes and C:N Data

Organic carbon and nitrogen isotopes were measured via elemental-analyzer continuous-flow isotope ratio mass spectrometry (EA-CF-IRMS) at the University of Washington's IsoLab facility. 25-45 mg of decalcified powders weighed into tin capsules were combusted in a Costech ECS 4010 Elemental Analyzer, coupled to a ThermoFinnigan 253 mass spectrometer, through a ThermoFinnigan CONFLO III gas interface. Measurements were corrected using calibrated internal laboratory standards. All stable isotope results are presented in standard delta (δ) notation:

$$\delta = (R_{sample}/R_{std.}-1) \times 1000$$

The standard for N isotope measurements was atmospheric air. Analytical precision based on repeated measurement of standards was 0.34 ‰. Samples were measured in triplicate and presented here with 1 standard deviation error bars.

2. Biomarker Background

All biomarker indices used in the main text and in this Data Repository are listed in DR Table 3.

2.1. Sources of compounds

This study utilizes steranes (derived from eukaryotic sterols), hopanes (derived from bacterial hopanoids), and isorenieratane (derived from Chlorobi) and its diagenetic products (aryl isoprenoids).

Steranes: Sterols are produced by unicellular and multicellular eukaryotes, where algae are known to produce specific classes. C_{27} sterols are more abundant in red algae, C_{28} sterols typically derive from chlorophyll-c containing algae (dinoflagellates, coccolithophores, diatoms) and prasinophytes, and C_{29} sterols derive from green algae (Peters et al., 2005; Volkman et al., 1994, 1998; Kodner et al., 2008). Therefore the relative contribution of algal steranes and their diagenetic products can be used to investigate relative changes in planktonic communities over geological time scales and across critical boundary transitions (DR Table 2).

Interpretation of C_{28} steranes: This paper argues that the C_{28} sterol record is indicative of prasinophytes. Although green algae are characterized by synthesizing C_{29} sterols in abundance, C_{28} compounds are the dominant sterols in some prasinophytes (Kodner et al., 2008). The fossil and geochemical record suggests that the major radiation of chlorophyll-c producing algae such as diatoms, dinoflagellates and haptophytes, the main producers of C_{28} sterols, occurred later in the Jurassic and Cretaceous (Grantham and Wakefield, 1988; Volkman et al., 1998; Schwark and Empt, 2006; Knoll et al., 2007). This hypothesis is supported by palynological records indicating a predominance of prasinophytes in the Early Jurassic (Prauss, 2007).

Hopanes: C_{27} - C_{35} hopanes are derived from bacteriohopanepolyols, which are produced almost exclusively by bacteria, and include cyanobacteria, purple non-sulfur bacteria, acetic acid bacteria, methanotrophs, methylotrophs, nitrifying bacteria, planctomycetes, sulfate reducers, and metal-reducing bacteria (Summons et al., 1999; Sinninghe Damsté, et al., 2004; Blumenberg et al., 2006, 2009; Fischer et al., 2005; Härtner et al., 2005; Talbot and Farrimond, 2007). Some classes of hopanes have been tied to specific biosynthetic origins. 2-methyl hopanes, expressed in this study as the 2methylhopane index (2-MeHI, DR Table 3 equation 6), are produced mainly by cyanobacteria (Summons et al., 1999). 3-methyl hopanes, expressed here as the 3methylhopane index (3-MeHI, DR Table 3 equation 7), have been tied to methanotrophic bacteria (Peters et al., 2005). Isorenieratene: Isorenieratene is produced as a light-harvesting carotenoid by Chlorobi (Peters et al., 2005). Chlorobi are obligate anaerobic photosynthesizing bacteria, and rely on the presence of reduced sulfur species and sunlight (Peters et al., 2005). Therefore, the presence of isorenieratene and its diagenetic and catagenic products (isorenieratane and aryl isoprenoids) in ancient depositional environments have been widely used as indicators for the occurrence of photic zone euxinia (Cao et al., 2009; Peters et al., 2005; Richoz et al., 2012; Grice et al., 2005). Absolute concentrations were calculated by comparison with an injection standard and normalized to TOC.

2.2. Proxies for redox and water column stratification

The Homohopane Index: The homohopane index (HHI, DR Table 3 equation 1) quantifies the abundance of the C_{35} extended side-chain hopanes relative to all other homohopanes (Peters et al., 2005; Köster et al., 1997). Extended side-chain homohopanes (C_{31-35}), originate from functionalized C_{35} bacteriohopanepolyols and other C_{35} hopanoids common in bacteria (references in Köster et al., 1997). C_{35} is preferentially preserved in sediments under reducing conditions through sulfurization reactions, whereas it undergoes more efficient degradation under oxic conditions (Köster et al., 1997). Therefore, the relative contribution of C_{35} can be used as an estimation of the degree of oxygen deficiency in the depositional environment.

 C_{28} 28,30 dinorhopanes: C_{28} 28,30 dinorhopanes (DNH, DR Table 3 equation 2) are thought to be produced by chemoautotrophic bacteria that grow in the anoxic-oxic interface (Peters et al., 2005). High relative contributions of DNH relative to C_{30} hopane are well correlated with anoxic, clay-poor sediments (Cao et al., 2009; Peters et al., 2005).

Gammacerane index: The gammacerane index (GI, DR Table 3 equation 3) quantifies the relative contribution of gammacerane over C_{30} hopane, and is highly specific for marine and non-marine water-column stratification (Peters et al., 2005; Sinninghe Damsté et al., 1995). Gammacerane derives from the reduction of tetrahymanol (Peters et al., 2005; Sinninghe Damsté et al., 1995), a compound found in ciliates thriving in highly stratified water columns, commonly associated with oxygen-reduced environments (Peters et al., 2005).

3. Preservation of lipid biomarkers

3.1. Biodegradation

DR Figure 2 shows the total ion chromatogram (TIC) of the aliphatic hydrocarbon fraction of representative samples. The presence of an elevated chromatographic baseline and a large unresolved complex mixture (UCM), the absence of alkanes, and the occurrence of steranes and hopanes suggest that many of our samples have been subject to moderate to elevated biodegradation (equivalent to scale 3 or 4) (Peters et al., 2005). The use of GC–MRM–MS allows for accurate identification of compounds such as

hopanes and steranes, which are more resistant to biodegradation than alkanes and other aliphatic compounds (Peters et al., 2005). In order to avoid potential biases by varying biodegradation, we focused on biomarker ratios rather than in absolute concentrations, except for PZE markers.

3.2. Thermal Maturity

Thermal maturity: We estimate thermal maturity by determining the degree of isomerization undergone by steranes and hopanes during burial. The proportion between stereoisomers with the biological (thermally unstable) and geological (thermally stable) configuration can be compared to equilibrium end-point values.

Index (bio/geo)	Range of Utility	Average Values in
		Our Samples
$C_{30} \beta \alpha / \alpha \beta$ ratio	100% (immature) to 5% (early	10 <u>+</u> 1%
DR 1E	peak oil generation) (Peters et	
	al., 2005)	
C ₃₂ S/R ratio	0% (immature) to 60% (early oil	58 <u>+</u> 1%
DR 1F	generation) (Peters et al., 2005)	
C ₂₇ aaa S/R	0% (immature) to 55% (early oil	54 <u>+</u> 3%
DR 1G	generation) (Cao et al., 2009;	
	Köster et al., 1997)	

These indices suggest maturity levels in the early peak oil generation range (Ro% of about 0.55-0.70; Peters et al., 2005). However, the close clustering of the C_{32} S/R ratio around 60% suggests that our samples might be close to the equilibrium end-point. We interpret the $C_{30} \beta \alpha / \alpha \beta$ as most likely representative of thermal maturity, as the non-thermal maturity factors that influence it, terrestrial source and hypersaline depositional environments (Peters et al., 2005), are likely not a significant factor for this section (Haggart et al., 2001). Hopane, sterane, and aryl isoprenoid biomarkers are relatively more resistant to thermal alteration than other aliphatic biomarkers (Peters et al., 2005), and this level of thermal maturity should not significantly affect their use as environmental and ecologic indicators.

4. Time span represented by our record

Our record begins at the base of the Rhaetian stage, defined here as the extinction of the bivalve *Monotis* (Haggart et al., 2001), a boundary currently estimated to be at ~204.0 Ma (Gradstein et al., 2012). The top of the section is bracketed by the initiation of the "positive isotope excursion" (Williford et al., 2007), an event estimated at ~200.8 Ma through correlation to the Newark Basin (Whiteside et al., 2010). The latter correlation implies a duration of ~3.2 million years for our record (see DR Figure 6). A Triassic-Jurassic boundary at 201.4 Ma, such as proposed by correlation to the Newark basin (Whiteside et al., 2010), implies that the period of environmental and ecologic stress following the transition into the Jurassic lasted at least 600,000 years. However, evidence

from longer records suggest that ecologic stress lasted up to 2 million years after the ETE (Bartolini et al., 2012).

5. Notes on Figure 1

Selected studies are those recent studies that best highlight the variability of environmental change in the Tethyan realm, covering the late Rhaetian to late Hettangian.

Location	Info	Evidence	Source
St. Audrie's Bay	Nutrient cycle	Rise in prasinophytes after the	van de Schootbrugge
	deposition	Triassic-Jurassic boundary	et al., 2007
St. Audrie's Bay	Redox	Chlorobi-derived biomarkers of	Jaraula et al., 2013
	conditions,	episodic and persistent photic zone	
	nutrient	euxinia associated with	
	cycling	perturbations of the carbon and sulfur cycles	
Pinhay Bay	Euxinia	Pyrite framboids and faunal data	Wignall, 2001
Frick Swiss Jura	Euxinia,	Geochemical evidence for	Schwab &
	nutrient	isorieneratane and increased	Spangenberg, 2007
	cycling	cyanobacterial production in	
		middle-Hettangian samples	
Eiberg Basin	Redox	Carbon isotopes and palynology	Bonis et al., 2009
sections, Austria	conditions	suggest that terrigenous input	Bonis et al., 2010
D'I D '	NT / · /	drove anoxic bottom waters	D : 1 0000
Eiberg Basin	Nutrient	Widespread occurrence of	Bonis et al., 2009
sections, Austria	cycling	prasinophytes concurrent with excursions in the bulk carbon	Bonis et al., 2010
		isotope signature, supporting an	
		alteration to nutrient cycles	
Csovar, Hungary	Nutrient	Prasinophyte bloom concurrent	Haas et al., 2010
Covar, Hungary	cycling	with a negative carbon isotope	11aas et al., 2010
	eyening	excursion and trilete land spore	
		bloom	
Rosswinkel,	Redox	Shift from oxygenated shales in	Richoz et al., 2012
Luxembourg,	conditions,	the late Rhaetian to anoxic shales	,
	euxinia,	with abundant evidence for the	
Mingolsheim,	nutrient	occurrence of PZE in the	
Mariental Germany	cycling	Hettangian for Rosswinkel and	
		Mariental. Euxinic depositional	
		conditions coincided with	
		increased prasinophyte dominance,	
		consistent with a nutrient	
		disruption scenario.	
		The southern German site of	
		Mingolsheim, however, did not	
		present evidence for nutrient	
		disruption or euxinia.	

6. Notes on Figure 2

A single sample at 125.6 meters exhibited particularly elevated (relative to the entire record) values for the GI, HHI, 28,30 DNH, 2-MeHI, 3-MeHI, as well as an enhanced concentration of isorenieratane. We interpret these signals to represent an interval of particularly exacerbated environmental and ecologic disruption, likely at the peak of the environmental conditions that lead to extinction across the ETE. Contamination by hydrocarbon migration is an unlikely explanation for this change, taking into account the rather homogeneous thermal maturity and lithology of this part of the section. Anomalous TOC, TIC, or hopane and sterane concentrations cannot be invoked either (DR Figure 1A-E), as they are similar to other samples. The potential contribution of soil bacteria to the total pool of bacterial hopanes appears unlikely due to the lack of biomarkers indicative of terrestrial sources, such as an elevated contribution of C₂₉ steranes (Peters et al., 2005), or enhanced concentrations of polycyclic aromatic hydrocarbons (PAHs; data not shown).

Accurate determination of biomarker concentrations in the sample at 32.5 m was not possible due to the loss of sample during processing. However, ratios derived from MRM-GC-MS data are provided.

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Precursor mass (Da)	Product mass (Da)	Biomarkers
358	217	C ₂₆ Steranes
372	217	C ₂₇ Steranes
386	217	C ₂₈ Steranes
404	221	D ₄ C ₂₉ Internal std.
400	217	C ₂₉ Steranes
414	217	C ₃₀ Steranes
426	205	Me-C ₃₀ Hopanes
370	191	C ₂₇ Hopanes
384	191	C ₂₈ Hopanes
398	191	C ₂₉ Hopanes
412	191	C ₃₀ Hopanes
426	191	C ₃₁ -Homohopanes
440	191	C ₃₂ Homohopanes
454	191	C ₃₃ Homohopanes
468	191	C ₃₄ Homohopanes
482	191	C ₃₅ Homohopanes
246	134	C ₁₈ Aryl Isoprenoid
260	134	C ₁₉ Aryl Isoprenoid
274	134	C ₂₀ Aryl Isoprenoid
288	134	C ₂₁ Aryl Isoprenoid
302	134	C ₂₂ Aryl Isoprenoid
546	134	Isorenieratane

Data Repository Table 1. MRM-GC-MS precursor-product reactions

Data Repository Table 2. Sources of Steranes

Sterane Class	Most Common Source(s)
C ₂₆ Steranes	Typically digenetic products (Peters et al., 2005). Found to be produced by cold-water diatoms (Rampen et al., 2007).
C ₂₇ Steranes	
	Red Algae (Peters et al., 2005; Volkman et al., 1998)
C ₂₈ Steranes	Chlorophyll-C Containing Algae
	(Peters et al., 2005; Volkman et al., 1998):
	Dinoflagellates
	Coccolithophores
	Diatoms
	Prasinophytes (Volkman et al., 1994, Kodner et al., 2008)
C ₂₉ Steranes	Green Algae, including terrestrial plants (Kodner et al., 2008)
C ₃₀ Steranes	Chrysophytes (Peters et al., 2005)

Data Repository Table 3. Calculation of indices and ratios. The compounds in the equations refer to the integrated area of the peak in the GC-MRM-MS trace.

	$HHI = \frac{C_{35}\alpha\beta S + R}{\sum C_{35}\alpha\beta S + R} \bullet 100$
1. Homohopane Index	$\sum c_{31-35} \alpha \beta \beta + R$
2. C ₂₈ 28,30 Bisnorhopane Index	DNH = $\frac{C_{28}28,30 \text{ dinor}}{C_{28}28,30 \text{ dinor} + C_{30}\alpha\beta}$
3. Gammacerane Index	$GI = \frac{C_{30}Gamma}{C_{30}Gamma + C_{30}\alpha\beta} \bullet 100$
4. C ₂₆ -C ₃₀ Steranes : C ₂₇ -C ₃₅ Hopanes	Ster : Ster + Hop = $\frac{\sum \text{Steranes (reg)}}{\sum \text{Steranes (reg) + Hopanes (reg)}}$
5. C _{xx} Steranes	C_{xx} Steranes = $\frac{\sum C_{xx}$ Steranes (all) $\sum C_{26-30}$ Steranes (all)
6. 2-Methyl Hopane Index	2 - MeHI = $\frac{C_{31}2\alpha - MeH}{C_{30} \alpha\beta + C_{31}2\alpha - MeH} \bullet 100$
7. 3-Methyl Hopane Index	$3 - \text{MeHI} = \frac{C_{31} 3\beta - \text{MeH}}{C_{30} \alpha\beta + C_{31} 3\beta - \text{MeH}} \bullet 100$ $C_{30} \beta\alpha / \alpha\beta = \frac{C_{30} \beta\alpha}{C_{30} \beta\alpha + C_{30} \alpha\beta} \bullet 100$
8. C ₃₀ βα/αβ	$C_{30}\beta\alpha/\alpha\beta = \frac{C_{30}\beta\alpha}{C_{30}\beta\alpha + C_{30}\alpha\beta} \bullet 100$
9. C ₃₂ S/R	$C_{32}S/R = \frac{C_{32}\alpha\beta S}{C_{32}\alpha\beta S + C_{32}\alpha\beta R} \bullet 100$
10. C ₂₇ ααα S/R	$C_{27}\alpha\alpha\alpha S/R = \frac{C_{27}\alpha\alpha\alpha S}{C_{27}\alpha\alpha\alpha S + R} \bullet 100$
11. Ts/Tm	$Ts/Tm = \frac{Ts}{Ts + Tm} \bullet 100$
12. C ₂₉ Dia/Reg	$C_{29}\text{Dia}/\text{Reg} = \frac{C_{29}\text{Dia} \text{ S} + C_{29}\text{Dia} \text{ R}}{C_{29} \text{ (Dia} + \text{Reg, all)}}$

								Sterane	Sterane		
Stratigraphic	KPF	Homohopane	Gamma/	28,30 BNH/	ug/g TOC	ug/g TOC	C26-30Ster/	C28/	C29/		
Position	Sample No.	Index	(C30+Gamma)	(28,30 BNH + C30)	[C18-22]	[ISO]	Hop+C26-C30	(C26-C30)	(C26-C30)	C31 2-MHI (%)	C31 3-MHI
7.6	-270	5.1	1.7	19.1	65.18	0.00	39.9	23.4	36.6	1.11	1.3
15.3	-250	6.5	1.8	15.6	19.12	0.00	36.9	21.4	42.7	4.14	1.6
23.1	-230	5.5	0.6	10.4	75.15	0.00	35.0	18.6	50.4	2.04	1.4
27.0	-220	6.0	1.1	11.9	40.64	0.00	36.6	20.9	44.0	1.40	1.6
30.9	-210	5.6	1.0	11.9			41.0	20.0	44.8	1.08	1.8
38.6	-190	4.7	0.7	9.7	16.85	0.00	37.0	18.7	45.0	1.09	1.9
42.4	-180	5.1	1.1	11.2	40.43	0.00	34.4	19.6	44.3	1.68	1.8
49.7	-160	5.0	1.6	10.2	87.44	0.00	32.2	19.9	45.3	1.08	1.6
68.1	-110	6.1	1.2	7.9	63.94	0.00	32.6	18.8	45.3	1.83	2.1
95.1	-49.8	4.7	1.8	11.3	37.08	0.31	22.0	18.7	51.0	1.48	2.3
96.9	-46.5	3.6	0.2	8.3	37.92	0.00	23.7	16.1	54.1	0.55	1.8
99.3	-41.9	3.9	0.6	15.1	24.79	0.00	26.3	17.9	51.9	0.77	1.7
100.8	-39.1	4.0	1.9	4.9	25.80	2.19	24.2	17.9	49.6	0.97	1.8
103.0	-35	8.2	2.4	22.4	8.09	0.00	31.2	21.3	45.0	2.76	1.7
105.1	-31	3.4	0.3	5.1	35.15	0.00	28.8	17.3	51.0	1.02	2.5
114.2	-14	14.8	0.7	12.2	84.13	20.13	18.4	20.9	40.9	6.17	2.2
115.2	-12	15.6	1.4	26.2	197.00	10.35	21.5	20.3	42.3	13.36	4.6
116.5	-9.5	7.7	2.1	18.6	78.58	0.43	24.3	21.1	42.2	2.70	2.4
117.4	-7.6	7.9	1.7	14.4	55.20	0.00	22.5	19.1	43.9	4.04	2.8
118.5	-5.3	8.7	1.5	15.1	37.23	0.00	23.7	20.6	44.3	4.28	2.3
119.0	-4.3	7.0	1.6	16.2	87.45	0.00	25.9	21.0	42.2	3.13	2.0
120.5	-1	7.4	1.3	14.9	7.65	0.00	24.9	21.4	45.3	1.55	1.7
121.9	2.8	13.0	2.1	28.6	67.19	5.28	29.6	22.2	45.8	4.12	3.3
122.5	5	7.5	0.9	26.9	101.78	1.04	23.4	22.4	47.1	1.28	2.2
123.2	7.4	8.1	1.4	30.8	168.40	2.96	24.4	21.7	47.5	2.22	2.4
123.9	10.1	15.2	1.5	26.4	70.42	4.83	22.7	19.9	43.5	13.05	4.7
124.7	13.2	10.2	1.9	31.3	136.86	7.22	26.5	21.4	46.8	2.74	1.7
125.1	15	9.3	1.6	32.8	217.51	6.21	19.6	21.5	48.3	1.61	2.2

Data Repository Table 4. Biomarker Data for Figure 2

125.6	17.3	20.3	9.4	43.8	279.49	86.03	22.2	19.5	44.0	28.68	13.4
126.4	21	6.7	1.9	25.6	6.52	0.00	30.4	21.4	46.1	1.18	1.6
126.6	22	9.8	1.8	25.0	244.22	6.95	29.9	21.9	47.9	2.36	1.9
126.9	23.6	12.3	3.0	29.9	158.71	8.38	28.1	22.0	45.4	3.45	2.2
127.1	24.5	10.4	1.9	26.5			26.6	21.4	46.2	2.49	1.8
127.5	26.8	9.0	1.7	22.1	61.31	1.23	27.7	23.0	46.2	2.73	1.8
128.2	30.5	12.3	2.3	18.0	154.24	15.79	32.8	20.8	45.2	4.52	3.4
128.5	32.2	10.2	1.3	23.0	40.10	8.33	38.1	21.1	43.3	9.14	4.4
128.9	34.6	9.7	2.3	22.1	32.25	4.03	32.0	22.0	43.9	5.14	3.5
130.3	42.7	9.2	2.2	21.1	64.26	3.64	28.8	21.1	43.1	8.78	4.1
130.6	44.1	6.5	0.9	22.8	102.21	1.63	23.8	21.5	46.5	2.20	1.4
131.2	47.9	9.5	1.8	17.5	134.59	3.84	31.9	21.9	44.1	0.86	1.3
133.7	60.4	4.0	0.3	14.7	0.00	0.00	24.4	15.2	52.4	0.54	2.3
134.6	64.3	5.2	1.1	25.0	151.54	0.00	21.9	21.2	46.5	3.15	1.8
136.0	70.4	6.7	0.7	10.7	4.30	0.00	28.0	18.6	47.0	2.15	1.7
139.6	82.9	9.1	1.7	5.5	190.56	3.44	28.4	22.0	45.3	2.79	1.8
139.7	83.2	5.2	0.6	12.5	78.31	0.00	34.9	19.4	47.1	1.24	1.7
140.5	85.5	8.9	1.4	17.1	50.77	0.57	27.5	22.9	45.0	1.96	1.3

Data Repository Table 5. N Isotopic Data for Figure 2

	δ15N vs	
Sample No.	Air N2 (permil)	SD
KPF -270	2.32	0.11
KPF -250	0.25	0.10
KPF -240	0.76	0.19
KPF -230	1.31	0.22
KPF -220	2.24	0.12
KPF -210	4.16	0.09
KPF -200	3.88	0.03
KPF -190	2.56	0.43
KPF -180	3.78	0.25
KPF -160	3.68	0.39
KPF -150	2.82	0.18
KPF -120	4.17	0.11
KPF -110	3.92	0.10
KPF -49.8	3.20	0.14
KPF -46.5	0.76	0.75
KPF -41.9	-0.23	0.45
KPF -39.1	-0.40	0.10
KPF -35	1.01	0.11
KPF -31	0.31	0.20
KPF -29.1	-1.29	0.40
KPF -14	0.84	0.22
KPF -12	0.88	0.02
KPF -9.5	2.24	0.11
KPF -7.6	2.04	0.05
KPF -5.3	-0.10	0.30
KPF -4.3	2.00	0.08
KPF -1	0.70	0.23
KPF 2.8	0.90	0.13
KPF 5.0	0.68	0.14
KPF 7.4	1.71	0.24
KPF 10.1	0.03	0.54
KPF 13.2	-0.48	1.13
KPF 15	1.45	0.02
KPF 17.3	0.68	0.70
KPF 21	0.27	0.45
KPF 22	0.96	0.29
KPF 23.6	2.37	0.06
KPF 24.5	0.75	0.14
KPF 26.8	1.51	0.10

KPF 30.5	0.62	0.14
KPF 32.2	0.85	0.26
KPF 34.6	0.30	0.17
KPF 42.7	-0.75	0.10
KPF 44.1	0.50	0.14
KPF 47.9	-0.23	0.22
KPF 60.4	-0.40	1.06
KPF 64.3	0.68	0.28
KPF 70.4	1.18	0.10
KPF 82.9	1.59	0.12
KPF 83.2	2.39	0.54
KPF 85.5	1.21	0.04

KPF	Stratigraphic			ug/g TOC	ug/g TOC
Sample No.	Position	%C Inorg	%C Org	[hopanes]	[steranes]
-270	7.6	2.420	9.903	2.05	23.41
-250	15.3	5.099	3.371	66.31	42.16
-230	23.1	1.920	1.194	453.11	275.81
-220	27.0	0.108	2.244	152.62	97.04
-210	30.9	0.300	2.933	188.73	141.94
-190	38.6	0.759	1.154	413.69	261.73
-180	42.4	0.147	3.110	437.07	253.94
-160	49.7	0.140	1.716	595.81	300.91
-110	68.1	0.277	3.786	298.34	165.08
-49.8	95.1	0.302	2.216	397.65	141.35
-46.5	96.9	0.151	0.733	422.86	160.60
-41.9	99.3	0.501	0.444	92.76	38.66
-39.1	100.8	0.001	0.151	315.94	121.36
-35	103.0	0.426	0.975	132.71	63.66
-31	105.1	0.232	0.369	301.65	140.60
-14	114.2	0.322	2.434	262.40	103.32
-12	115.2	0.700	1.885	38.80	17.28
-9.5	116.5	0.385	9.457	50.98	21.82
-7.6	117.4	0.000	4.061	80.75	31.49
-5.3	118.5	0.528	2.192	23.64	9.69
-4.3	119.0	0.036	5.824	76.11	34.46
-1	120.5	0.600	4.014	130.27	54.49
2.8	121.9	1.121	1.624	50.69	28.36
5	122.5	0.401	2.354	304.15	124.56
7.4	123.2	0.779	2.246	315.13	138.10
10.1	123.9	0.785	2.161	43.25	20.09
13.2	124.7	0.545	1.439	358.98	158.89
15	125.1	0.872	2.508	367.23	126.71
17.3	125.6	0.083	1.810	315.52	114.37
21	126.4	0.124	0.139	375.55	185.02
22	126.6	0.438	0.611	426.22	208.65
23.6	126.9	0.154	4.770	202.04	93.67
24.5	127.1	0.627	2.138		
26.8	127.5	0.385	1.556	371.20	167.30
30.5	128.2	0.211	1.557	302.23	166.46
32.2	128.5	1.040	1.576	42.29	28.74
34.6	128.9	0.742	1.827	286.71	161.37
42.7	130.3	0.376	1.049	137.72	69.09
44.1	130.6	0.440	1.845	369.31	138.60
47.9	131.2	0.006	0.403	313.04	158.68
60.4	133.7	0.366	0.655	471.49	197.76
64.3	134.6	0.309	2.001	106.62	36.86
70.4	136.0	1.271	3.257	238.49	101.11

Data Repository Table 6. Carbon, Biomarker, and Isotopic data for DR Figure 1 and DR Figure 6.

82.9	139.6	0.292	6.411	213.17	98.87
83.2	139.7	0.416	1.026	379.51	225.68
85.5	140.5	3.150	2.490	307.71	134.36

KPF	Stratigraphic	Hopane	Hopane	Sterane	Hopane	Sterane
Sample No.	Position	C30 ba/ab	C32 S/R	C27 Reg aaa S/R	Ts/Tm	C29 Dia/Reg
-270	7.6	6.5	56.5	53.7	50.3	11.8
-250	15.3	6.9	57.8	54.8	49.6	18.5
-230	23.1	7.3	57.6	55.8	41.6	17.3
-220	27.0	7.4	57.5	54.4	39.3	17.0
-210	30.9	8.0	57.5	54.3	27.5	14.3
-190	38.6	8.7	57.9	53.6	24.9	13.6
-180	42.4	8.4	57.9	54.7	25.6	16.1
-160	49.7	8.9	57.6	52.6	19.3	12.8
-110	68.1	8.5	58.4	55.7	22.6	19.3
-49.8	95.1	11.4	58.5	53.5	22.1	26.0
-46.5	96.9	9.3	58.6	52.0	27.5	21.5
-41.9	99.3	9.5	58.7	53.1	18.1	17.7
-39.1	100.8	8.8	57.9	49.7	26.9	20.4
-35	103.0	8.4	58.0	54.0	22.1	12.1
-31	105.1	9.4	57.6	54.8	23.5	17.9
-14	114.2	8.7	53.9	63.1	38.1	57.1
-12	115.2	13.4	56.3	57.7	35.5	50.0
-9.5	116.5	9.7	58.1	57.3	33.0	32.5
-7.6	117.4	9.4	57.9	57.2	33.2	33.0
-5.3	118.5	10.5	57.0	56.5	35.4	32.4
-4.3	119.0	9.5	58.1	56.5	33.8	29.6
-1	120.5	10.1	58.2	56.5	28.6	26.8
2.8	121.9	12.2	57.2	53.6	22.7	33.0
5	122.5	10.6	58.1	53.9	19.6	30.3
7.4	123.2	11.1	56.9	55.0	21.0	32.1
10.1	123.9	12.5	56.0	58.7	35.3	47.5
13.2	124.7	10.3	57.7	54.8	21.0	25.5
15	125.1	10.2	58.3	55.4	20.2	35.3
17.3	125.6	12.1	56.0	60.6	36.2	43.8
21	126.4	9.5	58.2	48.1	20.6	17.9
22	126.6	10.0	58.4	52.9	23.0	19.1
23.6	126.9	9.7	58.6	54.5	21.9	23.2
24.5	127.1	10.2	58.1	54.3	21.4	22.1
26.8	127.5	10.6	58.3	53.8	24.0	22.0
30.5	128.2	9.9	57.9	55.3	33.0	24.2
32.2	128.5	11.3	56.6	52.9	37.2	25.4
34.6	128.9	10.6	57.3	53.7	35.2	27.3
42.7	130.3	10.6	57.5	55.4	32.1	34.3
44.1	130.6	10.3	57.7	53.8	23.1	22.3
47.9	131.2	8.5	58.7	54.2	28.7	14.6

60.4	133.7	8.8	58.5	54.4	29.4	26.5
64.3	134.6	9.7	57.2	52.5	29.5	27.4
70.4	136.0	9.2	57.9	53.7	22.0	14.2
82.9	139.6	9.7	57.5	53.3	23.5	21.1
83.2	139.7	8.3	57.1	54.4	22.7	14.9
85.5	140.5	10.0	57.4	53.0	20.7	19.2

KPF	Stratigraphic	Sterane	Sterane	Sterane	Sterane	Sterane
Sample No.	Position	C26/(C26-30)	C27/(C26-30)	C28/(C26-C30)	C29/(C26-C30)	C30/(C26-C30)
-270	7.6	3.7	32.7	23.4	36.6	3.7
-250	15.3	3.2	26.7	21.4	42.7	6.0
-230	23.1	2.2	24.3	18.6	50.4	4.5
-220	27.0	3.0	27.6	20.9	44.0	4.5
-210	30.9	2.7	28.2	20.0	44.8	4.3
-190	38.6	2.3	30.0	18.7	45.0	4.0
-180	42.4	2.6	28.5	19.6	44.3	5.0
-160	49.7	2.0	28.4	19.9	45.3	4.3
-110	68.1	3.5	26.4	18.8	45.3	6.0
-49.8	95.1	2.3	24.0	18.7	51.0	4.0
-46.5	96.9	1.6	25.4	16.1	54.1	2.9
-41.9	99.3	2.1	25.4	17.9	51.9	2.7
-39.1	100.8	1.8	27.8	17.9	49.6	3.0
-35	103.0	2.0	27.1	21.3	45.0	4.7
-31	105.1	2.1	26.2	17.3	51.0	3.3
-14	114.2	6.6	21.6	20.9	40.9	9.9
-12	115.2	7.0	21.8	20.3	42.3	8.6
-9.5	116.5	5.7	25.7	21.1	42.2	5.3
-7.6	117.4	5.1	24.3	19.1	43.9	7.6
-5.3	118.5	4.8	23.6	20.6	44.3	6.6
-4.3	119.0	5.5	25.8	21.0	42.2	5.6
-1	120.5	4.1	24.2	21.4	45.3	5.0
2.8	121.9	3.6	22.3	22.2	45.8	6.1
5	122.5	2.6	23.5	22.4	47.1	4.4
7.4	123.2	2.7	23.1	21.7	47.5	4.9
10.1	123.9	6.7	21.5	19.9	43.5	8.4
13.2	124.7	2.3	23.4	21.4	46.8	6.0
15	125.1	3.1	22.0	21.5	48.3	5.1
17.3	125.6	4.2	19.1	19.5	44.0	13.2
21	126.4	1.8	27.2	21.4	46.1	3.5
22	126.6	2.3	22.3	21.9	47.9	5.6
23.6	126.9	2.7	23.1	22.0	45.4	6.7
24.5	127.1	2.5	24.0	21.4	46.2	5.9
26.8	127.5	2.4	23.6	23.0	46.2	4.9
30.5	128.2	3.0	24.2	20.8	45.2	6.8
32.2	128.5	4.3	24.4	21.1	43.3	6.9
34.6	128.9	4.4	22.7	22.0	43.9	7.0

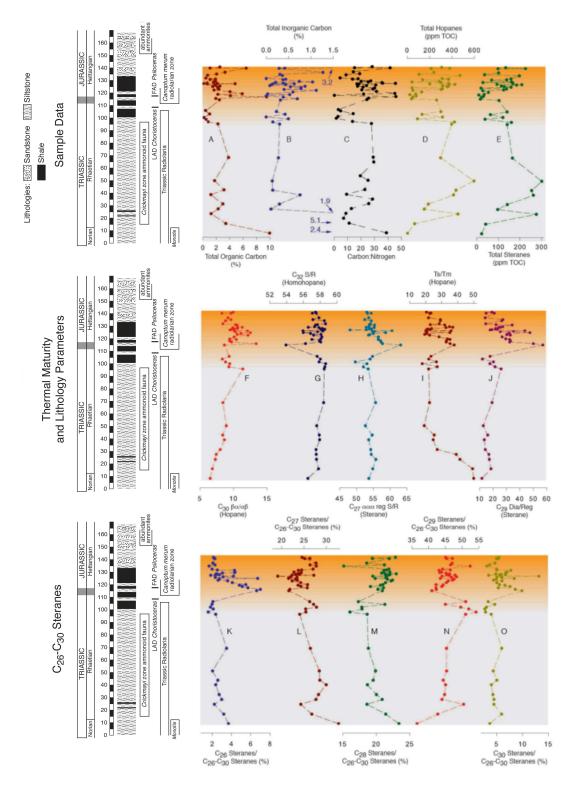
42.7	130.3	4.3	23.7	21.1	43.1	7.7
44.1	130.6	2.5	24.4	21.5	46.5	5.1
47.9	131.2	2.0	27.0	21.9	44.1	5.0
60.4	133.7	1.7	27.7	15.2	52.4	3.1
64.3	134.6	2.3	26.3	21.2	46.5	3.7
70.4	136.0	2.1	27.7	18.6	47.0	4.7
82.9	139.6	2.3	25.1	22.0	45.3	5.3
83.2	139.7	2.2	26.3	19.4	47.1	5.0
85.5	140.5	2.1	24.6	22.9	45.0	5.4

KPF	Stratigraphic		
Sample No.	Position	δ13 C	SD
-270	7.6	-28.388	0.077
-250	15.3	-30.072	0.013
-230	23.1	-29.318	0.397
-220	27.0	-29.963	0.085
-210	30.9	-29.928	0.019
-190	38.6	-28.866	0.174
-180	42.4	-30.077	0.168
-160	49.7	-29.813	0.000
-110	68.1	-29.639	0.128
-49.8	95.1	-29.447	0.154
-46.5	96.9	-29.180	0.518
-41.9	99.3	-29.243	0.730
-39.1	100.8	-30.283	1.009
-35	103.0	-29.542	0.213
-31	105.1	-30.153	0.180
-14	114.2	-30.341	0.151
-12	115.2	-30.090	0.080
-9.5	116.5	-29.982	0.074
-7.6	117.4	-29.707	0.116
-5.3	118.5	-29.744	0.064
-4.3	119.0	-29.949	0.055
-1	120.5	-30.365	0.025
2.8	121.9	-30.077	0.011
5	122.5	-29.372	0.072
7.4	123.2	-28.919	0.201
10.1	123.9	-30.014	0.007
13.2	124.7	-29.548	0.103
15	125.1	-29.352	0.278
17.3	125.6	-29.480	0.368
21	126.4	-28.401	1.522
22	126.6	-28.566	0.058
23.6	126.9	-29.973	0.117
24.5	127.1	-29.620	0.007
26.8	127.5	-28.498	0.116

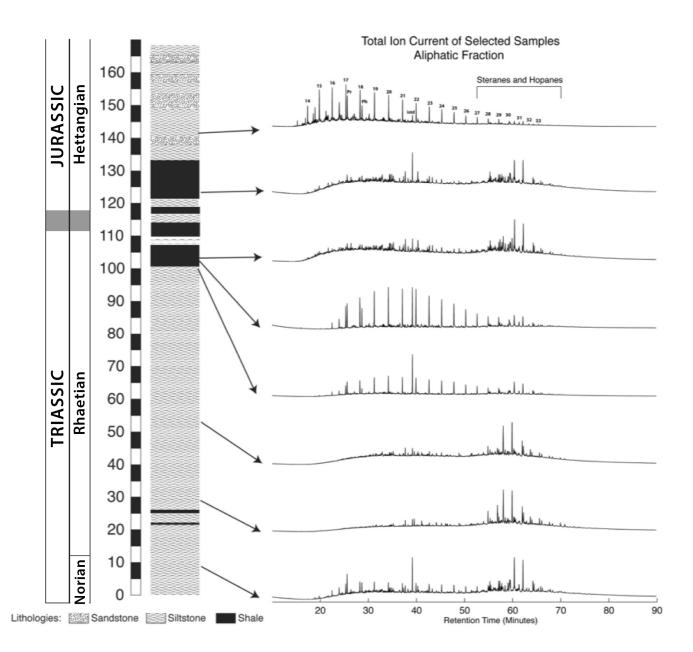
30.5	128.2	-29.851	0.072
32.2	128.5	-30.034	0.217
34.6	128.9	-29.965	0.121
42.7	130.3	-29.350	0.093
44.1	130.6	-30.034	0.145
47.9	131.2	-27.608	0.497
60.4	133.7	-28.202	0.464
64.3	134.6	-26.112	0.061
70.4	136.0	-30.013	0.010
82.9	139.6	-30.023	0.017
83.2	139.7	-29.344	0.116
85.5	140.5	-28.244	0.163

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Sample No.	C/N	SD
KPF -270	38.40	0.50
KPF -250	11.68	0.46
KPF -240	7.25	0.22
KPF -230	8.97	0.24
KPF -220	12.71	0.61
KPF -210	25.42	0.41
KPF -200	28.01	0.58
KPF -190	9.49	0.33
KPF -180	22.91	1.07
KPF -160	25.97	0.65
KPF -150	19.37	0.13
KPF -120	29.52	0.23
KPF -110	28.93	0.25
KPF -49.8	29.62	1.64
KPF -46.5	14.32	0.61
KPF -41.9	1.78	0.16
KPF -39.1	2.84	0.03
KPF -35	14.01	0.27
KPF -31	4.80	0.08
KPF -29.1	3.20	0.06
KPF -14	25.45	1.30
KPF -12	26.99	1.18
KPF -9.5	46.13	0.71
KPF -7.6	0.12	0.01
KPF -5.3	22.05	0.28
KPF -4.3	44.83	0.36
KPF -1	31.19	0.57
KPF 2.8	21.45	0.89
KPF 5.0	28.62	0.53
KPF 7.4	38.30	12.98
KPF 10.1	21.63	0.09
KPF 13.2	18.48	0.91

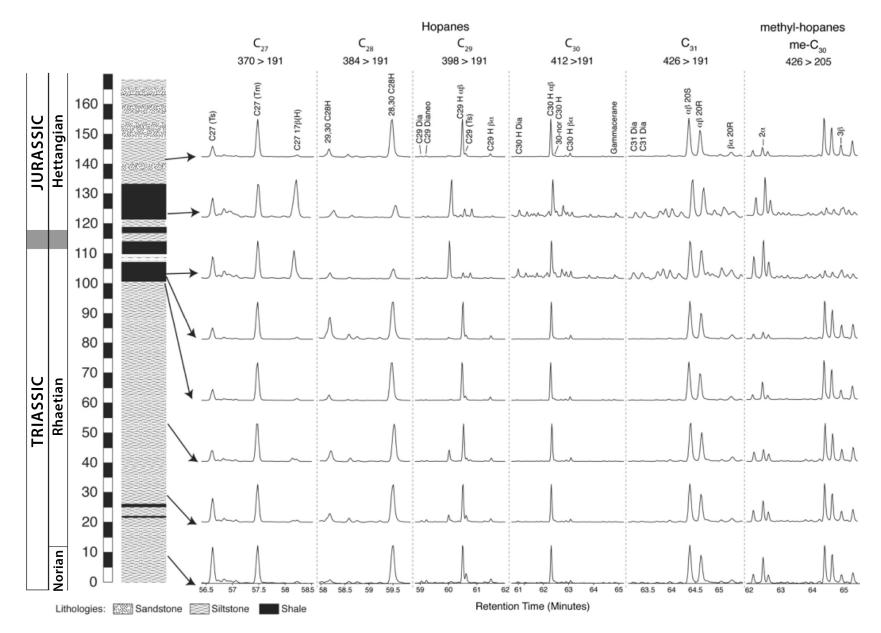
23.71	0.73
32.09	1.46
18.87	0.73
16.83	0.34
46.93	1.13
34.59	1.31
35.83	0.78
25.43	0.53
20.57	0.49
26.10	1.02
13.64	0.27
17.54	0.68
4.45	0.05
9.24	0.55
18.97	0.23
14.76	0.47
42.90	1.18
14.50	1.18
27.88	0.44
	32.09 18.87 16.83 46.93 34.59 35.83 25.43 20.57 26.10 13.64 17.54 4.45 9.24 18.97 14.76 42.90 14.50



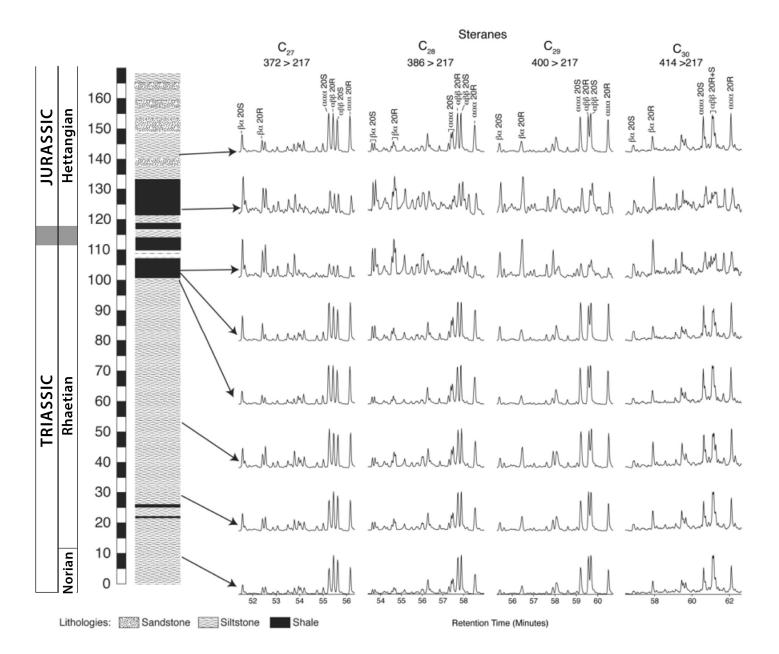
Data Repository Figure 1. (A-E) Elemental data and biomarker concentrations; (F-J) thermal maturity and lithology parameters; (K-O) C_{26} - C_{30} sterane ratios. Thermal maturity is further discussed in Section S2. The complete sterane data show that C_{26} and C_{30} steranes comprise a minor fraction of the sterane record.



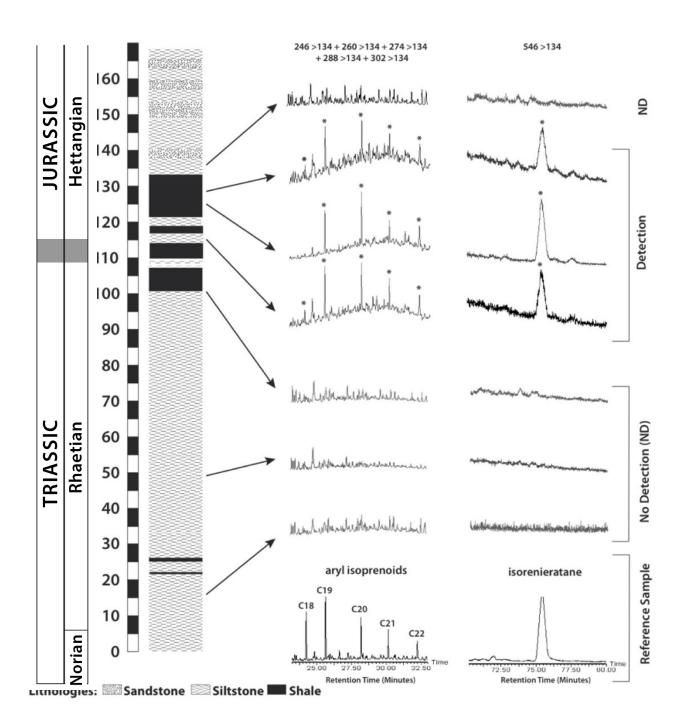
Data Repository Figure 2. Total Ion Chromatogram (TIC) of the aliphatic hydrocarbon fraction of representative samples. Some samples exhibit considerable evidence of biodegradation, as indicated by low alkane:(hopanes+sterane) ratios, and an elevated baseline. To overcome biodegradation we focused on the analysis of hopanes, steranes, and aryl isoprenoids by MRM-GC-MS. Biomarker ratios were used to account for the level of biodegradation present in each sample. Numbers indicate carbon number of *n*-alkanes, "Pr" pristane, "Ph" phytane,"istd" internal standard.



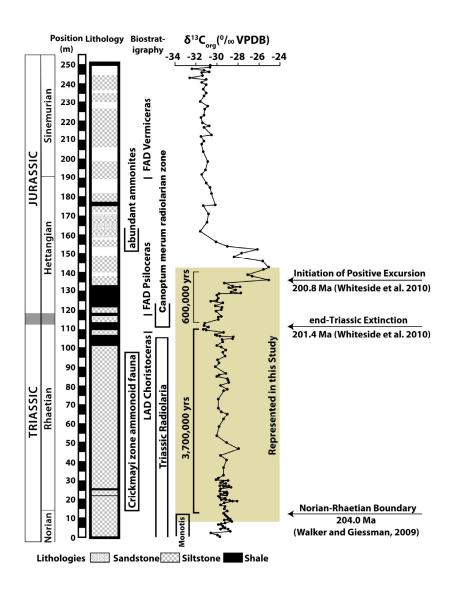
Data Repository Figure 3. GC-MRM-MS transitions of desmethyl and methyl pentacyclic triterpanes, plus gammacerane.



Data Repository Figure 4. MRM-GC-MS traces of C₂₇-C₃₀ steranes.



Data Repository Figure 5. GC-MRM-MS transitions of aryl isoprenoids (left) and isorenieratane (right) in selected representative samples. Data were compared to samples from two time intervals of widespread anoxia, the Cenomanian-Turonian Boundary in Central Jordan (bottom chromatogram; Sepulveda et al., 2009) and the end-Permian extinction from Meishan, China (data not shown; Cao et al., 2009). Numbers indicate carbon atom numbers for aryl isoprenoids, and * denotes unambiguously detected peaks used for integration. Numbers on top of chromatograms indicate MRM transitions used for compound identification.



Data Repository Figure 6. Time span represented by this study (colored area) in the context of the complete carbon isotope profile from Kennecott Point. Modified from Williford et al. (2009).