

Table FT1: Analytical data: Ag, Cu, and Pb isotopic compositions, Pb model ages T in million years (Ma), $^{238}\text{U}/^{204}\text{Pb}$ (μ) and $^{232}\text{Th}/^{238}\text{U}$ (κ), and description of the samples analyzed in this work.

Name	Dates	Monarch	$\epsilon^{109}\text{Ag}$ 10^{-4}	2s.e.m.	$\delta^{65}\text{Cu}$ ‰	2s.d.	$^{206}\text{Pb}/^{204}\text{Pb}$	2s.d.	$^{207}\text{Pb}/^{204}\text{Pb}$	2s.d.	$^{208}\text{Pb}/^{204}\text{Pb}$	2s.d.	Pb model age T Ma	$^{238}\text{U}/^{204}\text{Pb}$ (μ)	$^{232}\text{Th}/^{238}\text{U}$ (κ)
EdIa	1248-1250	Edward I	-0.14	0.08	-0.09	0.04	18.455	0.003	15.642	0.003	38.479	0.007	225	9.71	3.92
EdIIa	1307-1327	Edward II	-0.29	0.08	-0.05	0.04	18.454	0.003	15.638	0.003	38.448	0.007	221	9.70	3.91
EdIIb	1307-1327	Edward II	-0.22	0.08	-0.06	0.04	18.437	0.003	15.638	0.003	38.452	0.007	233	9.70	3.92
EdVIa	1551	Edward VI	0.26	0.04	0.11	0.04	18.415	0.003	15.646	0.003	38.467	0.008	257	9.72	3.94
MaIa	1553-1554	Mary I	0.28	0.05	0.03	0.04	18.659	0.003	15.661	0.003	38.719	0.007	101	9.72	3.92
ElIa	1558-1603	Elizabeth I	0.32	0.09	0.06	0.03	18.701	0.002	15.649	0.002	38.663	0.005	58	9.70	3.87
ElIb	1558-1603	Elizabeth I	0.01	0.07	0.05	0.04	18.695	0.003	15.650	0.003	38.665	0.008	64	9.70	3.87
ElIc	1565	Elizabeth I	0.27	0.04	0.04	0.04	18.604	0.003	15.641	0.003	38.575	0.007	118	9.69	3.88
ElId	1567	Elizabeth I	0.20	0.10	0.10	0.04	18.679	0.003	15.654	0.003	38.659	0.007	79	9.71	3.88
JaIa	1603-1625	James I	0.11	0.07	0.13	0.04	18.616	0.003	15.645	0.003	38.610	0.007	114	9.70	3.89
JaIb	1602-1607	James I	-0.16	0.07	0.17	0.04	18.636	0.003	15.641	0.003	38.624	0.007	95	9.69	3.88
ChIa	1641-1643	Charles I	0.16	0.09	-0.19	0.04	18.696	0.003	15.678	0.003	38.919	0.008	96	9.75	3.98
ChIb	1640	Charles I	0.09	0.07	0.06	0.04	18.659	0.003	15.654	0.003	38.707	0.007	95	9.71	3.91
ChIc	1625-1649	Charles I	0.71	0.08	0.12	0.04	18.414	0.003	15.632	0.003	38.502	0.007	243	9.70	3.95
ChId	1625-1649	Charles I	0.19	0.07	0.11	0.04	18.641	0.003	15.654	0.003	38.696	0.007	107	9.71	3.92

$$\epsilon^{109}\text{Ag} = [({}^{109}\text{Ag}/{}^{107}\text{Ag})_{\text{sample}} / ({}^{109}\text{Ag}/{}^{107}\text{Ag})_{\text{standard}} - 1] \times 10,000$$

$$\delta^{65}\text{Cu} = [({}^{65}\text{Cu}/{}^{63}\text{Cu})_{\text{sample}} / ({}^{65}\text{Cu}/{}^{63}\text{Cu})_{\text{standard}} - 1] \times 1000$$

s.e.m.: standard error of the mean

s.d.: standard deviation

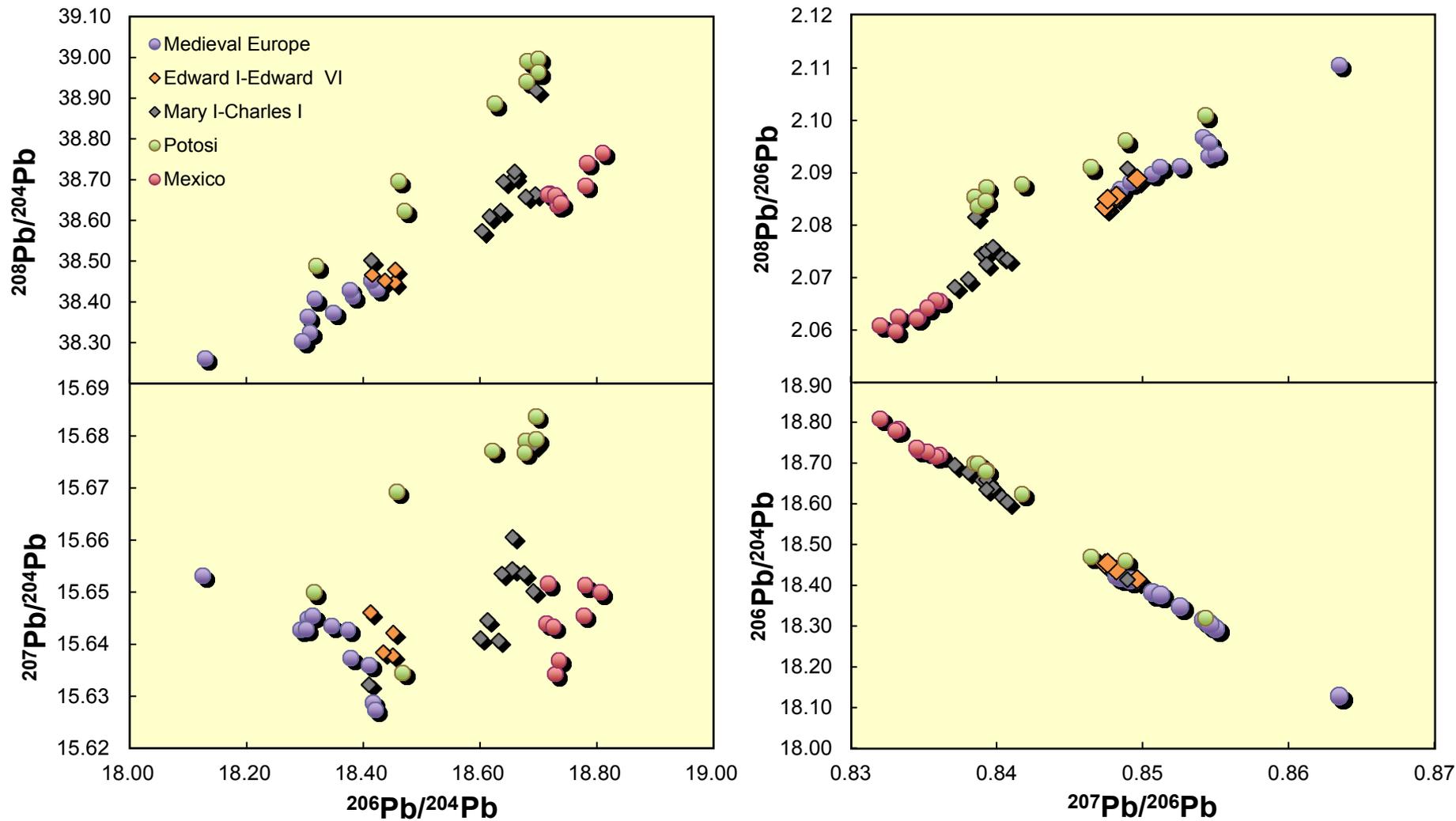


Figure FT1. Conventional plots of the Pb isotope compositions of English coins analyzed in this work (Table A1) and in coins from Hercynian Europe, Potosi, and Mexico by Desaulty et al. (2011)

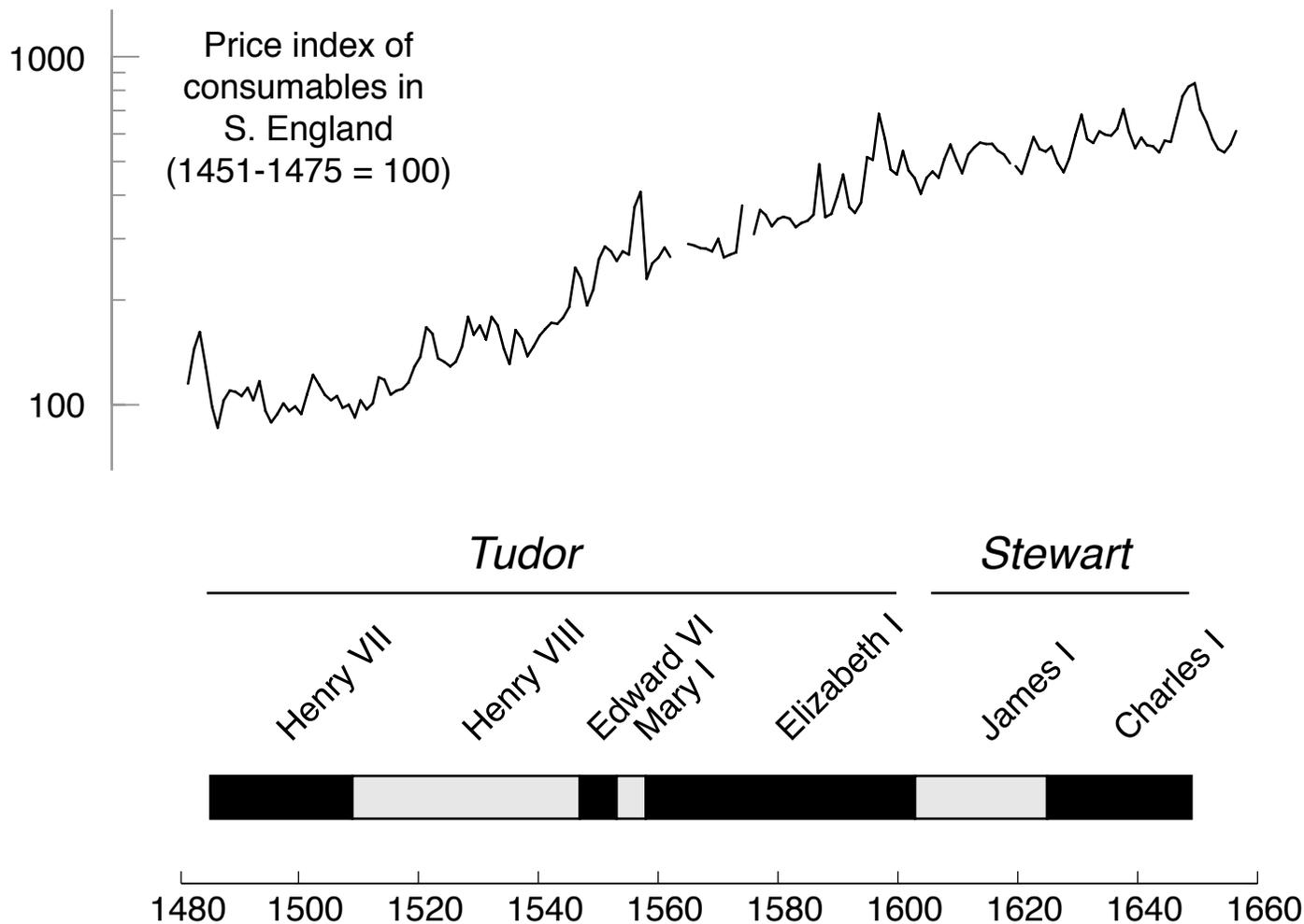


Figure S2. Prices of commodities (log scale) in Southern England between 1480 and 1660 as compiled by Phelps Brown and Hopkins (1956). The value of 100 represents the average over the 1451-1475 period.