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Late Archean euxinic conditions before the rise of atmospheric oxygen

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Geological and depositional setting of studied samples

The Marra Mamba Iron Formation is the stratigraphically lowest iron formation of the Hamersley Province in Western Australia. It is interpreted to have been deposited in a deep-water setting on a passive margin (Krap ež et al., 2003). The underly ing succession comprises, in descen ding or der, the Roy Hill Shale, the Warrie Member, the Woodiana Sandstone (collectively making up the Jeerinah Formation), and the Maddina Basalt. The Maddina Basalt, Woodian a Sandstone and the Warrie Mem ber together record the last stage of a protracted rifting event that opened an ocean to the south (Blake and Barley, 1992). The Roy Hill Shale records a rift-drift transition and its basal contact is a breakup surface. At most localities the contact between the Warrie Member and the Roy Hill Shale is conformable, but in some cases the contact is a subaerial erosion surface. The Roy Hill Shale is conformable with the overly ing Nammuldi Member, which is the low ermost division of the Marra Ma mba Iron Formation. The age of the Jeerinah Formation is defined by a $2,662 \pm 7$ Ma U-Pb zircon age from a thin, <1 mm-thick, ash-fall tuff at the top of the Roy Hill Shale Member, the upper Jeerinah Formation (Rasmussen and Fletcher, 2010).

FVG-1 was drilled between 1983 and 1984 by CRA Exploration Pty Ltd. in the Fortescue Valley between the North Pilbara Granite-Gre enstone Terrain to the nort h, and the H amersley Province to t he south at 22° 33'S, 119° 30'E. The drill-hole section, comprising the Namm uldi Member, Roy Hill Shale, Warrie Member, Woodiana Sandstone and Maddina Basalt, is typical of those successions, but m ost of the Marra Mamba Iron Formation is faulted out. The distinction in drill-core between the Warrie Member, Roy Hill Shale and Namm uldi Memb er is easily made. The Warrie Member comprises thin bedded carbonate turbidites and or ganic matter-rich (black) sh ales, but the shales lack abundant early diagenetic pyrite. The R oy Hill S hale consists of organic matter-rich (black) shale with abundant early diagenetic pyrite. The Nummuldi Member consists of stilpnomelane-rich shales and thin-bedded carbonate turbidites but contains little early diagenetic pyrite. The drill-hole also contains a meteorite-impact spherule bed (at drill-depth 750.68 m) that has now be en recognized in m any drill-hole sections throughout the Pilbar a region of Western Australia and also in South Africa (Simonson et al., 2000).

The Hamersley Province, the site of de position for the Jeerinah Formation, is mildly deformed and metamorphosed, with depth of buri al being the primary control on meta morphic grade (S mith et al., 1982). The FVG-1 drill-hole was drilled in the area that experienced prehnite-pumpellyite-epidote facies of metamorphism.

Archaean Sulfur Budget

Presence of large MIF in S isotopes s uggests that photochem ically-processed and volcanically-derived S O_2 was the m ain sulf ur source f or pyrite in shales of the Jee rinah Formation. We envision, based on the range of MIF in pyrites, that volcanic em issions with SO_2 :H₂S ratios ranging between 2 a nd 10 delivered sulfur to the ca. 2.66 Ga anoxic atm osphere (cf. Halevy et al., 2010). SO $_2$ was photolyzed in the oxygen- and ozone-free atm osphere and sulfate and elem ental sulfur aeros ols carried sulfur with MIF to the ocean (cf. Farquhar et al., 2001; Pavlov and Kasting, 2002; Za hnle et al., 2006). Hydrogen sulfid e and sulfur dioxide that escaped photolysis in the atm osphere did not carry MIF and ther efore diluted MIF in sulfur compounds delivered from the atmosphere. W e develop below a sim ple mass-balance model to account for sulfur concentrations in shales of the Jeerinah Form ation using volcanic sulfur as the single sulfur flux to the Archean an oxic atm osphere-ocean and reason able sed imentation rates for deposition of organic matter-rich shales.

Estimates of the m odern volcanic SO ₂ flux ra nge widely, from values of 1.5 x 10 ¹² g SO₂/yr¹⁴ to those approaching 50 x 10 ¹² g SO ₂/yr (Lambert et al., 1988). However, recent extensive c ompilations that in tegrate d irect f lux m easurements on re latively lon g tim escales yield estimates of ~15-20 x 10 ¹² g SO ₂/yr (Berresheim and Jaeschke, 1983; Bluth et al., 1993; Andres and Kasgnoc, 1998; Halm er et al., 2002). Although the bulk of the m odern volcanic sulfur flux is typically considered to be in the f orm of SO₂, the flux of H ₂S from volcanoes is poorly constrained and has been es timated to be as high as 37 x 10 ¹² gH₂S/yr (Halmer et al, 2002).

Assuming an SO $_2$ outgassing rate of 15 x 10 12 g SO $_2$ /yr and a conservative H $_2$ S outgassing rate of 3 x 10 12 g H₂S/yr yields a total sulfur outgassing rate of 1.8 x 10 13 g S/yr. We adopt this as our m inimum modern flux rate and assum e a value of 3.2 x 10 13 g S/yr as our maximum modern flux (Holland, 2002). Assuming that roughly half of this is mantle sulfur (e.g., Habicht et al., 2002), we obtain a mode rn mantle flux of between 9.0 x 10 12 and 1.60 x 10 13 g S/yr. Rather than m aking assumptions regarding th e areal extents of continents versus oceans during the Archean, we integrate this flux over the entire surface of the Earth to y ield a globally averaged m ass flux of 1.77 – 3.14 x 10 $^{-2}$ g/m²yr for volcanic sulfur of m antle origin g iven modern outgassing rates.

For a sedim ent accumulating at ~1 m /Myr (a reasonable po st-compaction accumulation rate for a distal black shal e (Leventhal, 1991; Schieber and Baird, 2001; Schieber, 2009), assuming a typical shale density of 2.7 g/cm³, this globally averaged mass flux would correspond to the accumulation of 0.65 - 1.16 wt% pyrite sulfur, slightly lower than the average pyrite sulfur content of the euxinic Roy Hill Mem ber (2.41±1.09 wt%). If we assume that the bulk of this

mantle sulfur flux was sequestered as pyrite in depositional environm ents similar to thos e represented by the Roy Hill M ember (i.e., $\sim 2.41 \text{ wt\%}$ pyrite sulfur accumulating at $\sim 1 \text{ m/Myr}$), then the ar eal extent of this depositional mode could have been $\sim 22 - 68\%$ of t he modern seafloor area.

The upper end of this range is probably extr eme, as it is not likely that the entire inventory of volcanic sulfur w ould be sequestered in a single depositional facies. Additionally, bulk sedim ent accum ulation rates m ay have been som ewhat faster than our assum ed postcompaction rate of 1 m/Myr, which would result in lower calculated pyrite sulfur concentrations and a more restricted areal extent of sulfidic deposition. For example, doubling the accumulation rate to 2 m/Myr would yield a range of 0.19 - 0.58 wt% pyrite sulfur and an areal extent of 11 - 0.58 wt% pyrite sulfur area extent of 10 - 0.58 wt% pyrite sulfur area extent of 10 - 0.58 wt% pyrite sulfur area extent of 10 - 0.58 wt% pyrite sulfur area extent of 10 - 0.58 wt% pyrite sulfur area extent of 10 - 0.58 wt% pyrite sulfur area extent of 10 - 0.58 wt% pyrite sulfur area extent of 10 - 0.58 wt% pyrite sulfur area extent of 10 - 0.58 wt% pyrite sulfur area extent of 10 - 0.58 wt% pyrite sulfur area extent of 10 - 0.58 wt% p 34% of the modern seafloor, using the sam e mantle flux. On the other hand, Archean outgassing rates were probably elevated relative to the m odern (Kasting, et al., 1989; Zahnle et al., 2006), which would presum ably have resulted in a commensurate increase in volcanic SO₂ and H₂S fluxes, and, regardless of the m ode of subduction assum ed for this period of Earth's history, it seems likely that som e sulfur of sedim entary origin would have contributed to the overall volcanic flux. These calculations also neglect contributions from other gases such as CS₂ and OCS, whose combined modern flux m ay be as high as \sim 2.5 x 10⁻¹¹ g S/yr (Halm er et al., 2002), and direct em issions of SO $_4^{2-}$ in volcanic plum es, which m ay be significant (Berresheim and Jaeschke, 1983). These vagaries aside, it seem s clear that even a m odest volcanic sulfur source could have provided f or the su lfur burial f lux recorded by the Roy Hill Shale, resulting in potentially widespread euxinic de position given variations in the availability of organic carbon, Fe_{HR}, and volcanic activity.

Iron Speciation

Although ferruginous deposition for the lower R oy Hill M ember is clearly indicated by the combination of elev ated Fe $_T$ /Al ratios and low values f or Fe $_{PY}$ /Fe $_{HR}$, these samples also display the unusual combination of high Fe $_{T}$ /Al ratio s and relatively low Fe $_{HR}$ /Fe_T values. Although this relationship could result from auth igenic and/or m etamorphic form ation of Fecontaining silicate m ineralogies, it is also possible that the extraction of Fe_{HR} is incomplete for samples with high concentrations of crystalline or coated siderite, a phenom enon suggested by Mt. McRae Shale. To test this Reinhard et al. (2009) for the possibility, we perform ed a n additional, aggressive boiling HCl ex traction (Fe_{HCl}), which quantitatively m obilizes all side rite and all other Fe _{HR} phases, as well as any authigenic Fe silicate m inerals and som e amount of poorly reactive silicate phases. Samples from the Roy Hill Me mber show a reasonably good correspondence between se quentially extracted Fe HR co ntent and Fe HCI content, with the exception of som e samples from the lower ferrugi nous interval (Fig. DR1). W e interpret this inconsistency to reflect an underestim ation during the sequential procedure of Fe _{HR} present as

refractory siderite and/or secondary sili cates. We emphasize that this additional Fe $_{HR}$ would increase Fe $_{HR}$ /Fe $_{T}$ ratios while decreasing Fe $_{PY}$ /Fe $_{HR}$ values, thereby strengthening our argum ent for an Fe(II)-containing water column prior to the onset of euxinia.



Figure DR1. Cross-plot of HCl soluble Fe and sequentially extracted Fe for the Jeerinah Formation. Blue diamonds: euxinic interval; black diamonds: upper ferruginous interval; green diamonds: lower ferruginous interval. Open field encompasses samples where Fe_{HR} is underestimated by the sequential extraction

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Table DR1.

FVG-1	тос	Pyrite S	δ ³⁴ S	δ ³³ S	∆ ³³ S	AI	Мо	Total Fe	Acetate Fe	Dithionite Fe	Oxalate Fe	Pyrite Fe	Fe _{нк} /Fe _т	Fe _{PY} /Fe _{HR}	Fe _⊺ /Al
(m)	(wt. %)	(wt. %)	(‰)	(‰)	(‰)	(wt. %)	(wt. %)	(wt. %)	(wt. %)	(wt. %)	(wt. %)	(wt. %)			
748.3	1.8	0.9	1.2	1.37	0.27	8.4	2	6.6	0.3	0.3	0.7	0.7	0.3	0.4	0.8
749.65	3.3	2.7	2.2	1.96	0.42	8.6	2	4.7	0.3	0.2	1	2.3	0.8	0.6	0.5
750.56	6.1	1.2	0.9	0.99	-0.02	9	3	2.4	0.1	0.1	0.3	1	0.6	0.6	0.3
751.15	4.5	0.5	2.5	-	-	8.8	0	1.5	0.1	0.2	0.2	0.4	0.5	0.5	0.2
752.65	7.5	1.4	12.7	13.37	5.94	8.8	3	1.4	0.1	0.2	0.1	1.2	1.1	0.8	0.2
753.95	4.8	0.6	7.7	-	-	6	3	1.7	0.2	0.1	0.2	0.5	0.6	0.5	0.3
755.2	2.4	0.1	3.8	1.72	-0.47	2.9	1	2.2	0.4	0.1	0.6	0.1	0.6	0.1	0.8
756.6	3.8	0.3	8.6	2.04	0.16	5.4	1	1.4	0.4	0.1	0.1	0.3	0.6	0.3	0.3
756.8	5.6	0.7	4	1.66	-1.35	5.5	-	1.4	0	0.1	0	0.6	0.5	0.9	0.3
760.7	7.6	2	3.3	1.72	-0.47	7.3	2	2.4	0	0.1	0	1.7	0.8	0.9	0.3
761.8	5.9	3.6	2.6	2.86	1.08	6.2	3	3.9	0	0	0	3	0.8	1	0.6
764.45	4.6	1.7	2.3	0.77	-0.82	6.1	-	3.7	0.1	0	0	1.4	0.4	0.9	0.6
765	10.7	1.9	6	9.52	6.76	5.5	3	2.7	0.4	0.2	0.1	1.6	0.8	0.7	0.5
767.6	4.6	1.6	5	5.91	2.58	4.5	2	2.1	0.1	0	0.1	1.3	0.7	0.8	0.5
774	4.4	3.4	1.7	0.74	-0.73	5.7	3	4.5	0.2	0.1	0.1	2.9	0.7	0.9	0.8
775.55	4.7	2	3.3	1.6	-0.43	6	2	2.5	0.1	0.1	0.1	1.7	0.8	0.8	0.4
776.4	5.8	3	5.6	5.72	2.5	6.2	4	3	0.2	0.1	0.1	2.5	1	0.9	0.5
777.8	11.3	3.5	-0.7	1.07	1.04	5.8	3	3	0	0.1	0	3	1	1	0.5
779.45	10.1	3.3	0.2	2.24	1.87	5.8	2	3	0	0.1	0	2.7	1	0.9	0.5
780.3	1.2	0.5	-0.1	2.06	1.9	9	1	4	2	0.1	0.7	0.4	0.8	0.1	0.4
780.95	10	2.3	8.9	13.33	7.94	7.2	3	2.5	0.2	0.2	0.1	1.9	0.9	0.8	0.3
782.9	7.8	3.8	10.2	13.1	7.17	6.1	-	3.1	0.2	0.1	0.1	3.2	1.1	0.9	0.5
787.4	7.6	3.8	6.5	9.33	5.69	6.4	6	4.5	0.7	0.1	0.1	3.2	0.9	0.8	0.7
791	4.7	1.1	-0.4	-	-	7.7	4	3.6	0.6	0.1	0.3	0.9	0.5	0.5	0.5
794.1	4.7	2.9	1.7	6.4	4.82	6.7	4	5	0.2	0.1	0.3	2.4	0.6	0.8	0.8

796.2	4.4	0.8	1.1	-	-	7.1	4	4.5	1	0.1	0.8	0.7	0.5	0.3	0.6
797.05	4	1.1	7.5	11.3	6.98	7.9	6	4.5	0.3	0.1	0.7	0.9	0.4	0.5	0.6
802.1	1.7	0.5	3.5	-	-	5.6	3	5.1	0.4	0.2	0.6	0.4	0.3	0.3	0.9
813.25	1.5	0.2	1.5	-	-	6.3	-	4.1	2.5	0.2	0.4	0.2	0.8	0.1	0.7
813.6	2.5	0.8	3.7	-	-	6.4	3	4.3	0.2	0.1	0.9	0.7	0.5	0.3	0.7
815.2	2.7	0.5	2.4	5.06	3.44	7.7	4	4.9	0.5	0.1	0.9	0.4	0.4	0.2	0.6
818.8	2.2	0.2	2.5	-	-	7.4	4	5.2	0.6	0.1	0.6	0.2	0.3	0.1	0.7
822.45	2.3	0.8	0.2	1.56	1.19	5.8	3	4.7	0.7	0.1	1.1	0.6	0.5	0.2	0.8
823.3	3.4	0.5	2.8	-	-	6.6	3	4.5	0.3	0.1	0.2	0.4	0.2	0.4	0.7
829.9	2.3	0.6	2.6	4.33	2.65	6.8	5	5.1	1.2	0.2	0.4	0.5	0.5	0.2	0.8
849.6	1.4	0.5	0	-	-	6.7	-	4.7	0.2	0.1	0.1	0.4	0.2	0.5	0.7
855.65	0.7	0.8	-0.3	-	-	7.8	3	7.5	1.1	0.1	0.4	0.7	0.3	0.3	1
861.05	0.9	0.4	-0.7	-	-	8.4	2	5.8	0.4	0.1	1.1	0.4	0.3	0.2	0.7
881.7	1.5	0.4	1.3	-	-	8.3	2	4.8	0.2	0.1	1.1	0.3	0.4	0.2	0.6