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- 3 The Discrepancy between Mineral Residence Time and Soil Age: Implications for
- 4 the Interpretation of Chemical Weathering Rates

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11 Derivation of Eq. 3

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- The mass loss rate of mineral species j per unit ground surface area via chemical 13
- weathering ( $w_i$ , with the unit of ML<sup>-2</sup>T<sup>-1</sup>) is described as: 14

15

16 
$$w_j = -\frac{\partial m_j}{\partial \tau_j} = R_j A_{s,j} \omega_j$$
. (A)

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- where  $m_i$  is the mass of mineral j per a given ground surface area [ML<sup>-2</sup>],  $\tau$  is the time 18
- that the minerals are exposed to chemical weathering reactions, R is the chemical 19
- weathering rate in moles reacted per mineral surface area per time [mol L<sup>-2</sup> T<sup>-1</sup>],  $A_s$  is the 20
- mineral surface area per ground surface area [unitless], and  $w_m$  is molar weight of the 21
- mineral j [M mole<sup>-1</sup>], 22

The surface area of the minerals per ground surface area  $[A_s, unitless]$  is related to the 1

2 mineral grain roughness by [White and Brantley, 2003]:

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$$A_{s,j} = \frac{6\lambda_j}{D_j \rho_{\mu,j}} m_j, \tag{B}$$

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where D is the grain diameter [L],  $\lambda$  is the mineral surface roughness [unitless], and  $\rho_m$  is 6

the density of the minerals [ML<sup>-3</sup>]. 7

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9 Both the mass and the surface area of the minerals may evolve due to chemical

10 weathering. In the model presented here we adopt the White and Brantley [2003]

11 weathering model, which describes the chemical weathering rate per mineral surface area

and the mineral grain roughness as a function of the mineral's exposure time to chemical

13 weathering as

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15 
$$R_j = a_j \tau_j^{\alpha_j}$$
, (Ca)

16

17 and

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19 
$$\lambda_j = b_j \tau_j^{\beta_j}$$
, (Cb)

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21 where a, b,  $\alpha$ , and  $\beta$  are mineral species specific empirical coefficients.

1 Thus the mass loss rate of mineral species *i* per ground surface area via chemical

2 weathering ( $w_i$  in Eq. A) can be rewritten as in the Eq.3 in the text:

3

$$4 w_{j} = -\frac{\partial m_{j}}{\partial \tau} = \underbrace{\frac{\partial a_{j}b_{j}\omega_{j}}{D_{j}\rho_{\mu,j}}\tau_{j}^{\alpha_{j}+\beta_{j}}}_{time-dependent rate coefficient[yr^{-1}]} \times \underbrace{m_{j}}_{current mass}. (D)$$

5

6 This equation describes a first order decay with time-dependent decay rate; the rate Wi is

7 the product of the time-dependent rate coefficient and the current mass of i. If the initial

mass per area of the mineral j is  $m_{j,0}$ , the  $m_j$  in Eq. D can be solved as a function of 8

9 weathering exposure time ( $\tau$ ) and the initial mass ( $m_{i,0}$ ):

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11 
$$m_j = m_{j,0} \exp\left[-\frac{6a_j b_j \omega_j}{D_i \rho_{\mu,j} (1 + \alpha_j + \beta_j)} \tau_j^{1 + \alpha_j + \beta_j}\right].$$
 (E)

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# **Derivation of Eq. 4:**

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15 Over the course of soil formation, primary mineral grains of different species are

continuously incorporated into the soil. The amount of mineral, i, per unit ground surface

area that enters the soil from the underlying fresh parent material via soil production

between the time period of t and t+dt [ $m_{i,t}$  with the unit of ML<sup>-2</sup>], where the t is the soil

19 age, is:

$$1 m_{j,t} = (\rho_r C_{j,r} P_t) dt, (F)$$

3 where  $\rho_r$  is the bulk density of the parent material [ML<sup>-3</sup>],  $C_{jr}$  is the mass concentration of

4 the primary mineral j in the parent material [MM<sup>-1</sup>], and P<sub>t</sub> is the soil production rate at

5 time t [LT<sup>-1</sup>].

6

Now we consider a soil of age T. As described in the time-dependent dissolution

8 rate law described in Eq. D, the mass loss rate via chemical weathering from the mass

9 fraction,  $m_{i,t}$  in Eq. F, is the product of (1) the time-dependent reaction coefficient for the

weathering exposure time of T-t and (2) the present mass of j that entered the soil

between t and t+ $\Delta$ t and has survived the chemical weathering loss until time T (i.e.,

12  $m_{it\rightarrow T}$ ).

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14 
$$w_{j,t,T} = \frac{6a_j b_j \omega_j}{D_j \rho_{\mu,j}} (T - t)^{\alpha_j + \beta_j} \times m_{j,t \to T}.$$
 (G)

15

As was done for Eq. E,  $m_{j,t\to T}$  can be solved as a function of initial mass  $(m_{j,t}$  in Eq. F)

17 and the residence time, *T-t*.

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19 
$$m_{j,t\to T} = (\rho_r C_{j,r} P_t) dt \exp\left[-\frac{6a_j b_j w_j}{D_j \rho_{\mu,j} (1 + \alpha_j + \beta_j)} (T - t)^{1 + \alpha_j + \beta_j}\right]$$
 (H)

1 The soil of age T is comprised of primary minerals that have been incorporated into the

soil during the entire duration of soil formation. The mass loss from the mineral species i 2

in the entire soil profile ( $w_{i,T}$  with the unit of ML<sup>-2</sup>T<sup>-1</sup>) can be thus obtained by integrating 3

 $\mathbf{w}_{i,t,T}$  (Eq. G) over the time period from t=0 to t=T. 4

5

6 
$$w_{j,T} = \int_{t=0}^{t=T} w_{j,t,T} dt$$
, (Ia)

7

$$8 = \frac{6a_{j}b_{j}\omega_{j}}{D_{j}\rho_{\mu,j}} \int_{t=0}^{t=T} \left[ (T-t)^{\alpha_{j}+\beta_{j}} (\rho_{r}C_{j,r}P_{t}) \exp\left[-\frac{6a_{j}b_{j}\omega_{j}}{D_{j}\rho_{\mu,j}(1+\alpha_{j}+\beta_{j})} (T-t)^{1+\alpha_{j}+\beta_{j}}\right] \right] dt, \quad (Ib)$$

9

$$10 = \frac{6a_j b_j \omega_j}{D_j \rho_{\mu,j}} \int_{t=0}^{t=T} \left[ (T-t)^{\alpha_j + \beta_j} m_{j,t \to T} \right] dt, \qquad (Ic)$$

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Eq. (Ic) follows from Eq. (Ib) by inserting Eq. (H) and is featured as Eq.4 in the text. 12

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# Derivation of Eq. 5

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16 If we incorrectly equate the mineral residence time ( $\tau = T - t$ ) and soil age (T), as has

been done in the past, the chemical weathering rate of mineral species i from a soil of age

18 *T* is described in Eq.5 in the text:

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$$1 w_{j,T} = \underbrace{\frac{6a_j b_j \omega_j}{D_j \rho_{\mu,j}} T^{\alpha_j + \beta_j}}_{time-dependent} \times \underbrace{(\rho_s C_{j,s} h_s)}_{current \ mass}. (J)$$

3 where the current mass of mineral i is calculated as the product of soil bulk density, mass

4 concentration of *j*, and soil thickness at time *T*.

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## **Simulation Procedure**

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- In the simulations, mineral grains are introduced into the soil during each timestep. The 8
- 9 mass per unit ground surface area of the mineral grains introduced during a timestep of
- 10 time *dt* is

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12 
$$m_{j,0} = P\chi_j \rho_{\mu,j} (1-\phi) dt$$
, (K)

- where P is the soil production rate, determined by the soil thickness (see text),  $\chi_j$  is the 14
- volume fraction of the mineral in the parent material (mass fraction and volume fraction 15
- 16 are assumed to be the same in the simulations presented here because quartz, plagioclase,
- and K-feldspar have similar mineral densities),  $\rho_{\mu}$  is the mineral density of species j, and 17
- $\phi$  is the porosity, assumed to be 0.4 for all simulations.  $\rho_{\mu}\chi_{i}(1-\phi)$  is identical to  $\rho_{r}C_{ir}$  in 18
- 19 Eq. (F). It should be noted that the simulated particles represent a cohort of individual
- 20 mineral grains that are produced in the time period  $\Delta t$ . Each mineral grain is then tagged
- 21 with an age; the clock starts once the mineral grain is introduced into the soil layer.

2 As time passes, the soil-profile-integrated mass and weathering rate of the mineral j is

3 updated according to Eq. (I). Thus the mass in the total soil profile and also of individual

4 mineral grains of a range of residence times is tracked. The soil thickness is determined

by summing the contribution of all the mineral grains that reflect mineral species groups 5

6 and number of mineral particles within each group:

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$$h = \sum_{j}^{\# species \# particles} \frac{m_{j,n}}{\chi_{j,i} \rho_{\mu,j,n} (1-\phi)}.$$
 (L)

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This soil thickness is then used to determine P in the following time-step with Eqs. 2a or

11 2b.

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## A Note on Residence Time

14 In the simulations, we track the time lengths that individual mineral grains spend in the

15 soil. The residence time is calculated based on the average of those time lengths. The

mineral grains, however, may have different masses (due to, for example, differential

weathering of varying soil production) such that the residence time is calculated by

weighting the time lengths of the mineral grains by their masses.

# Reference

- 1 White, A., and S.L. Brantley (2003), The effect of time on the weathering of silicate
- 2 Minerals: why do weathering rates differ in the laboratory and field?, Chemical Geology,
- 3 *202*, 479-506.