

1 **GSA Data Repository 2020008**

2
3 Calcium isotope evidence for environmental variability before and across the Cretaceous –
4 Paleogene mass extinction
5 *Linzmeier et al.*
6
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11 ○ Figure DR2 – Comparison of $\delta^{44/40}\text{Ca}$ to $\text{Wt}\%_{\text{carb}}$ from ODP 690
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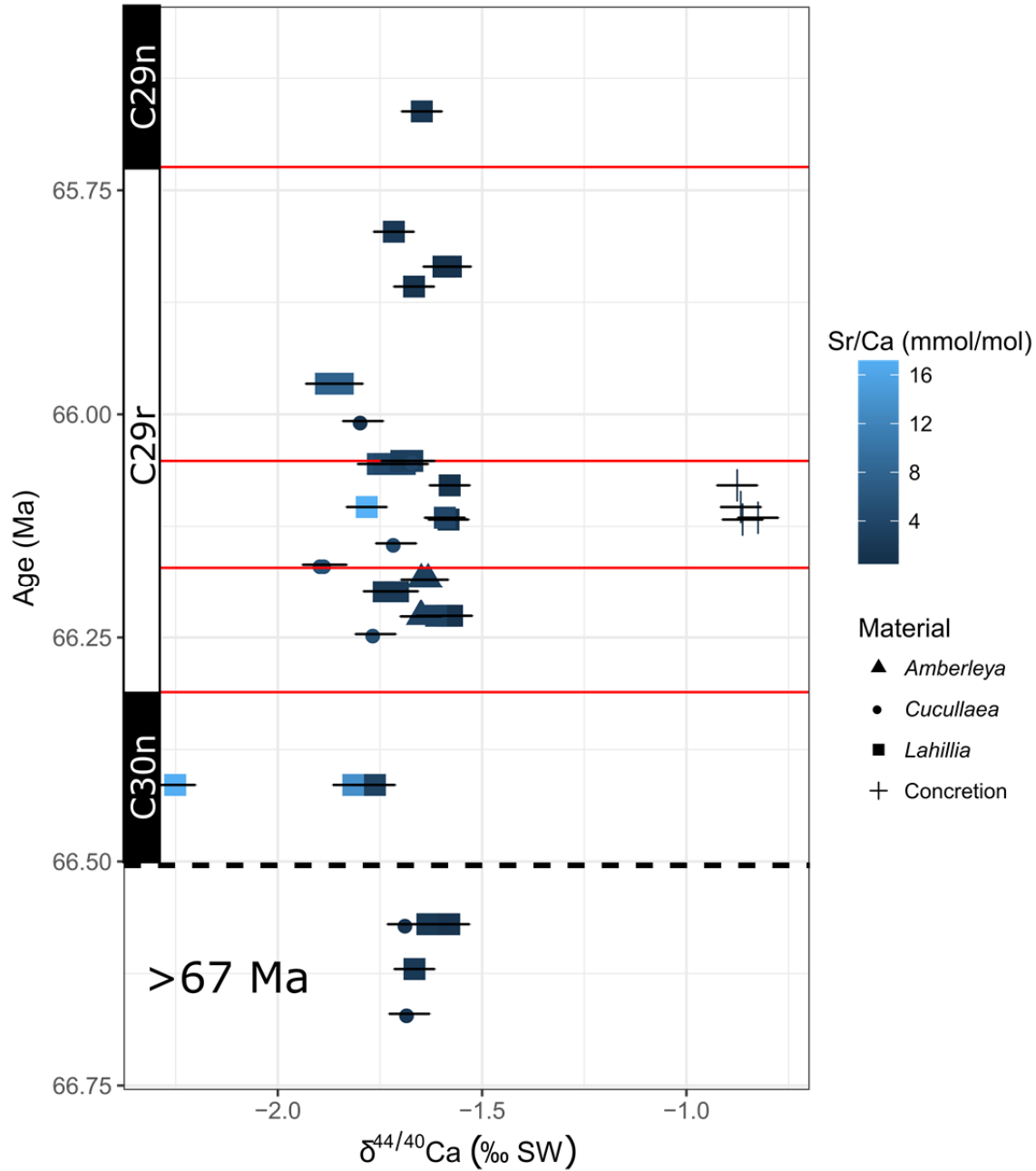
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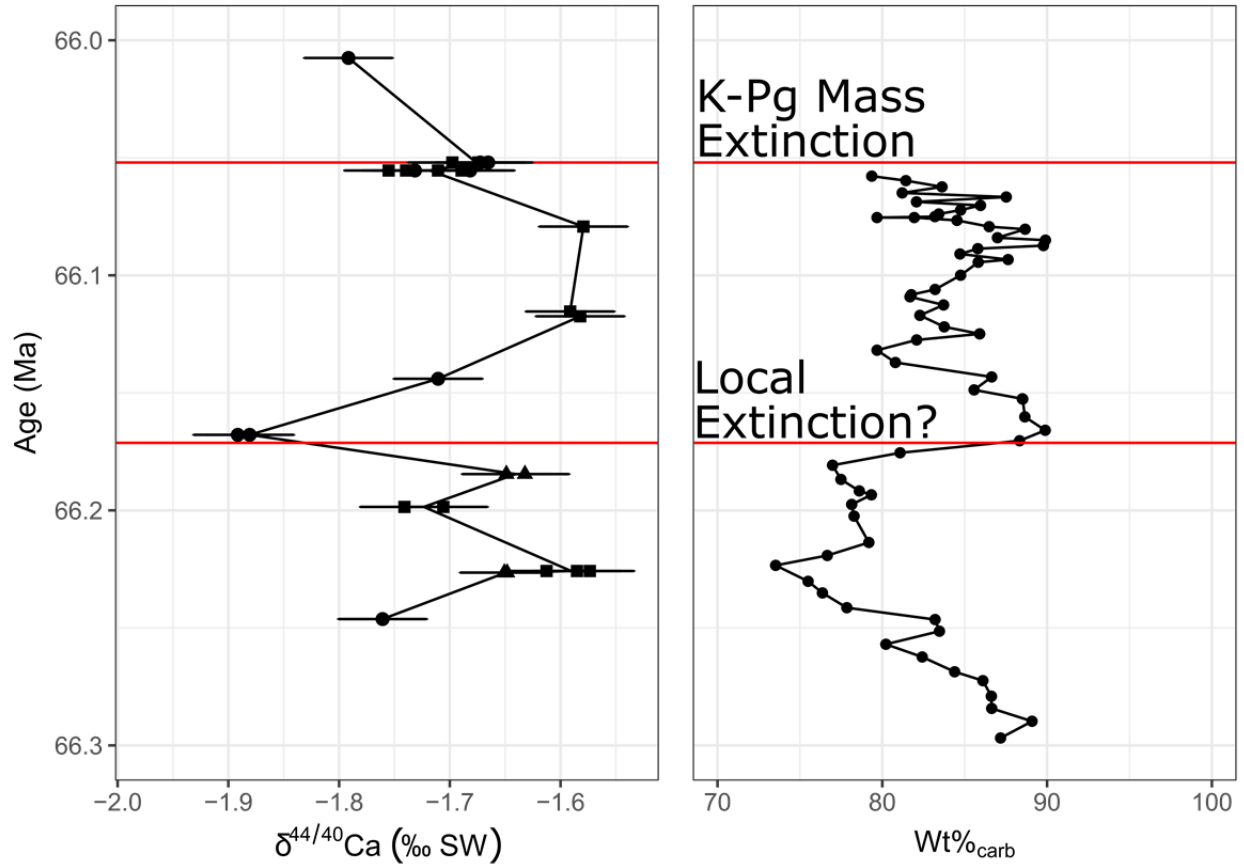
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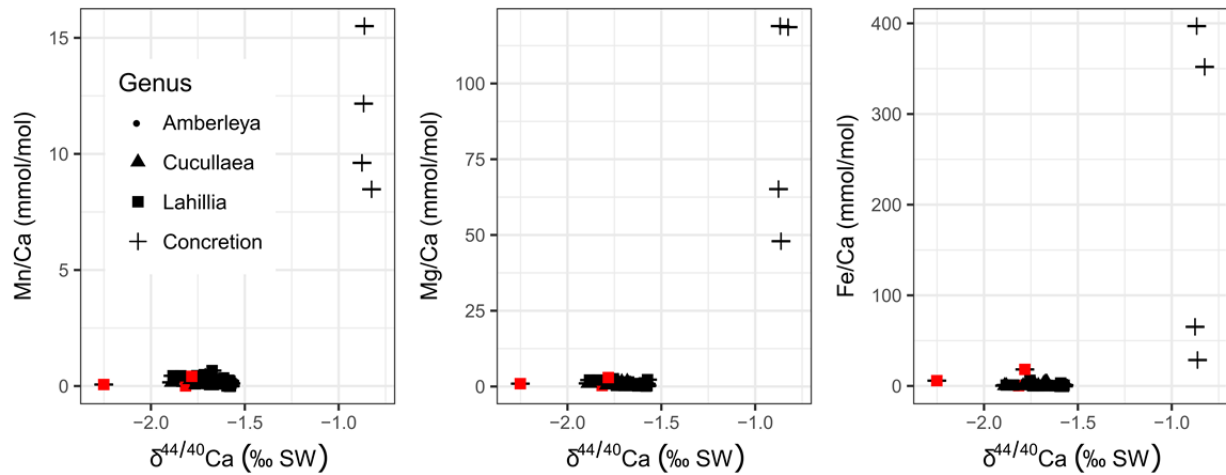
Data Repository Figure DR1. All $\delta^{44/40}\text{Ca}$ through section

All analyses of $\delta^{44/40}\text{Ca}$ through time at Seymour Island. Shells with high Sr/Ca and analyses of carbonate cement are excluded from Figure 3 in the main text. Note that shells with high Sr/Ca have lower $\delta^{44/40}\text{Ca}$ compared to closely contemporaneous samples. Concretion carbonate cement has consistent $\delta^{44/40}\text{Ca}$ near -0.8‰ across several samples, which may reflect the difference between biogenic aragonite and inorganic calcite (Blättler et al., 2012). The three red bars from bottom to top indicate the start of Deccan Traps volcanism, a putative local extinction, and the K-Pg extinction, respectively.



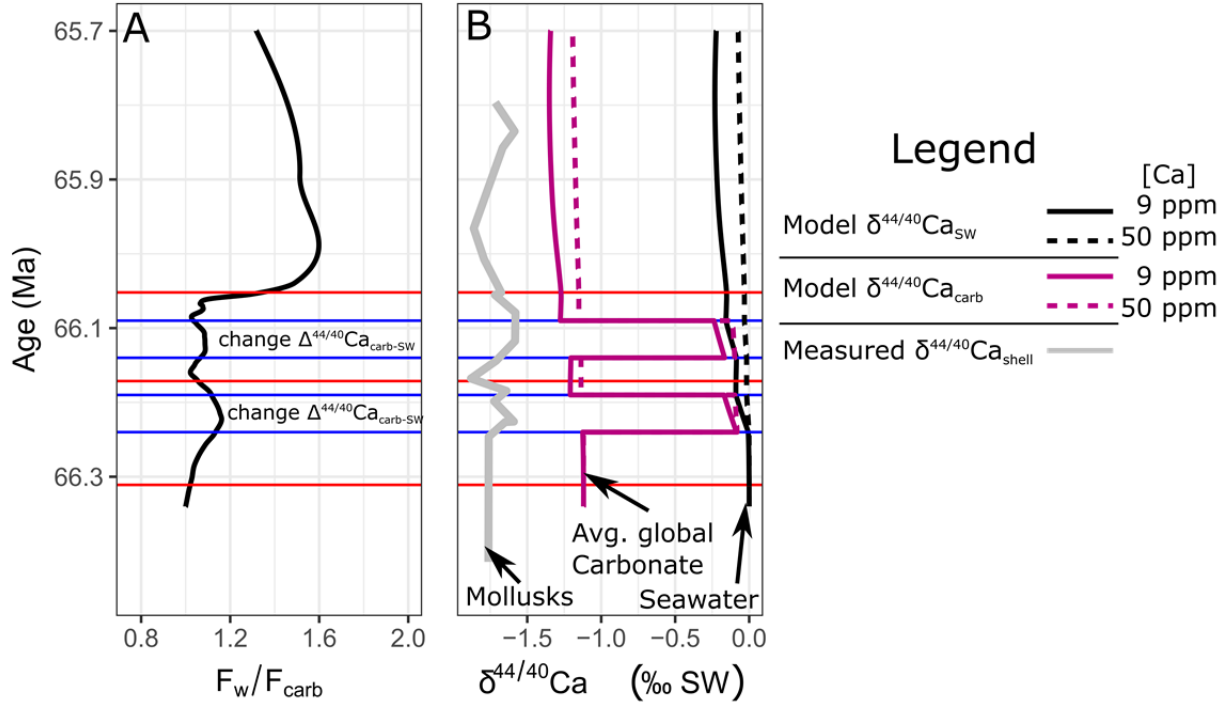
Data Repository Figure DR2. Comparison of $\delta^{44/40}\text{Ca}$ to $\text{Wt}\%_{\text{carb}}$ from ODP 690

Bivalve $\delta^{44/40}\text{Ca}$ from low Sr/Ca samples near to the K-Pg boundary compared to variability in $\text{Wt}\%_{\text{carb}}$ measured at ODP 690. Correlation based on the K-Pg boundary and 30N-29R reversal horizon with linear age interpolations between these horizons; see below for more specifics. The $\text{Wt}\%_{\text{carb}}$ record from this site (Maud Rise) reflects both saturation state and carbonate production in the high latitude ocean (O'Connell, 1990; Ehrendorfer, 1993; Henahan et al., 2016). Modern high latitude oceans are likely more sensitive to saturation state change (Fabry et al., 2009). The slight decrease in $\text{Wt}\%_{\text{carb}}$ below the K-Pg may be due to bioturbation (Henahan et al., 2016). Local minor extinction horizon from Tobin (2017).



Data Repository Figure DR3. Comparison of $\delta^{44/40}\text{Ca}$ to elemental ratios

All analyses of $\delta^{44/40}\text{Ca}$ vs other elemental ratios measured by ICP-OES. Shells with Sr/Ca higher than 10 (mmol/mol) are highlighted in red. For most solutions the [Mn], [Mg], and [Fe] were below calibration range for solutions optimized for Ca and Sr concentrations. These data highlight the difference between shell material and concretion carbonate cement. The low Mn/Ca, Mg/Ca, and Fe/Ca also supports the preservation of unaltered shell.



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62 **Data Repository Figure DR4. $\delta^{44/40}\text{Ca}_{\text{SW}}$ sensitivity to global change in both flux imbalance**
 63 **and $\Delta^{44/40}\text{Ca}_{\text{carb-SW}}$**

64 A) Flux estimates are based on scaling the burial flux of carbonate (F_{carb}) using $Wt\%_{\text{carb}}$ at ODP
 65 690 assuming constant weathering flux (F_w). The magnitude of the global carbonate fractionation
 66 factor ($\Delta^{44/40}\text{Ca}_{\text{carb-SW}}$) is reduced by 1.0‰ between the pairs of blue lines and then returns to the
 67 previous value (-1.12‰) outside of these lines. B) Modeled $\delta^{44/40}\text{Ca}_{\text{SW}}$ and $\delta^{44/40}\text{Ca}_{\text{carb}}$ at two
 68 [Ca] through time compared to the measured $\delta^{44/40}\text{Ca}_{\text{shell}}$. Even with large (~1.0‰) stepwise
 69 change to the global fractionation factor (between the blue lines), minimal $\delta^{44/40}\text{Ca}_{\text{SW}}$ change
 70 (<0.02‰) for the global ocean occurs over the timescale studied.

71 **Data Repository Table DR1. Sample List and Museum Information**

Sample	Museum*	Museum #	Class	Genus
C1174A.2	UM		Bivalvia	<i>Cucullaea</i>
C1184A2	UM		Bivalvia	<i>Cucullaea</i>
C1467A	UM		Bivalvia	<i>Cucullaea</i>
C1517A	UM		Bivalvia	<i>Cucullaea</i>
C1555B	UM		Bivalvia	<i>Cucullaea</i>
C477B	UM		Bivalvia	<i>Cucullaea</i>
C757B	UM		Bivalvia	<i>Cucullaea</i>
C757C2	UM		Bivalvia	<i>Cucullaea</i>
C915A.2	UM		Bivalvia	<i>Cucullaea</i>
JRB-16-0351	UW	UWBM-109960	Bivalvia	<i>Lahillia</i>
JRB-16-0729	UW	UWBM-109958	Bivalvia	<i>Lahillia</i>
JRB-16-0877	UW	UWBM-109964	Bivalvia	<i>Lahillia</i>
JRB-16-0330	UW	UWBM-109965	Bivalvia	<i>Lahillia</i>
L1134A	UM		Bivalvia	<i>Lahillia</i>
L1161B	UM		Bivalvia	<i>Lahillia</i>
L1161Sed	UM		Sediment	Sediment
L1430A1	UM		Bivalvia	<i>Lahillia</i>
L1474A1	UM		Bivalvia	<i>Lahillia</i>
L1480C	UM		Bivalvia	<i>Lahillia</i>
L1480D	UM		Bivalvia	<i>Lahillia</i>
L1516A	UM		Bivalvia	<i>Lahillia</i>
L1516B	UM		Bivalvia	<i>Lahillia</i>
L1516C	UM		Bivalvia	<i>Lahillia</i>
L1529B	UM		Bivalvia	<i>Lahillia</i>
L1529C	UM		Bivalvia	<i>Lahillia</i>
L1609	UM		Bivalvia	<i>Lahillia</i>
L477B	UM		Bivalvia	<i>Lahillia</i>
L477D	UM		Bivalvia	<i>Lahillia</i>
L757A	UM		Bivalvia	<i>Lahillia</i>
L757C	UM		Bivalvia	<i>Lahillia</i>
SI-11-466A-1	UW	UWBM-109966	Gastropoda	<i>Amberleya</i> (<i>Ambercyclus</i> ? Witts et al., 2016)
SI-11-528A-1	UW	UWBM-109967	Gastropoda	<i>Amberleya</i> (<i>Ambercyclus</i> ? Witts et al., 2016)
SI-11-570A-1	UW	UWBM-109968	Bivalvia	<i>Lahillia</i>
SI-11-570S-2	UW	UWBM-109969	Sediment	Sediment
SI-11-624A-2	UW	UWBM-109970	Bivalvia	<i>Lahillia</i>
SI-11-624S-1	UW	UWBM-109971	Sediment	Sediment
SI-11-722A-1	UW	UWBM-109972	Bivalvia	<i>Lahillia</i>
SI-11-722S-1	UW	UWBM-109973	Sediment	Sediment

72 *Institution name: UW – University of Washington, UM – University of Michigan

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74 **Data Repository Table DR2. Seymour Island age model tie points**

Tie Point	Stratigraphic Height (m) Tobin et al. 2012	Stratigraphic Height (m) Zinsmeister 2001	Age (Ma)	Reference
29R-29N	924	1118	65.724	Sprain et al., 2018
KPg Boundary	865	1059	66.052	Sprain et al., 2018
30N-29R	789	983	66.311	Sprain et al., 2018
30R-30N	607	801	68.196	GTS 2012

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76 **Data Repository Table DR3. Global age model tie points**

Tie Point	ODP 690 Stratigraphic Height (m) Hamilton, 1990	ODP1209 Age (Ma) Barnet et al., 2018	ODP1262 Age (Ma) Henehan et al., 2016	ODP1262 Age (Ma) Westerhold et al., 2008
29R-29N	247.55	65.608	-	64.192
KPg Boundary	247.82	66.022	66.04	65.28
30N-29R	252.58	66.407	66.398	65.625
30R-30N	-	-	-	-

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78 **Data Repository Table DR4. Trace element and $\delta^{44/40}\text{Ca}$ dataset [See Excel sheet]**

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80 **Data Repository Table DR5. $^{87}\text{Sr}/^{86}\text{Sr}$ dataset [See Excel sheet]**

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82 **Data Repository Table DR6. Summary statistics of replicates**

Replicate type	Average range* $\delta^{44/40}\text{Ca}$ (‰)	Number of samples	Number of analyses
Sample	0.0257	10	21
Horizon	0.0383	11	32

83 *Average maximum minus minimum value for replicate analyses as defined in the table.

84 **Data Repository Table DR7. Range of calcium concentration for the Late Cretaceous**

Residence time τ_{Ca} (Myr)	Scale for τ_{Ca}	Reservoir size (mol)	[Ca] (mmol/kg)	Reference
0.45	1	1.44E+19	6.00E-03	Du Vivier et al., 2015
1.27	2.83	4.08E+19	1.70E-02	Horita et al., 2002
0.68	1.5	2.16E+19	9.00E-03	Lasaga et al., 1985; Wilkinson and Algeo, 1989
1.50	3.34	4.80E+19	2.00E-02	Stanley and Hardie, 1998
2.25	5	7.20E+19	3.00E-02	Hardie, 1996
3.75	8.34	1.20E+20	5.00E-02	Wallmann, 2001

85 **Data Repository Table DR8. Flux balance model parameters**

Symbol	Description	Value	Reference
F_w	Weathering input flux	$3.2E+13$ mol/yr	Sum input flux Du Vivier et al., 2015
δ_w	Weathering input isotope ratio	-1.12‰	Du Vivier et al., 2015
Δ_c	Carbonate Fractionation factor	-1.12‰	Du Vivier et al., 2015
δ_{swi}	Initial isotope ratio of seawater	0.00‰	Zhu and Macdougall, 1998

86
87 **Supplementary Discussion: Full description of geological setting**

88 The López de Bertodano Formation (LBF) was deposited in the James Ross Basin during
89 the Maastrichtian-Danian and predominantly comprises siliciclastic clays and silts with rarer
90 sandy horizons. While the LBF has commonly been considered as deposited in an open-ocean
91 facing shelf environment (Macellari, 1988), redox proxies have implied some degree of
92 circulation restriction (Schoepfer et al., 2017). Carbonate concretions are common within the
93 mudrock facies, and sandier beds often have calcareous cements. The upper 300 m of the section
94 comprise bioturbated siltstones with occasional mappable glauconite-rich horizons capped by
95 lags of mollusks and other fossils (Macellari, 1988; Olivero, 2012; Witts et al., 2015).
96 Glauconite-rich horizons suggest sediment starvation and possibly record flooding surfaces
97 formed in response to short-duration, low-amplitude sea level rises (Chafetz and Reid, 2000).
98 Sharp boundaries above glauconitic layers may suggest shallower deposition with storm
99 scouring. Mean sedimentation rates are estimated to be between 10 and 30 cm/kyr using
100 dinoflagellate cyst and ammonite biostratigraphy, with comparison to magnetostratigraphy and
101 the K-Pg boundary horizon (Witts et al., 2015). Magnetostratigraphically constrained
102 sedimentation rates range from 10 to 20 cm/kyr (Tobin et al., 2012).

103 There is broad agreement on a deepening upward trend during the deposition of the LBF,
104 but specific depth interpretations vary. Estimates of water depth for deposition of the upper LBF
105 range from 100-200 m (Huber, 1988; Macellari, 1988); however, Crame et al. (2004) argued for
106 deeper outer shelf conditions. The uppermost portion of the section containing samples analyzed
107 in this study has a generally uniform sedimentology, which suggests no appreciable differences
108 in water depth between the specimens (Crame et al., 2004). Despite a lack of sedimentological
109 change, Macellari (1988) used paleoecological change to infer a regression roughly 50 m below
110 the K-Pg boundary, which may instead be a biological event (Tobin, 2017).

111 Post-depositional alteration of the LBF appears minimal; the formation dips shallowly at
112 10° to the southeast, and many mollusks are preserved with nacreous aragonite that maintains
113 iridescence, indicating that temperatures remained below 60°C (Tobin et al., 2012). The absence
114 of any smectite-illite clay transformation and the presence of apatite fission tracks also suggests
115 low burial temperatures (<80°C) and shallow depths (Pirrie, 1994). This evidence, combined
116 with published data, indicates that many shells from this location are well-preserved and suitable
117 for application to paleoclimate and paleoenvironmental reconstructions (Tobin et al., 2012;
118 Tobin and Ward, 2015; Petersen et al., 2016; Witts et al., 2018).
119

Supplementary Discussion: Extended analytical methods

Elemental concentrations

Approximately 50 mg samples powdered by hand drilling or crushing chips were weighed into acid-cleaned centrifuge tubes and dissolved in ultrapure 5% HNO₃. Outgassed CO₂ was released after ~10 minutes of dissolution, when most powder had dissolved. Centrifuge tubes containing mixtures were then placed on a rocker table overnight. The mixtures were centrifuged and passed through 0.45 µm filters to obtain stock solutions. An aliquot of each stock solution was diluted to an estimated 12 ppm [Ca] in 7 mL of 5% HNO₃. Elemental concentrations were measured using a Thermo Scientific iCAP 6500 ICP-OES in the Aqueous Geochemistry Laboratory at Northwestern University. Standardization and instrument performance were checked with repeated measurements of NIST SRM 1643f.

Calcium isotope ratios (⁴⁴Ca/⁴⁰Ca)

Stable calcium isotope ratios (⁴⁴Ca/⁴⁰Ca) were measured in the Radiogenic Isotope Clean Laboratory at Northwestern University using an optimized ⁴³Ca-⁴²Ca double-spike TIMS technique (Lehn et al., 2013). Measurements were done using a Thermo-Scientific Multicollector Triton Thermal Ionization Mass Spectrometer equipped with 10¹¹ Ω amplifier resistors. All procedures employed ultrapure reagents. Total procedural blanks determined by isotope dilution averaged 29 ng of Ca (n = 8), which is negligible compared to the amount of Ca processed for isotopic analysis (50 µg).

Sample aliquots containing 50 µg of Ca were weighed into Teflon vials, spiked, and equilibrated at 60 °C overnight on a hotplate. Solutions were dried at 90 °C, and the residues were dissolved in 0.5 mL of 1.55N HCl. Calcium was separated from other elements by eluting samples through Teflon columns packed with Bio-Rad AG MP-50 cation exchange resin. The purified fractions were dried, reacted with 2 drops of 35% H₂O₂ to oxidize organic compounds, reacted with 2 drops of 16N HNO₃ to convert Ca to nitrate form, and dried. The residues were dissolved in 0.4 µL of 8N HNO₃ and then split into 4 equal beads containing ~12 µg of Ca. A single bead was loaded onto outgassed, single Ta filament assemblies between thin parafilm “dams” spaced ~0.5 mm apart. The beads were dried slowly at 1.6 amps, and then 1.0 µL of 10% H₃PO₄ was added before a final dry down at 3.5 amps.

In the mass-spectrometer, a 20V ⁴⁰Ca ion-beam was achieved after warm-up, and ⁴⁰Ca/⁴²Ca, ⁴³Ca/⁴²Ca, and ⁴³Ca/⁴⁴Ca ratios were measured using a three-hop duty cycle having variable integration times. Amplifier biases were cancelled using amplifier rotation, and the ⁴¹K beam was monitored to confirm that ⁴⁰K did not isobarically interfere with ⁴⁰Ca (⁴⁰K/⁴¹K = 0.00174). Datasets comprising 120 scans were reduced using an iterative procedure. All runs were evaluated to ensure a steady or increasing ⁴⁰Ca beam, an increasing raw ⁴²Ca/⁴⁰Ca fractionation pattern, and the absence of filament reservoir effects, which appear as “reverse-fractionation” in uncorrected ratios or residual trends in fractionation-corrected ratios. Changes in collector cup efficiency were monitored and accounted for by analyzing at least 6 Ocean Scientific International Ltd. (OSIL) seawater (SW) standards and 2 NIST 915b standards every 30 samples. All results are reported in delta notation relative to SW. The internal precision of the sample runs ranged from ±0.016‰ to ±0.027‰ (2SEM). During the period of study, repeated analyses of OSIL SW and NIST 915b yielded mean δ^{44/40}Ca values of 0.00‰ ± 0.009‰ (2SEM, n = 37) and NIST -1.13 ± 0.016‰ (2SEM, n = 10). These results correspond to a short-term,

external reproducibility of $\pm 0.05\%$ (2SD), which is the uncertainty adopted for the present study. As described in the main text, replicates of samples were better than $\pm 0.05\%$.

Radiogenic strontium isotope ratios ($^{87}\text{Sr}/^{86}\text{Sr}$)

Strontium isotope ratios ($^{87}\text{Sr}/^{86}\text{Sr}$) were also measured in the Radiogenic Isotope Clean Laboratory at Northwestern University, following the TIMS procedure outlined in Andrews et al. (2016). Sample aliquots containing 100 ng of Sr were dried in Teflon vials, and the residues were dissolved in 8 M HNO_3 . Strontium was separated from matrix elements by eluting samples through inverted pipet tips packed with Eichrom Sr-Spec resin. The purified fractions were dried, and the residues were dissolved in 1.5 μL 3 N HNO_3 . The aliquots were loaded onto outgassed, single Re filament assemblies, together with 1 μL of a TaCl_5 activator solution, and dried at 1.0 amps before heating to ~ 2.2 amps. In the mass-spectrometer, an 8V ^{88}Sr ion-beam was achieved before collecting data in multi-dynamic mode for 120 duty cycles having a 4 s integration time. Measurements were made with amplifier rotation. The ^{85}Rb beam was monitored to ensure that ^{87}Rb did not isobarically interfere with ^{87}Sr ($^{87}\text{Rb}/^{85}\text{Rb} = 0.3856$). Instrumental mass fractionation was corrected by normalizing $^{86}\text{Sr}/^{88}\text{Sr}$ ratios to 0.1194 using an exponential law. During the period of study, measurements of NBS 987 yielded an $^{87}\text{Sr}/^{86}\text{Sr}$ of 0.710252 ± 0.000008 (2SD, $n=5$).

Supplementary Discussion: Description of age model and correlation

Seymour Island Age model:

The age model for many of the samples builds on the stratigraphic framework outlined by Zinsmeister (2001), where plane projection was used to place georeferenced samples distributed across the island onto a unified stratigraphic section. Several samples analyzed in this study were collected within a measured stratigraphic section (Tobin et al., 2012). Placement of samples from the Tobin et al. (2012) framework into the Zinsmeister (2001) framework was done by adding 194 m to the stratigraphic height. This same stratigraphic framework has been used for previous geochemical studies (Petersen et al., 2016) and the analysis of the K-Pg extinction (Wang and Marshall, 2004; Tobin, 2017). Stratigraphic thicknesses for Molluscan Units, as defined by Macellari (1988), are within ~ 4 m between the Tobin et al. (2012) and Zinsmeister (2001) frameworks (Petersen et al., 2016). Analyses of $\delta^{18}\text{O}$ and $\delta^{13}\text{C}$ from samples in both frameworks are comparable (Tobin et al., 2012). Given fixed sedimentation rates, ± 4 m uncertainty is equivalent to age uncertainties of ~ 13 to ~ 40 kyr. It should also be noted that the relative ordering of samples is robust relative to the K-Pg horizon due to significant lithological changes and the high sedimentation rates in the location.

Numerical ages for the section are derived assuming constant rates of sediment accumulation between chron reversals and the K-Pg boundary. Magnetostrigraphic reversals (C30R-C30N-C29R-C29N) are documented in the López de Bertodano Formation (Tobin et al., 2012). The age of one sample (JRB-16-0877) from 11 m above the base of C29N is extrapolated assuming the same sedimentation rate. Ages used for these reversals are from Sprain et al. (2018) and Gradstein et al. (2012). The age used for the K-Pg boundary is 66.052 Ma (Sprain et al., 2018). Correlation to other globally distributed datasets are outlined in the following paragraphs, and Table DR2 provides a complete list of ages used for the age model construction.

Correlation to ODP 690:

Data from ODP 690 are correlated to Seymour Island using chron reversals and the K-Pg boundary. The positions of chron reversals are from Hamilton (1990). Data on carbonate weight % are from several sources (O’Connell, 1990; Ehrendorfer, 1993). Some bioturbation near the K-Pg boundary may have produced a pre-impact decrease in carbonate weight percent due to moving carbonate up and non-carbonates lower in the core (Henehan et al., 2016), but perhaps the decrease is real and related to an increase in Deccan volcanism just before the K-Pg boundary (Ehrendorfer, 1993; Westerhold et al., 2011; Henehan et al., 2016; Dameron et al., 2017; Schoene et al., 2019; Sprain et al., 2019). For the purposes of the null hypothesis model based on flux imbalances (see below), the impact of this bioturbation is minimal and only matters slightly for low seawater calcium concentrations.

Correlation to ODP 1262:

Data from ODP 1262, as presented in Barnet et al. (2018), were correlated to the Seymour Island section by adjusting all data based on a slightly different K-Pg boundary age of 66.052 Ma (Sprain et al., 2018) rather than the astronomically tuned age of 66.02 Ma (Dinarès-Turell et al., 2014) used by Barnet et al. (2018). We also tie the C30N-C29R and C29R-C29N reversal ages to that from Sprain et al. (2018) and linearly scale astronomically tuned numeric ages between the two. Similar correlations to this dataset have been done recently by other authors (Schoene et al., 2019; Sprain et al., 2019). It should also be noted that the sedimentation rates differ greatly between 1 – 2 cm/kyr at ODP sites and 10 – 30 cm/kyr at Seymour Island.

Correlation to ODP 1209:

Data from ODP 1209, as presented in Westerhold et al. (2011) and Henehan et al. (2016), are correlated to our section by adjusting the ages of the K-Pg, C30N-C29R, and C29R-C29N to updated ages from Sprain et al. (2018) and scaling their published ages between these tie points.

Correlation to Deccan Traps flow volumes:

Deccan Traps flow volumes are from Richards et al. (2015), and durations of flows are from Sprain et al. (2019). Eruption rate estimates are from U-Pb dating of zircons done by Schoene et al., (2019) and are scaled linearly between the K-Pg and C30N-C29R. Correlation between these records is done using the same chron reversals and the K-Pg.

Uncertainty in the age model:

Correlation of data from Seymour Island to other records hinges on the assumption of continuous steady rate sedimentation in this location and other locations, if solely tied together using chron reversals and the K-Pg boundary. Future work at Seymour Island, including astronomically tuned age models, will help determine the validity of this assumption.

Supplementary Discussion: Description of illustrative box model

A simple numerical model of the marine Ca isotope cycle is used to constrain drivers of $\delta^{44/40}\text{Ca}$ variation, specifically imbalances between primary input and output fluxes (Fig. 4, Fig. DR4). The model presents a simple null hypothesis for $\delta^{44/40}\text{Ca}$ variation caused by changes in the ‘weathering flux’ relative to the ‘carbonate flux.’ Flux imbalance is hypothesized to be one of the major drivers of $\delta^{44/40}\text{Ca}$ evolution through Earth history (DePaolo, 2004; Fantle and

DePaolo, 2005; Fantle, 2010; Fantle and Tipper, 2014; Tipper et al., 2016). Although more complex modeling—coupling to seawater carbonate chemistry and $p\text{CO}_2$ weathering feedbacks, for example—can be done for the marine $\delta^{44/40}\text{Ca}$ system, difficulties persist in reproducing $\delta^{44/40}\text{Ca}$ excursions measured in carbonates because isotope fractionation mechanisms remain incompletely understood (Komar and Zeebe, 2016; Jost et al., 2016; Silva-Tamayo et al., 2018). Relative uniformity of the $^{87}\text{Sr}/^{86}\text{Sr}$ record (Fig. 3) suggests that the weathering flux did not substantially change over the sampled interval, so in the subsequent forward model, we assume a constant weathering flux and a variable carbonate burial flux. This model is parameterized using the same fluxes and Ca isotope compositions as Du Vivier et al. (2015), but higher $[\text{Ca}]$ values and longer residence times. This model is intended to explain horizon-to-horizon variations through changes in the $\delta^{44/40}\text{Ca}$ composition of seawater, but not reproduce the absolute $\delta^{44/40}\text{Ca}$ values of mollusk shell due to unknown controls on biogenic fractionation (see extended discussion below).

The forward model for the $\delta^{44/40}\text{Ca}$ of seawater based on flux imbalances is done assuming equality of the weathering input flux (F_w) and carbonate export flux (F_{carb}) at 66.5 Ma. F_{carb} is then scaled by changes in carbonate weight percent (wt_{carb}) from this point in time. For example, if wt_{carb} is reduced by 50% from the initial value (e.g., 90% to 45%), then F_{carb} is reduced by 50%. In this modeling, the initial mass of calcium in the ocean is run at the limits of estimated ranges for Late Cretaceous seawater (~ 10 to 50 mmol/kg, Lasaga et al., 1985; Wallmann, 2001). To create an initial steady state with a fixed F_w across the range of masses modeled here, initial F_{carb} must remain constant. In this model, F_{carb} is equivalent to the moles of Ca in the ocean (N_{Ca}) divided by the residence time of Ca (τ_{Ca}), so to maintain an initial steady state any adjustment to N_{Ca} also requires an adjustment to τ_{Ca} . We scale both by multiplying by a constant scaling factor to allow a fixed F_w for all model runs (Table DR 7). The model equations are:

$$\begin{aligned} 1) \quad \frac{dN_{Ca}}{dt} &= F_w - \frac{wt_{carb_t}}{wt_{carb_i}} * \frac{N_{Ca_t}}{\tau_{Ca}} \\ 2) \quad \frac{d(N_{Ca}\delta_{SW})}{dt} &= F_w * \delta_w - (\delta_{SW} + \Delta_C) * \frac{wt_{carb_t}}{wt_{carb_i}} * \frac{N_{Ca_t}}{\tau_{Ca}} \end{aligned}$$

where fixed variables were set using values in Table DR8. The carbonate weight percent that varies F_{carb} in the model is a loess-smoothed and interpolated record of ODP site 690 combining several data sources (O’Connell, 1990; Ehrendorfer, 1993), which is correlated into the Seymour Island timescale as outlined above. With this method, F_{carb} reduction at the K-Pg boundary is likely overestimated because indications of carbonate preservation (e.g., coarse fraction, unbroken planktic foraminifera) increase in multiple cores immediately after the boundary (Henahan et al., 2016). This occurs because wt_{carb} may have decreased in any single core, while another area could have preserved carbonate due to increased saturation of the whole ocean (Henahan et al., 2016; Luo et al., 2016; Boudreau et al., 2018). In other words, although wt_{carb} goes down in one location, alkalinity buildup in the global ocean may enhance carbonate preservation in others. Additionally, changes in wt_{carb} are sensitive to carbonate production, including biological productivity, saturation state at the seafloor, riverine runoff, and the proportion of foraminifera to coccolithophores (Henahan et al., 2016). Because of these complex

controls, wt_{carb} is likely a better flux estimator than a highly sensitive saturation state proxy (Henehan et al., 2016).

To reproduce the relative mollusk $\delta^{44/40}\text{Ca}$ pattern by maintaining a fixed fractionation factor and allowing $\delta^{44/40}\text{Ca}_{\text{SW}}$ to vary due to flux imbalances, substantial (4x or more) and high frequency (<200 kyr) shifts in weathering and carbonate fluxes are required. Although weathering rates may have increased somewhat through the interval (Martin and Macdougall, 1991; McArthur et al., 1998; Dessert et al., 2001; Tobin et al., 2017), it seems unlikely that either weathering or carbonate fluxes can vary by such large magnitudes over brief time spans. Moreover, the absence of variation in the measured $^{87}\text{Sr}/^{86}\text{Sr}$ record precludes substantial changes in weathering.

Reduced aragonite export is expected during OA (Orr et al., 2005), but decreases in the proportion of aragonite-to-calcite burial drive $\delta^{44/40}\text{Ca}_{\text{SW}}$ in the opposite direction relative to the observed record. Moreover, OA is expected to reduce fractionation for all carbonate minerals (Tang et al., 2008; Nielsen et al., 2012), which causes similar discrepancies between measured and modeled data.

Supplementary Discussion: Discussion of calcium isotope fractionation and biomineralization

Mollusk carbonate offers a valuable geochemical archive because of the abundant methods available to screen shells for diagenetic alteration (Cochran et al., 2010; Witts et al., 2018). Nonetheless, biological control of Ca isotope fractionation (Gussone and Heuser, 2016) and averaging timescales of growth and shell accumulation (Kidwell, 2002; Judd et al., 2018; Linzmeier et al., 2018) add complexity to interpreting data derived from these materials.

Data presented in this paper span several temporal scales. First, the analysis of each shell averages multiple years of growth (Petersen et al., 2016). This masks any seasonal variability that might be coupled to temperature (Immenhauser et al., 2005; Steuber and Buhl, 2006; Hippler et al., 2013) or seasonal changes in the carbonate chemistry of local seawater (Kelly and Hofmann, 2013; Waldbusser and Salisbury, 2014). Second, replicate shells from the same or closely spaced stratigraphic levels demonstrate $\delta^{44/40}\text{Ca}$ variability on the timescales of hundreds to thousands of years necessary for shell bed accumulations (Kidwell, 2002). Similarity between shells from the same horizons suggest environmental conditions that control $\delta^{44/40}\text{Ca}$ persist over shell accumulation timescales. Third, level-to-level variability through the section represents mean changes in environmental parameters that drive $\delta^{44/40}\text{Ca}$ across timescales of tens of thousands of years, as derived from the linear sedimentation age model.

More study of modern mollusks is needed to address Ca isotope fractionation by these organisms but placing our data in context of globally distributed proxies suggests a CO_2 -linked driver. It is beyond the present dataset to determine the exact mechanism of fractionation in these mollusks, but our findings suggest investigations of covariation between carbonate saturation state and $\delta^{44/40}\text{Ca}$ are merited. Samples from natural environments with different $p\text{CO}_2$ and therefore carbonate saturation states may provide the best analog to changes experienced by organisms on intergenerational timescales as the K-Pg mollusks experienced. Existing datasets may be reinterpreted in this context given poor fit to temperatures (Hippler et al., 2013). Given covariation with sedimentary indicators of saturation, fractionation may mimic responses seen in abiotic precipitation experiments, at least in sign (Tang et al., 2008; Nielsen et al., 2012). The

331 mechanism of fractionation may lie in calcium transport into the extrapallial fluid (EPF).
332 Precipitation of shell is isolated from seawater in the EPF between the shell and mantle.
333 Mollusks regulate the pH and saturation state of EPF (Heinemann et al., 2012). Intracellular
334 calcium channels move most of the calcium into the EPF, and they enable high, selective
335 diffusive fluxes which are controlled by biomineralization rates (Carré et al., 2006). Other
336 pathways, such as passive, non-selective intercellular Ca transport and active enzymatic (Ca²⁺-
337 ATPase and carbonic anhydrase) pumping, may also fractionate calcium isotopes (Carré et al.,
338 2006).

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